

Artificial Metalloenzymes: Challenges and Opportunities (An Introduction)

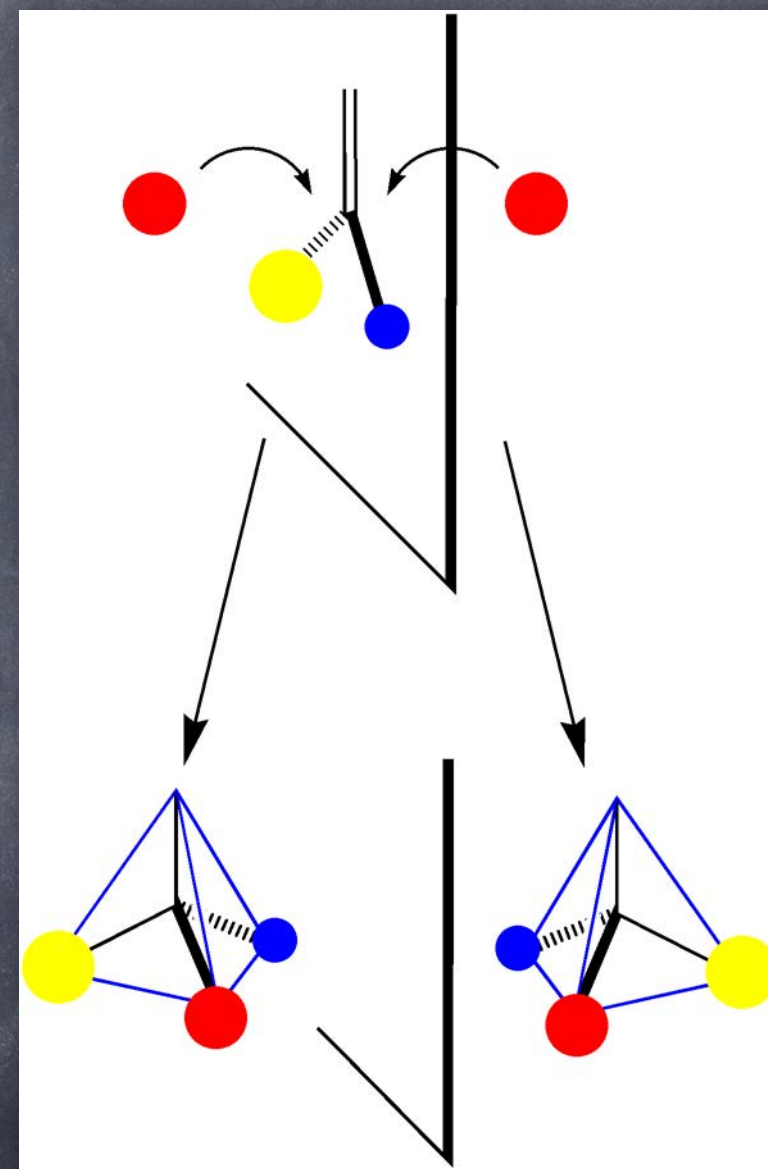
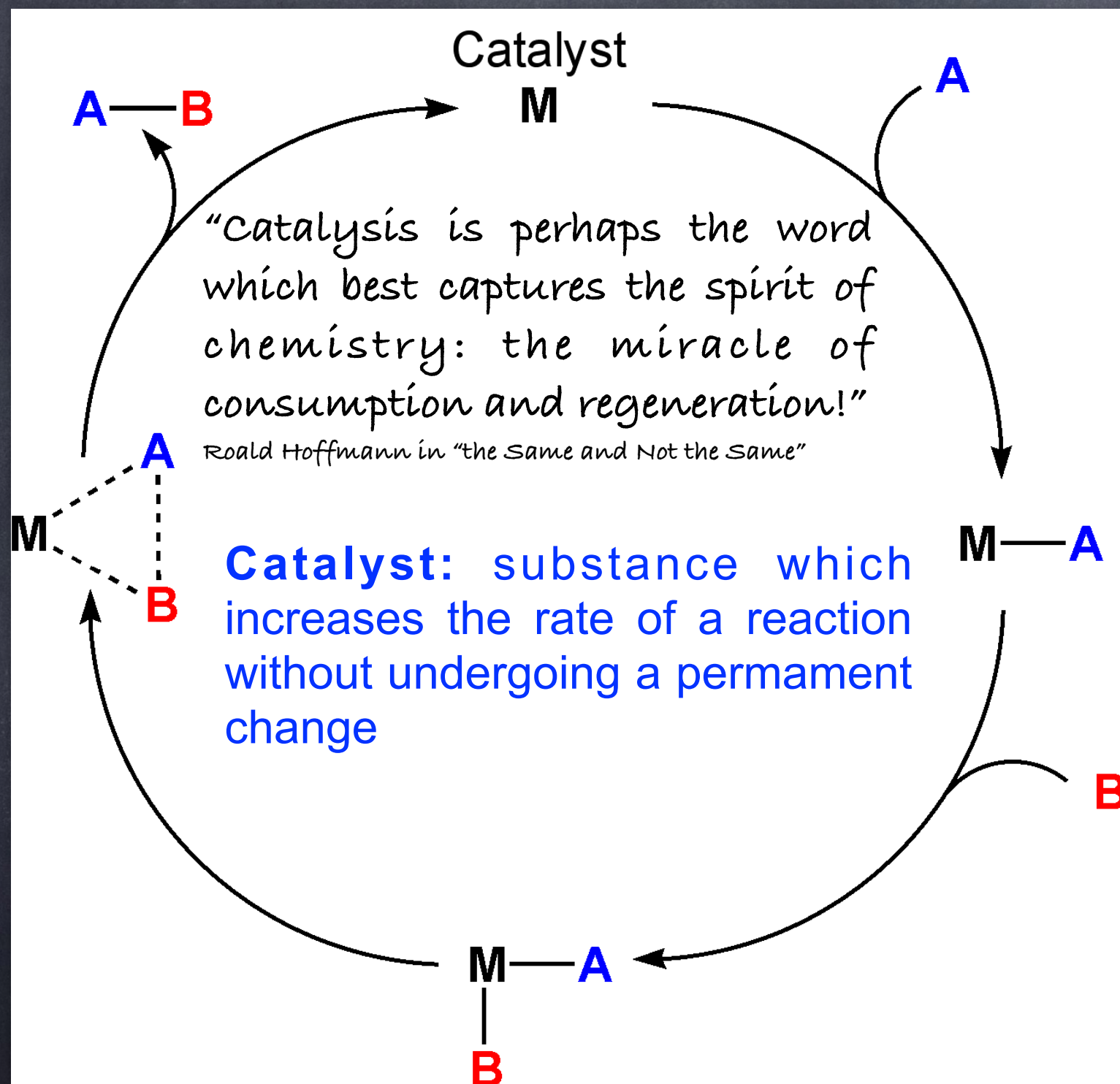
Theracat Workshop

Sept 25, 2019

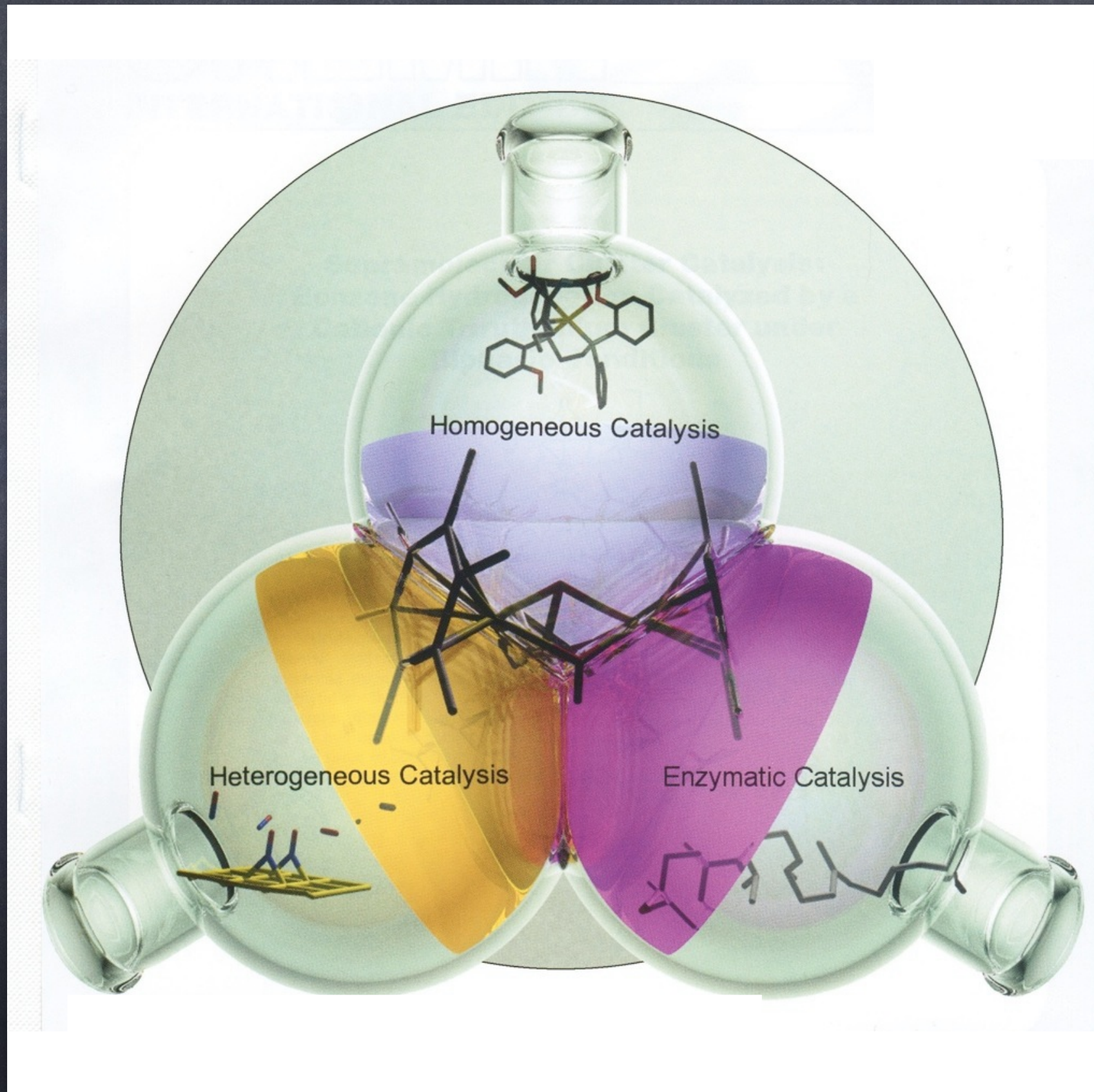
Thomas Ward

University of Basel

Catalysis: Definition



Heterogeneous-, Homogeneous- and Enzymatic Catalysis: Three Kingdoms



The Most Important Discovery of the 20th Century: Haber-Bosch Nitrogen Fixation

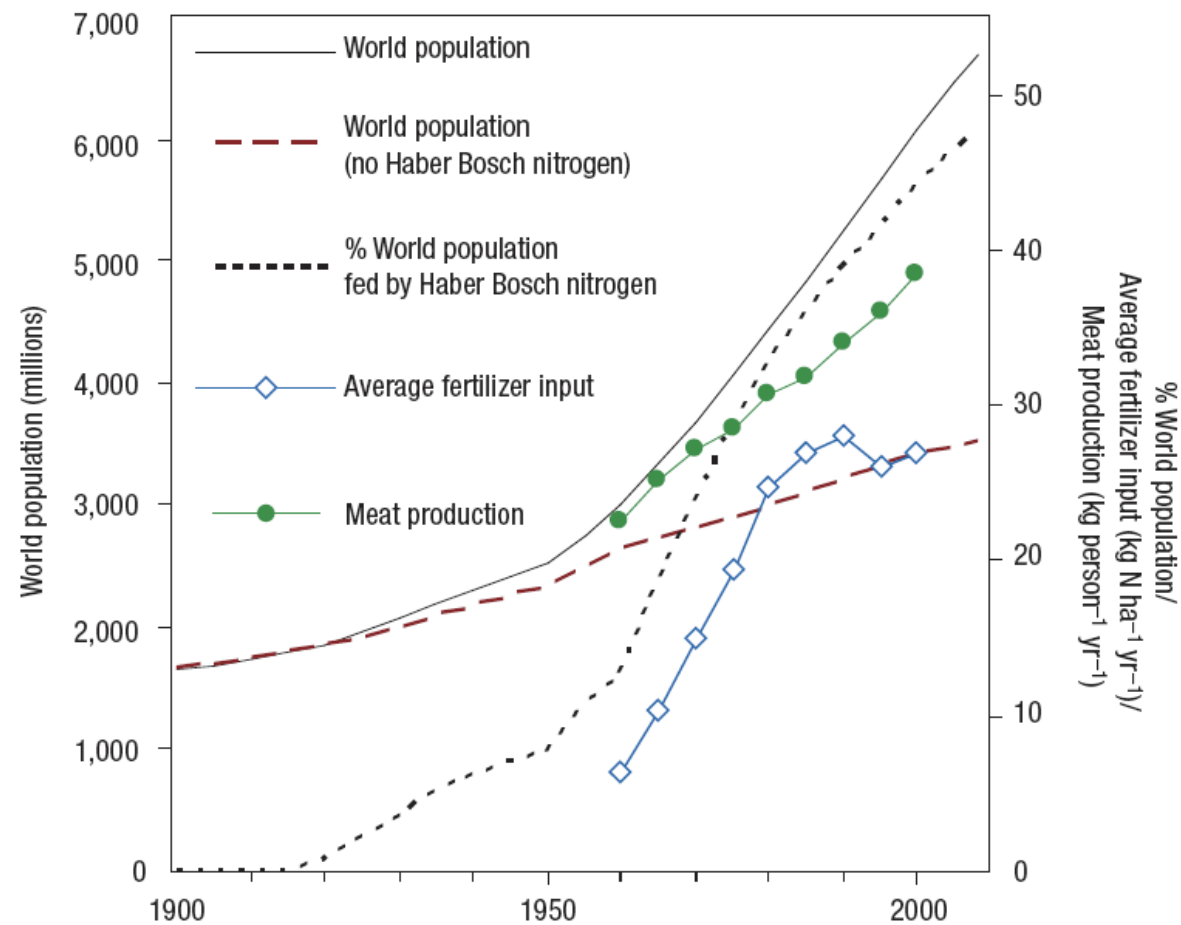
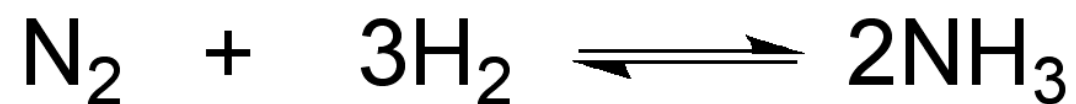
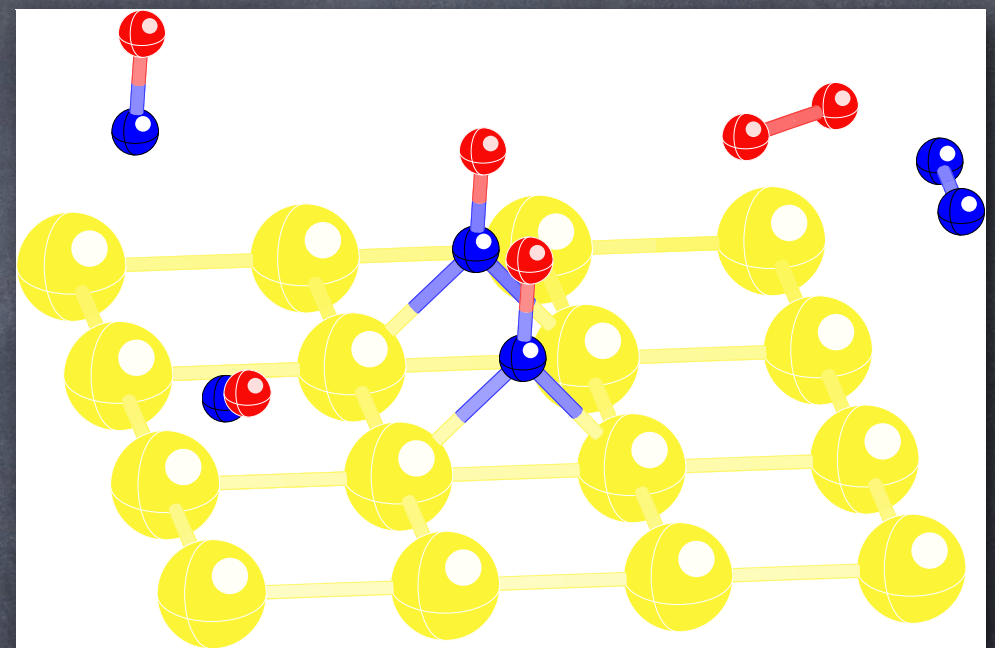
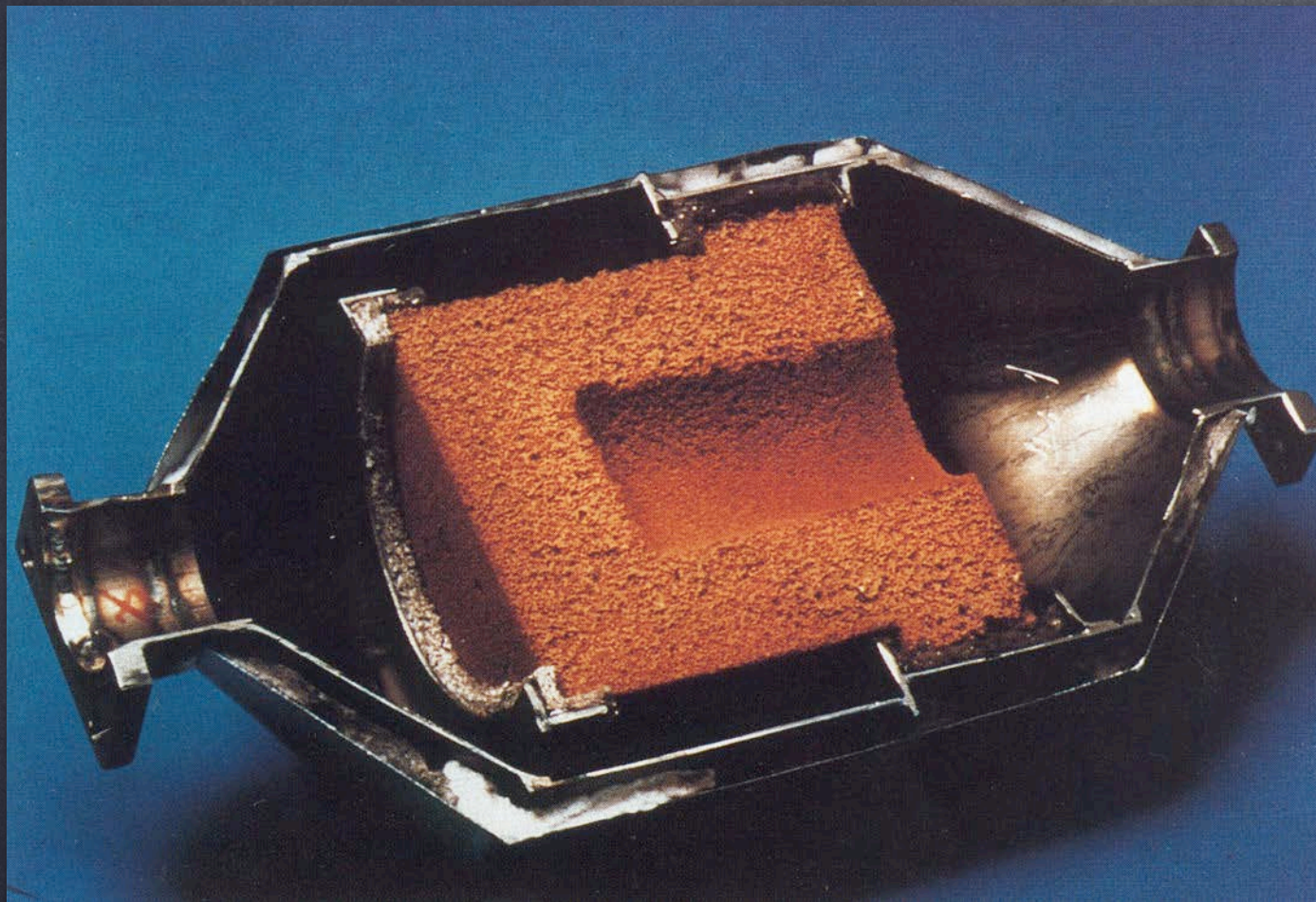


Figure 1 Trends in human population and nitrogen use throughout the twentieth century. Of the total world population (solid line), an estimate is made of the number of people that could be sustained without reactive nitrogen from the Haber-Bosch process (long dashed line), also expressed as a percentage of the global population (short dashed line). The recorded increase in average fertilizer use per hectare of agricultural land (blue symbols) and the increase in per capita meat production (green symbols) is also shown.



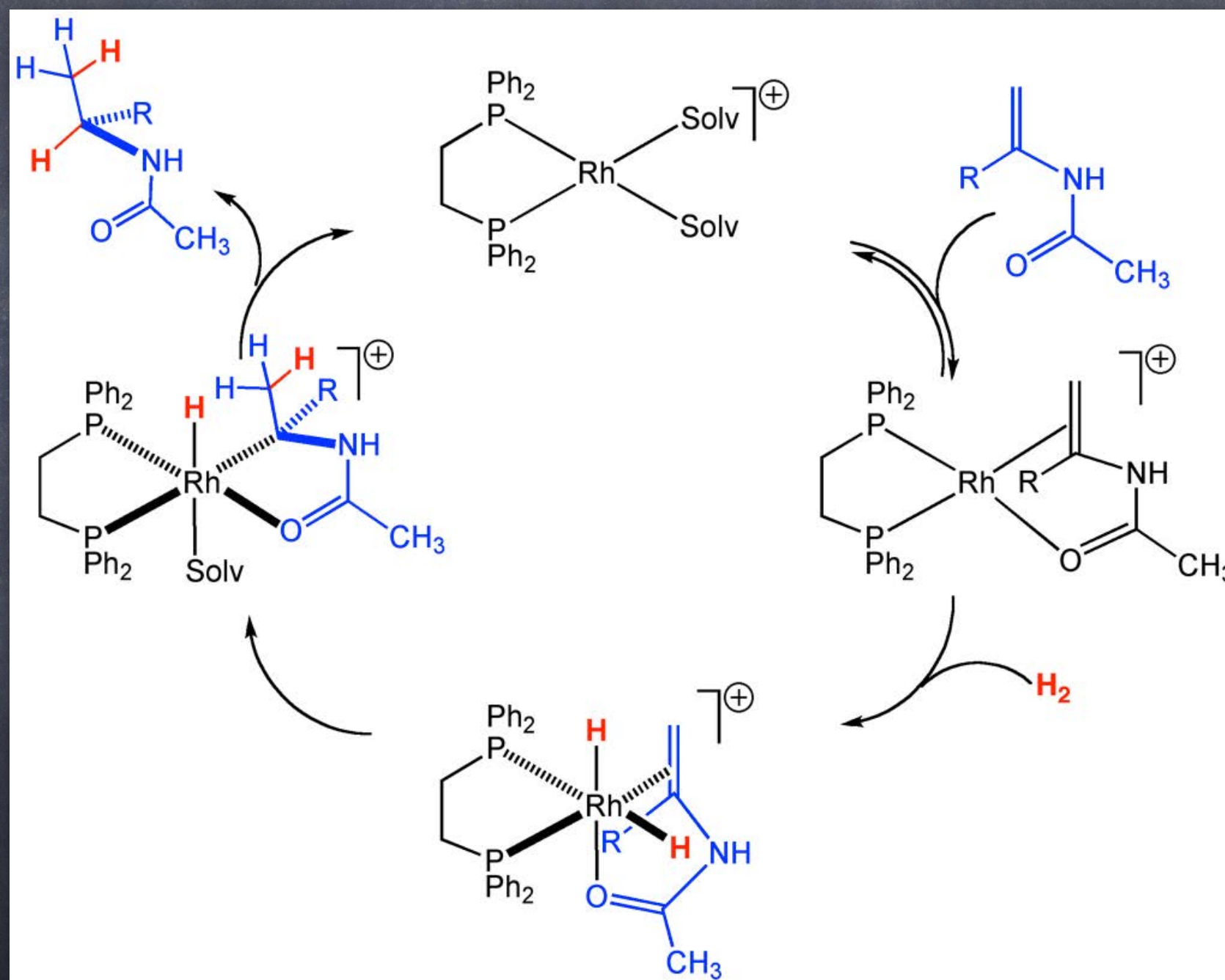
A Tiger in your Engine and...
A Catalyst in your Exhaust Pipe

Ertl Nobel Prize 2007 for Heterogeneous Catalysis

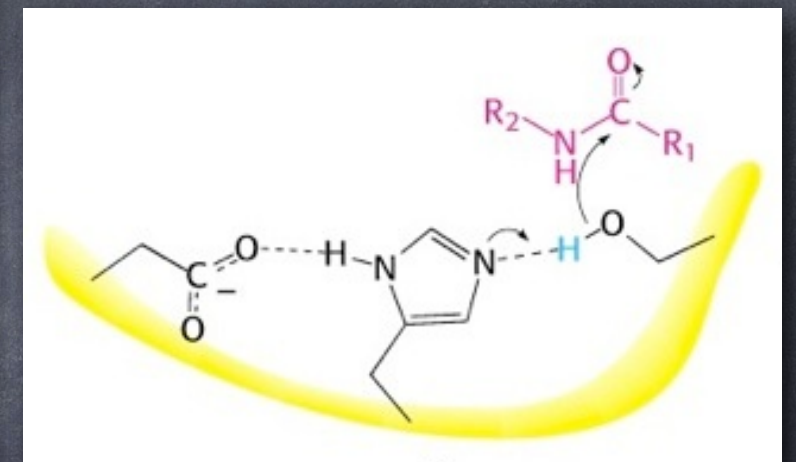
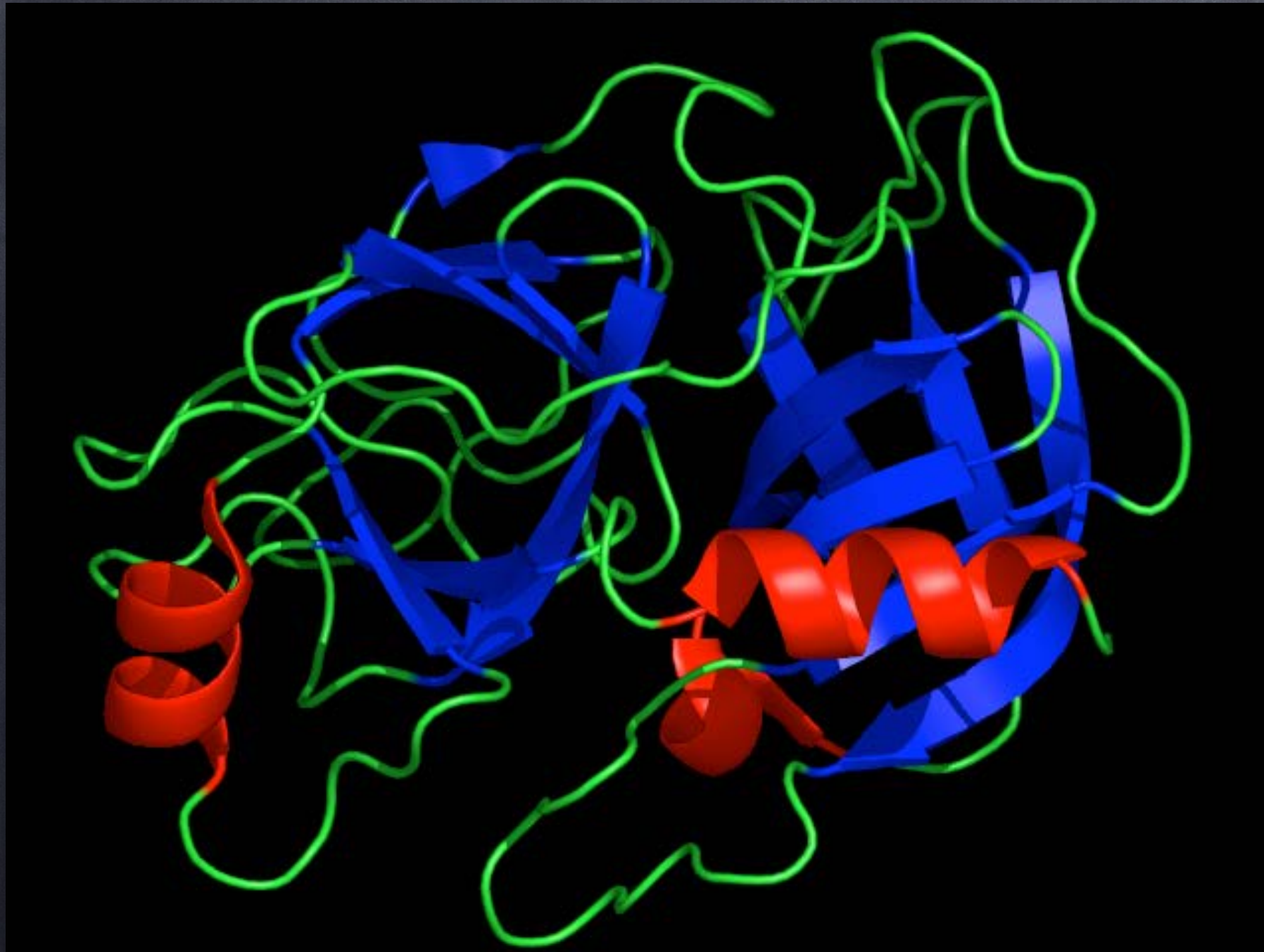


Homogeneous Catalysis: The Elegance of a Man-Made Catalyst

Three Nobel Prizes since 2001

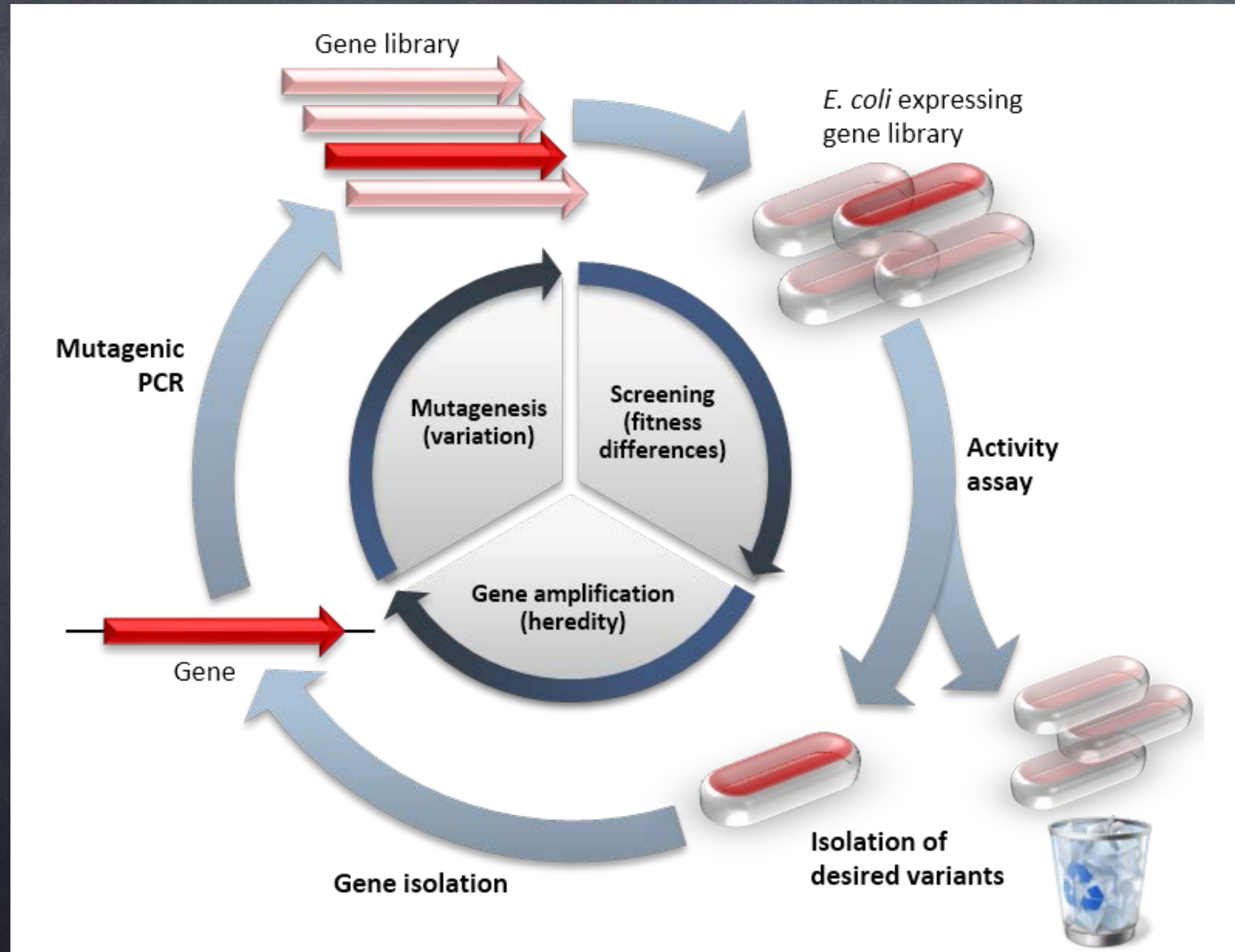


Subtilisin: A Serine Protease Used in your Laundry- and Dishwashing Powder

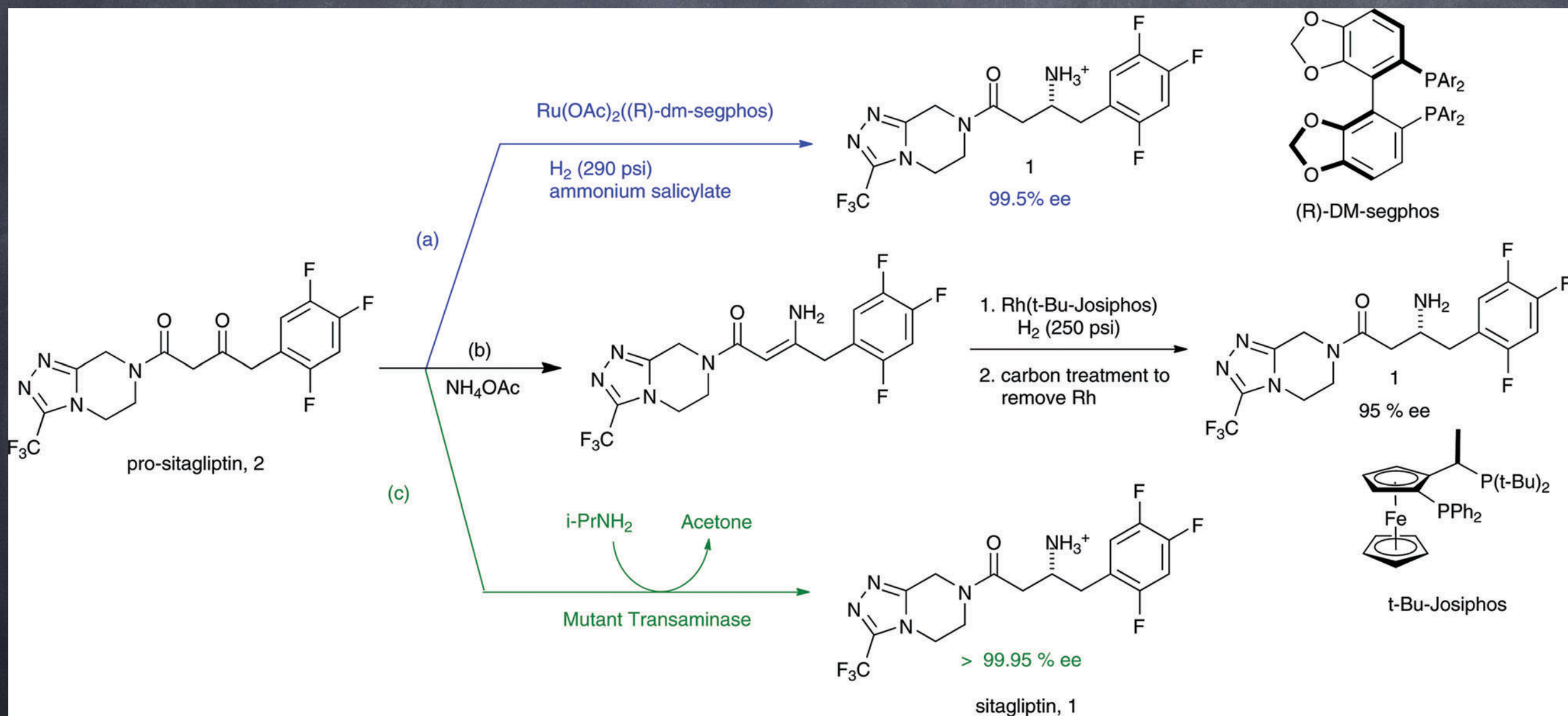


900 Tons/Year in EU

2018 Nobel Prize: Frances Arnold (1/2) for Directed Evolution



Sitagliptin (Merck): Two Green Chemistry Awards



Homogeneous- vs. Enzymatic Catalysis

Homogeneous

Enzymatic

Enantiomers

Both

Single

Solvent Tolerance

Organic

Aqueous

Substrate Specificity

Broad

Narrow

Optimization

Chemical

Genetic

Catalyst Lifetime

Limited

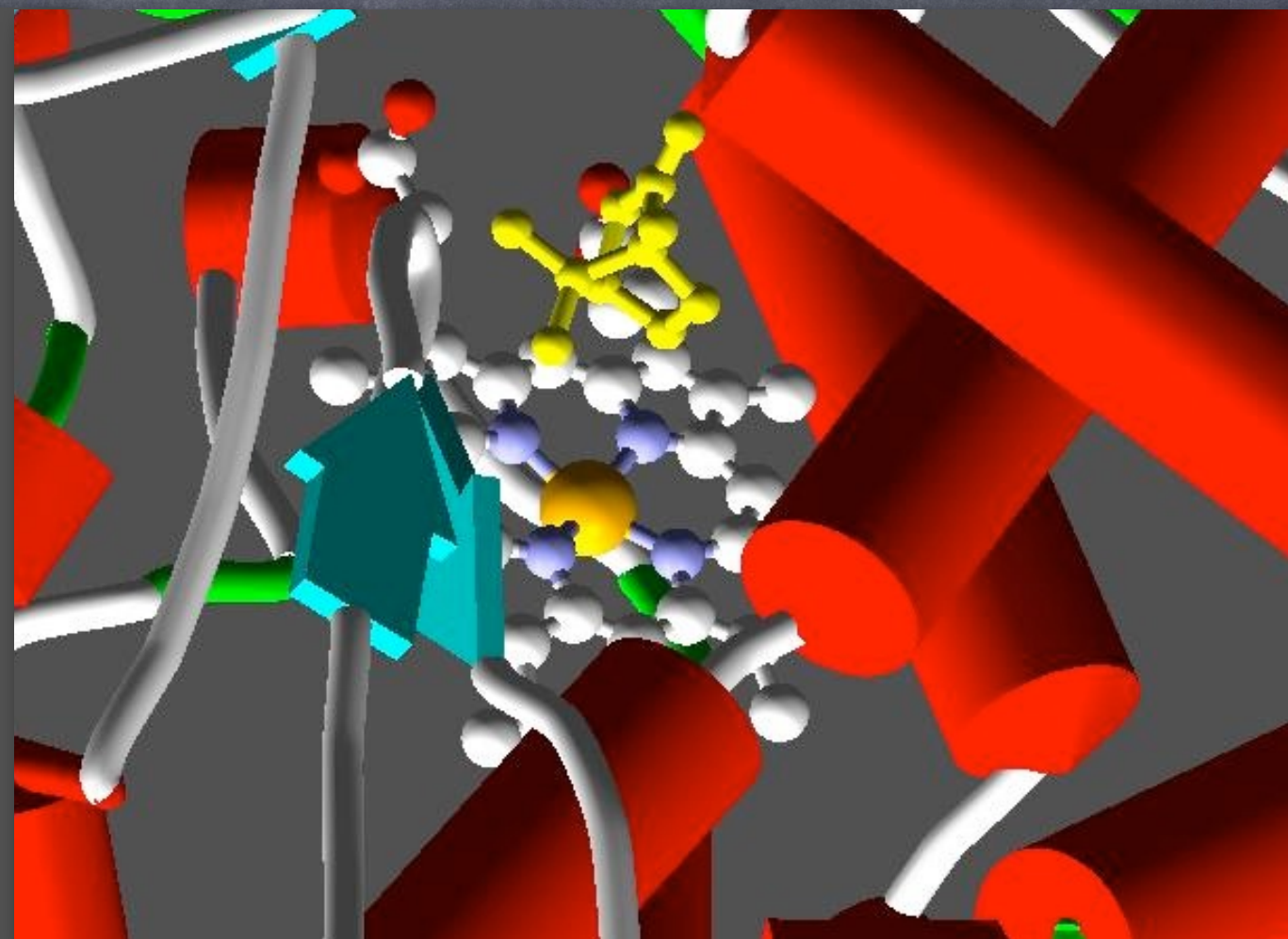
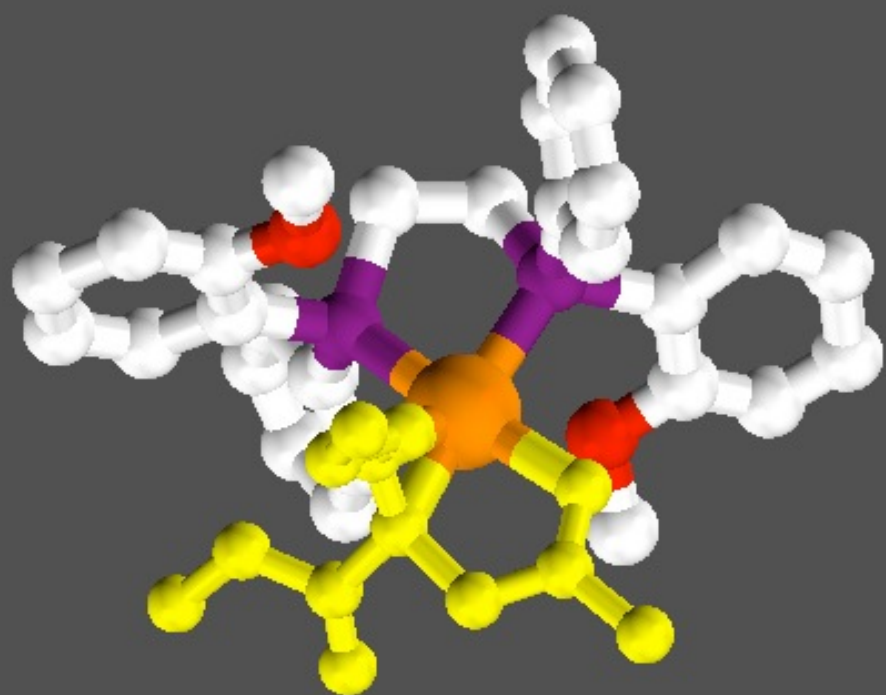
Extended

In vivo Compatibility

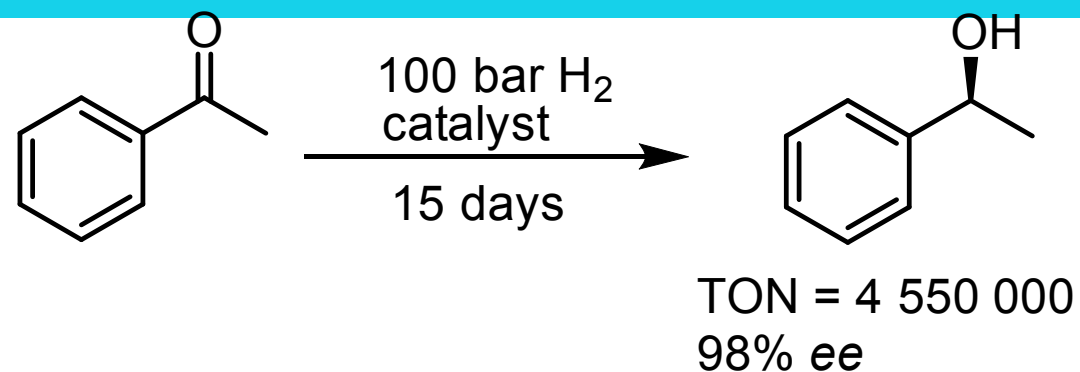
Limited

Excellent

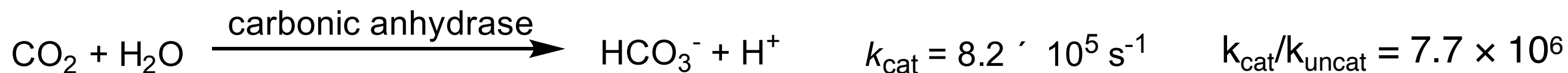
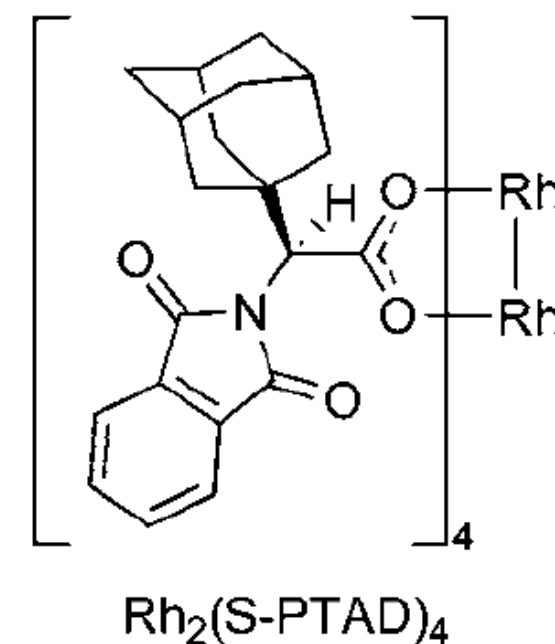
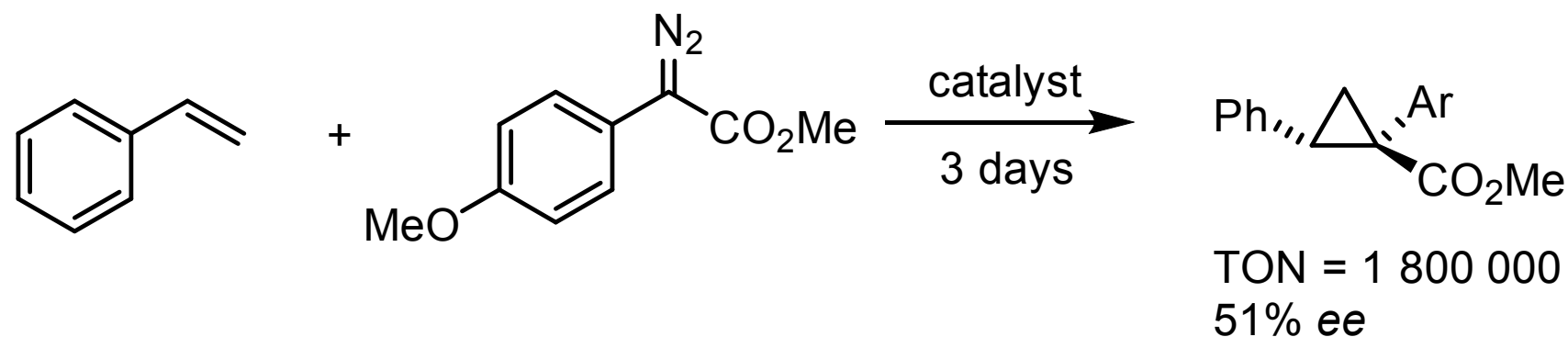
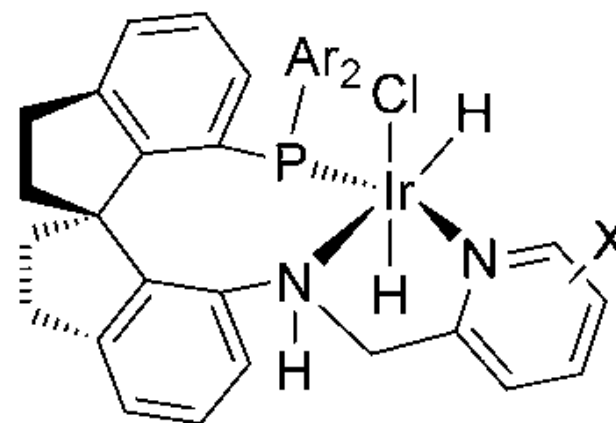
Second Coordination Sphere Environments



what is a good catalyst? a few benchmarks



cat.



Q. L. Zhou, Angew. Chem. Int. Ed. 2011, 50, 7329.

H. M. L. Davies, Chem. Sci. 2010, 1, 254.

A. Fersht >Structure and Mechanism in Protein Science<.

what is the motivation?

ii) make better catalysts (more efficiently)

rate

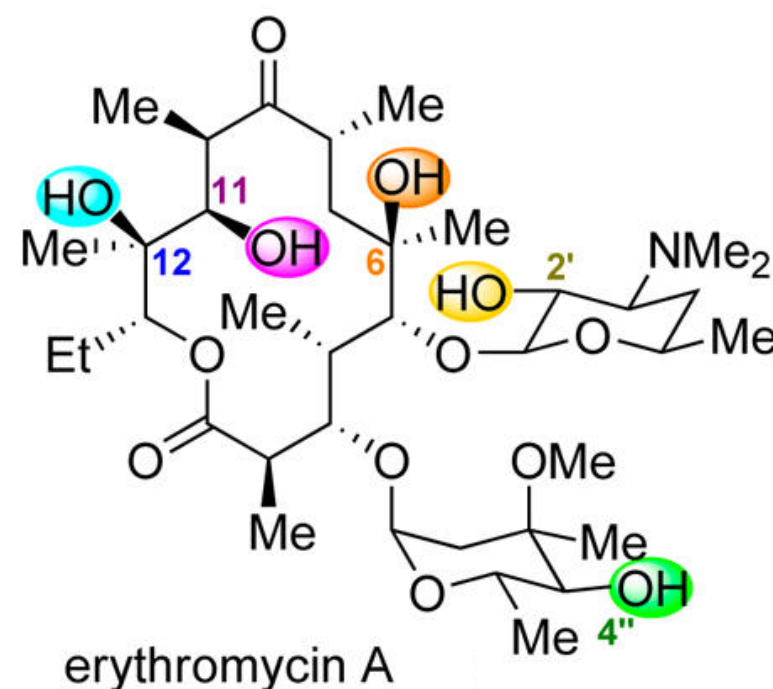
robustness

selectivity (chemo-, regio-, and stereoselectiv

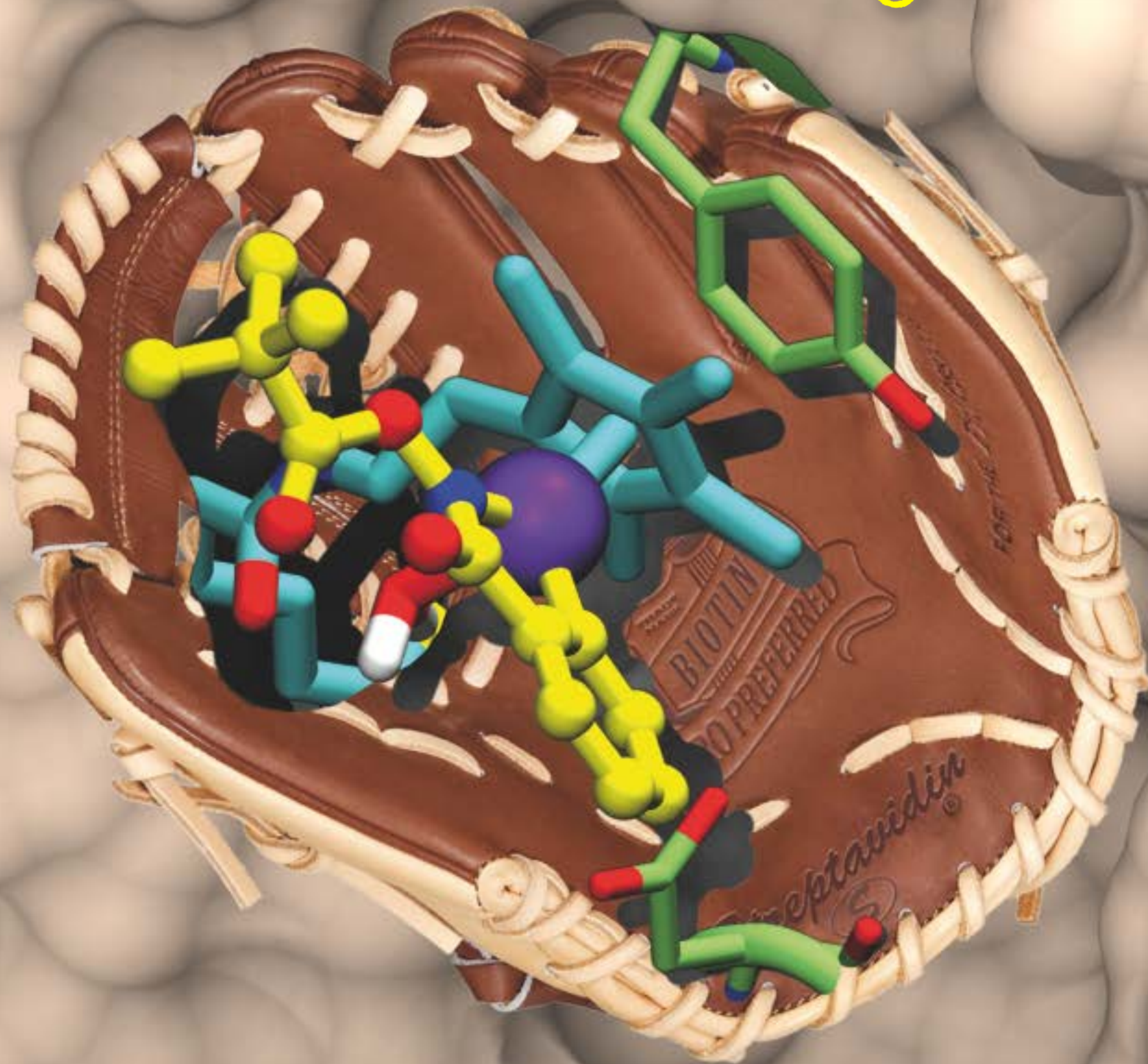
compatibility (with other catalysts, biological environment)

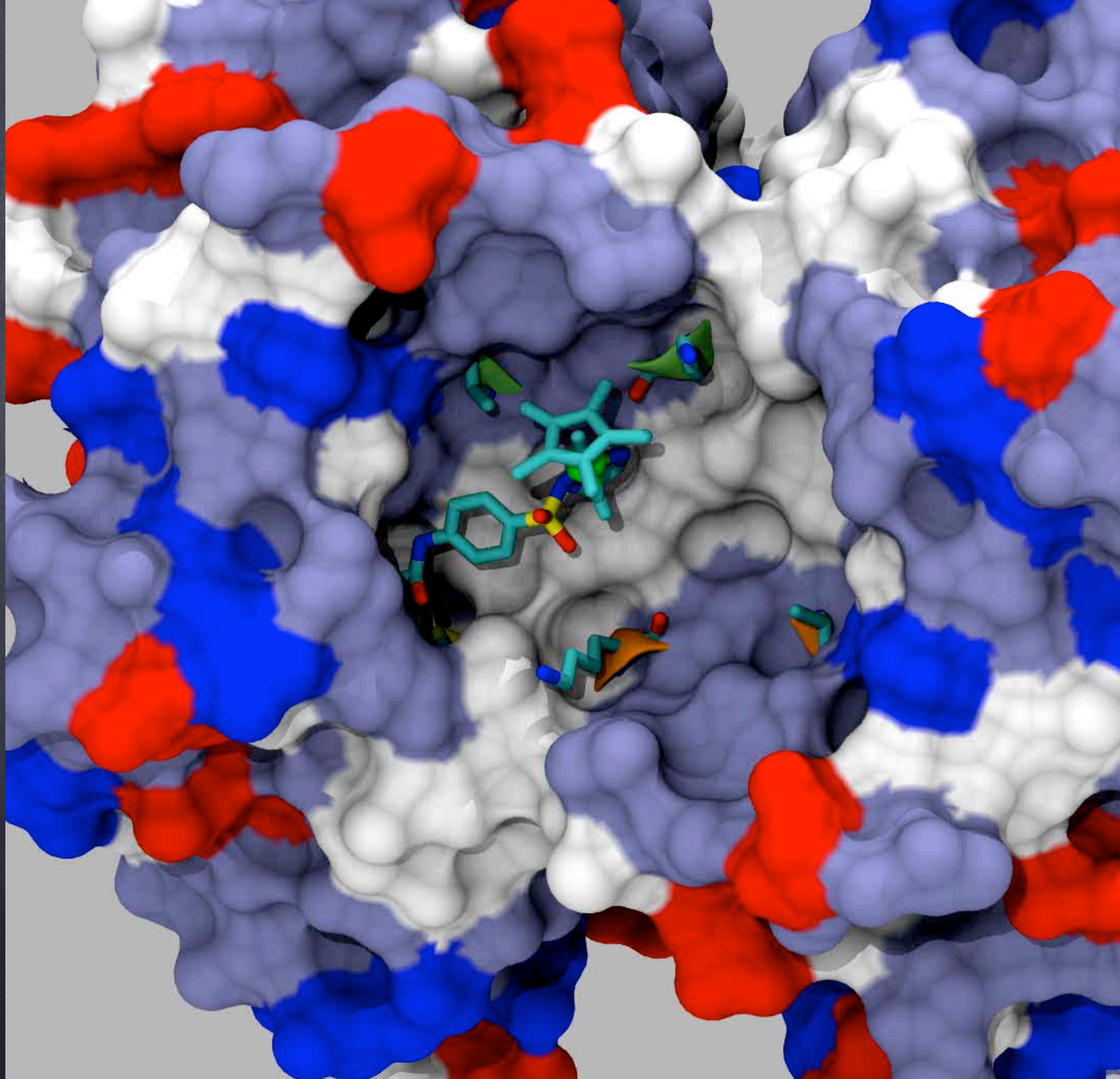
accessibility (e.g. hydrogenases)

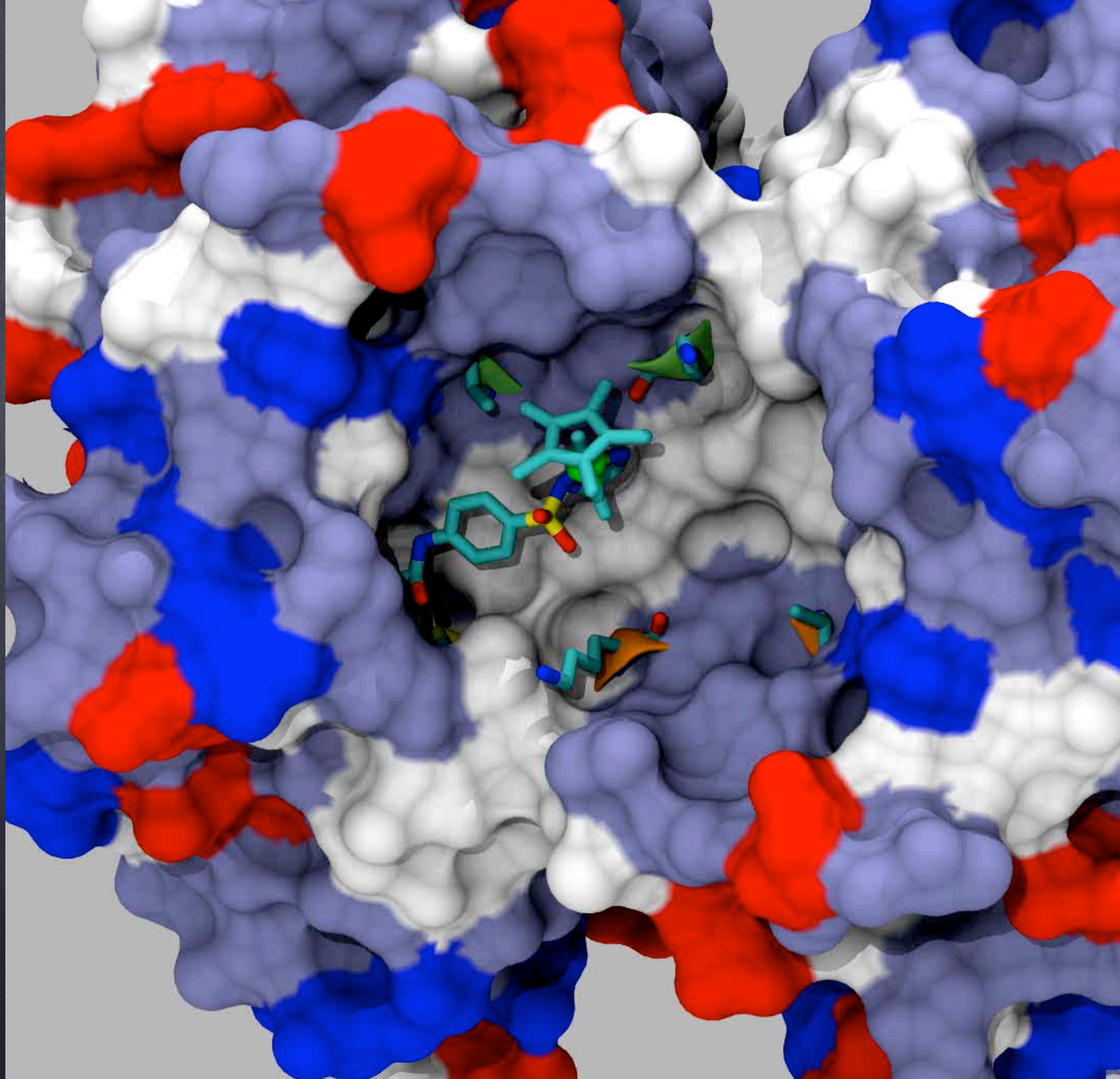
controllability (allostery)



Proteins as Host for Abiotic Cofactors: Artificial Metalloenzymes?







The Periodic Table of The (Metalloenzymes) Elements

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11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 B Boron 10.811	14 C Carbon 12.011	15 N Nitrogen 14.007	16 O Oxygen 15.999	17 F Fluorine 18.998	18 Ne Neon 20.180											
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.61	33 As Arsenic 74.922	34 Se Selenium 78.09	35 Br Bromine 79.904	36 Kr Krypton 84.80	
37 Rb Rubidium 84.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.907	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.29	
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [208.982]	85 At Astatine 209.987	86 Rn Radon 222.018	
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103	104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [269]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [269]	111 Rg Roentgenium [272]	112 Cn Copernicium [277]	113 Uut Ununtrium unknown	114 Fl Flerovium [289]	115 Uup Ununpentium unknown	116 Lv Livermorium [298]	117 Uus Ununseptium unknown	118 Uuo Ununoctium unknown	
		Lanthanide Series	57 La Lanthanum 138.906	58 Ce Cerium 140.115	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.966	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.26	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967	
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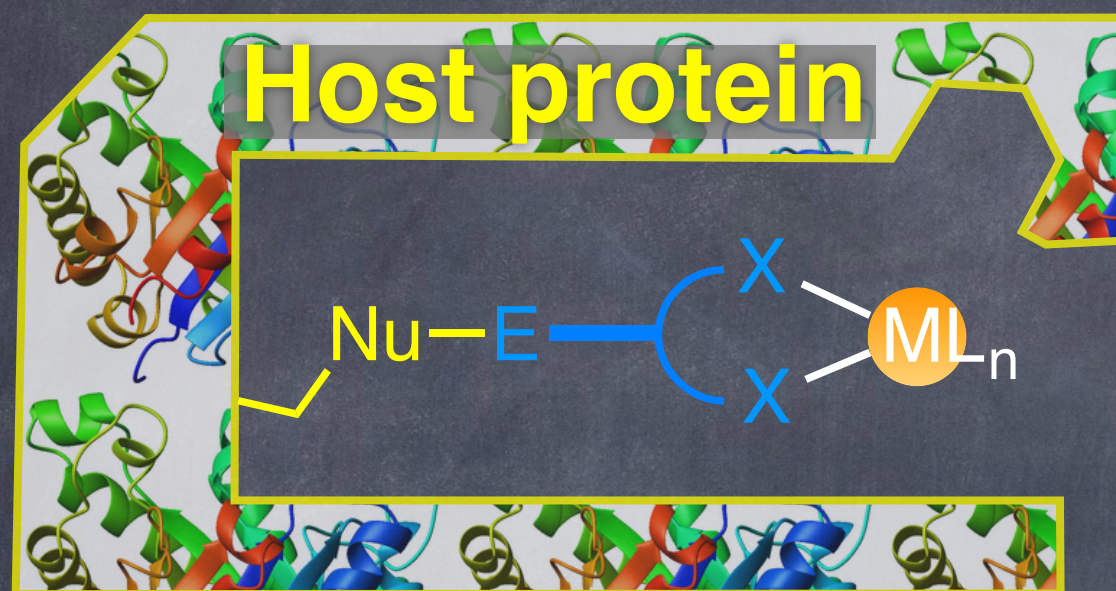
The Periodic Table of The (Metalloenzymes) Elements

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11 Na Sodium 22.990	12 Mg Magnesium 24.305	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8 VIII 8			9 VIII 9	10 VIII 10	11 IB 1B	12 IIB 2B	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.066	17 Cl Chlorine 35.453	18 Ar Argon 39.948				
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.61	33 As Arsenic 74.922	34 Se Selenium 78.09	35 Br Bromine 79.904	36 Kr Krypton 84.80						
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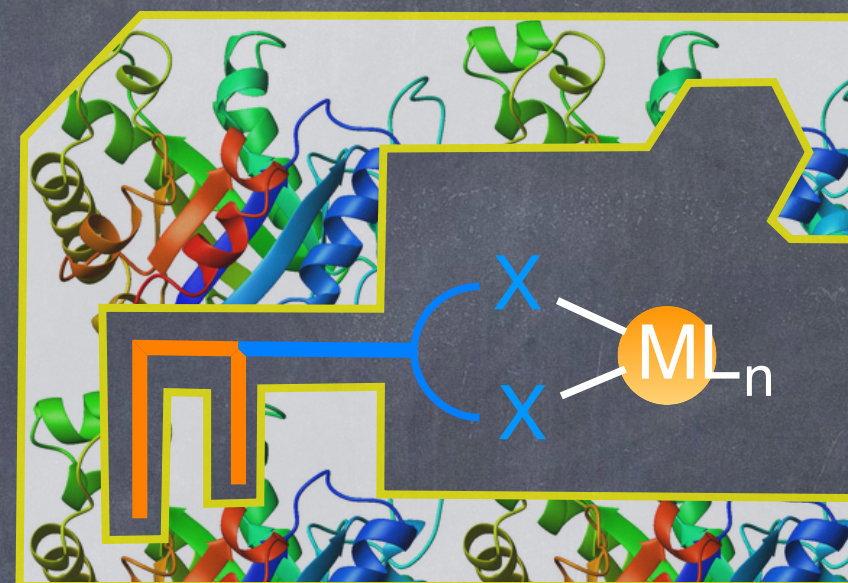
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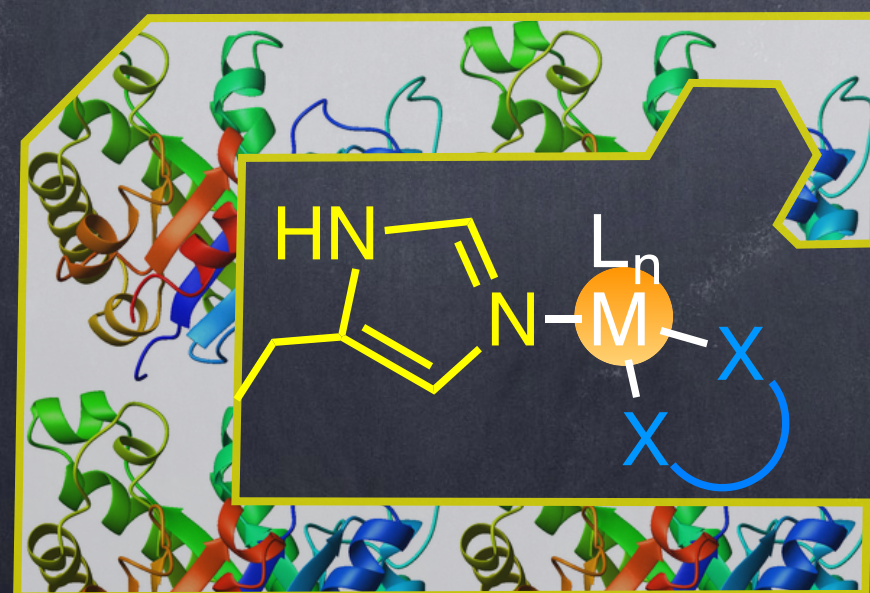
Anchoring of the Catalyst: Four Alternatives to Ensure Localization



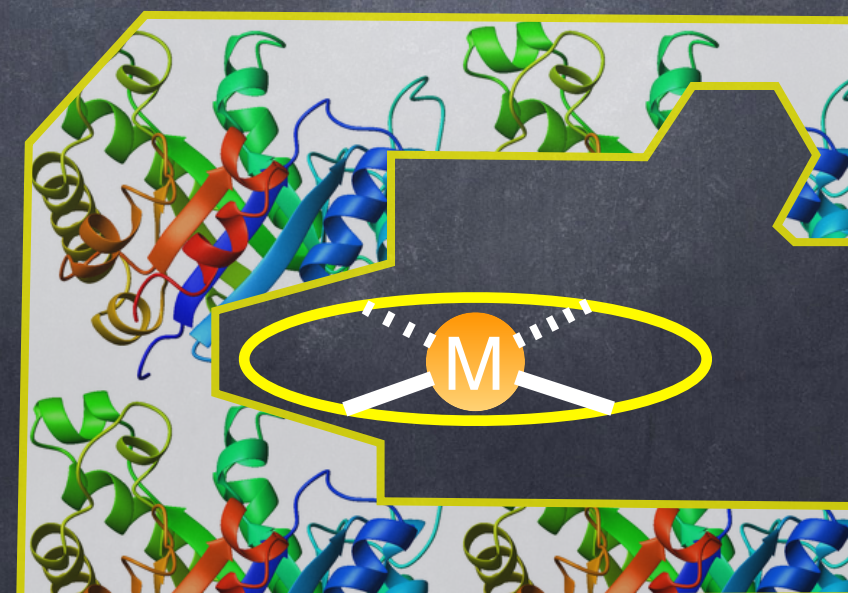
Covalent anchoring



Supramolecular anchoring

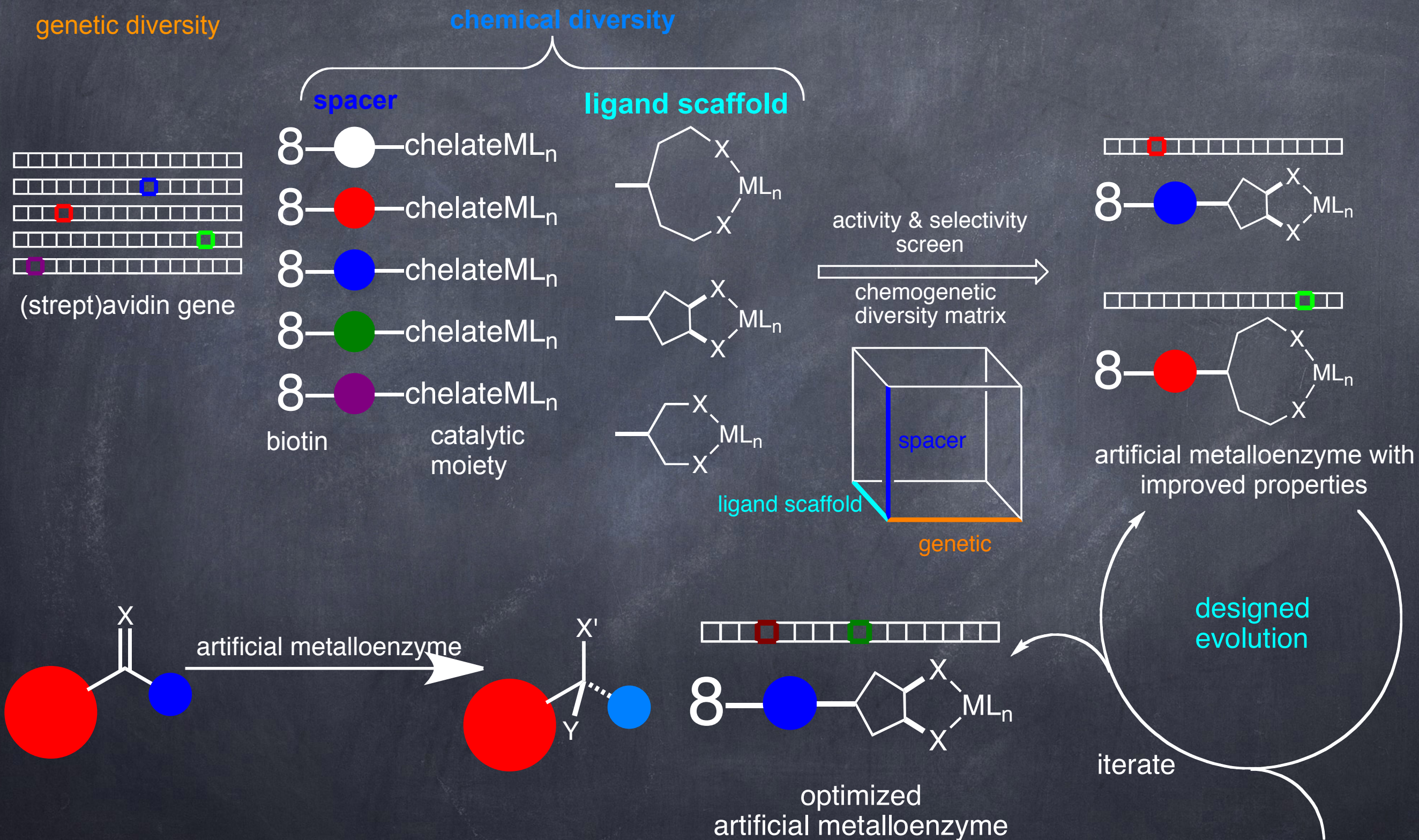


Dative anchoring



Metal substitution

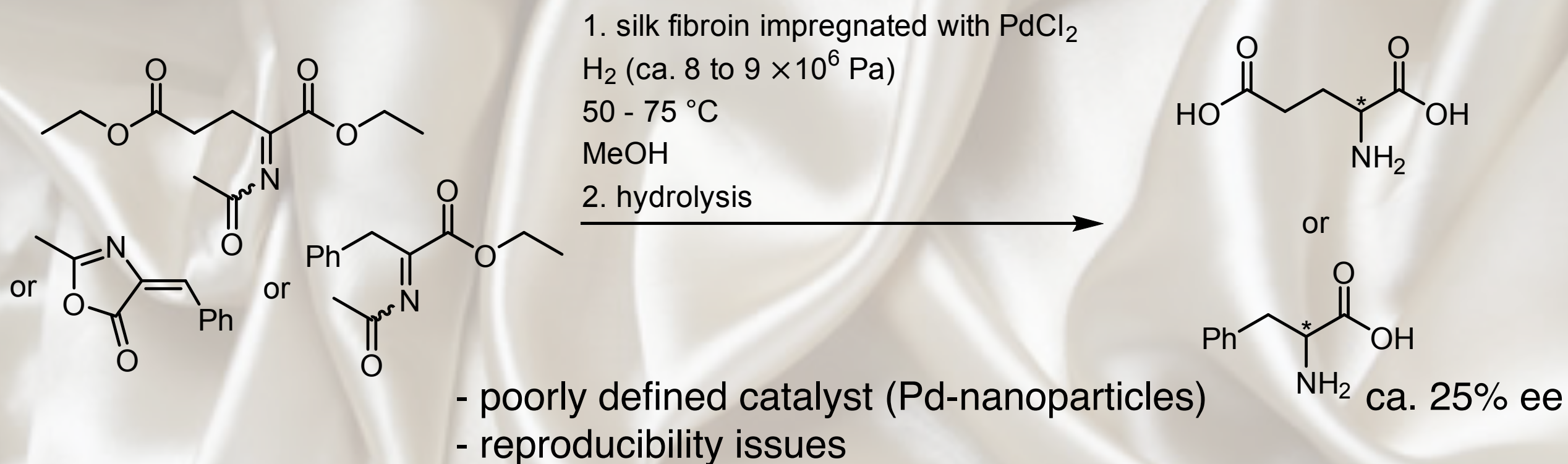
ArMs: Chemo-Genetic Optimization Strategy



standing on the Shoulders of
Giants:
A Historical Perspective

historical perspective – a few milestones

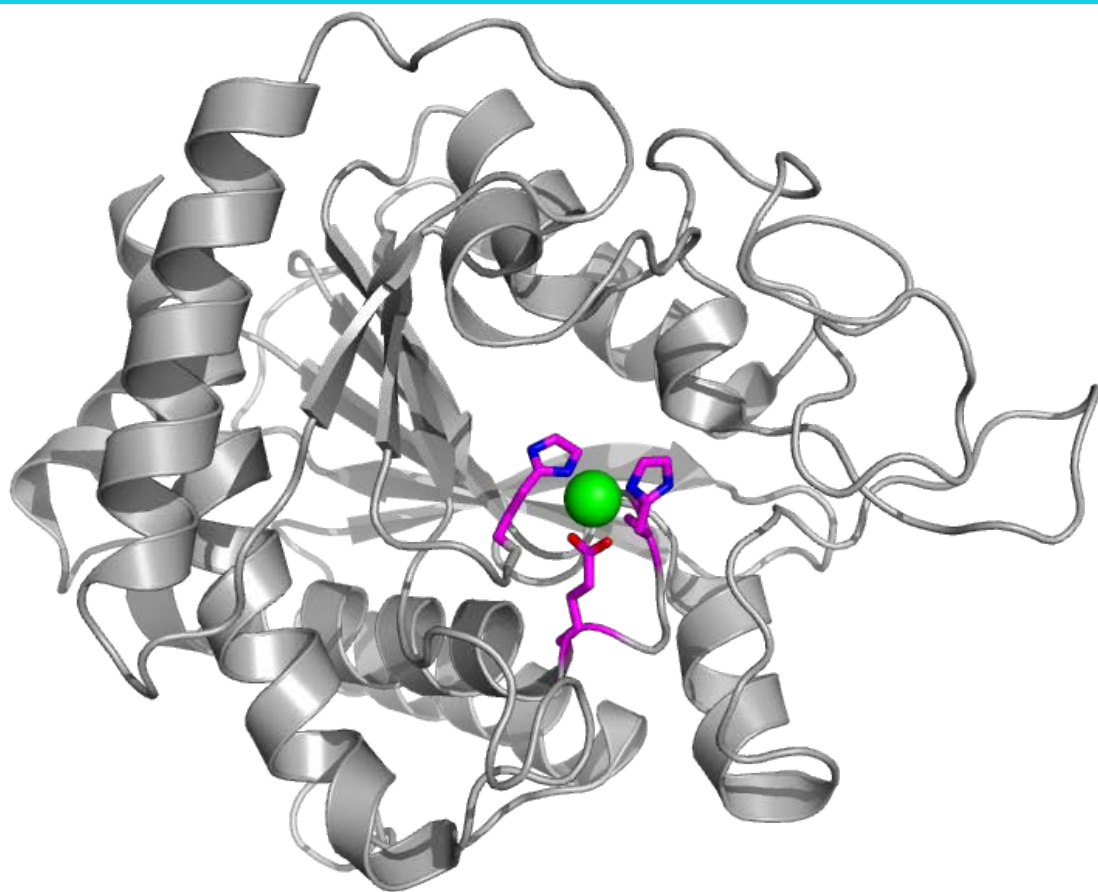
artificial metalloenzymes were at the beginning of asymmetric transition metal catalysis by chemical means! Akabori et al. 1956



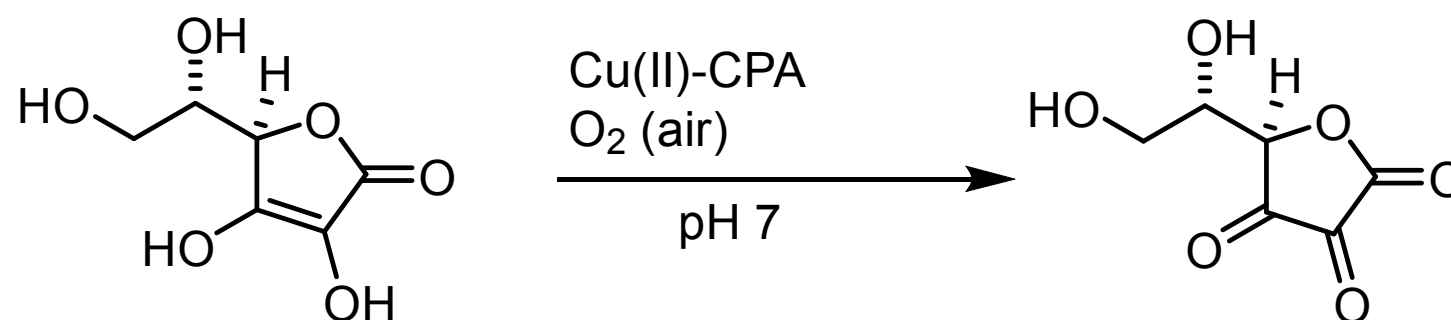
S. Akabori et al. >An Asymmetric Catalyst< Nature 1956, 178, 323-324.

H. B. Kagan in >Compr. Asym. Catal.< 2000, Eds. A. Pfaltz/E.N. Jacobsen, Springer Berlin

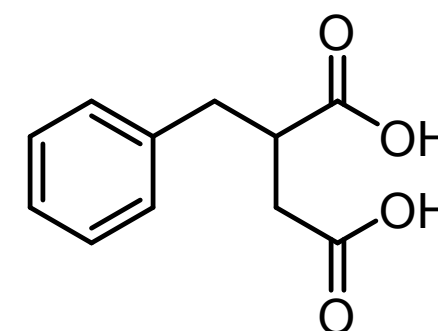
non-native reaction after metal exchange: E. T. Kaiser (1976)



carboxypeptidase A
PDB 1M4L



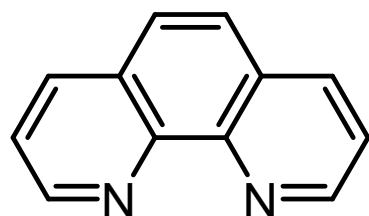
effective inhibitor



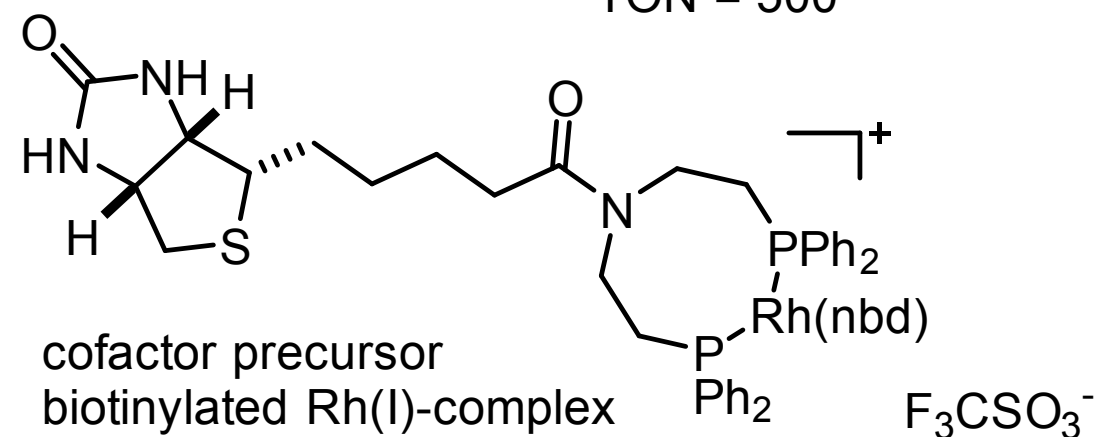
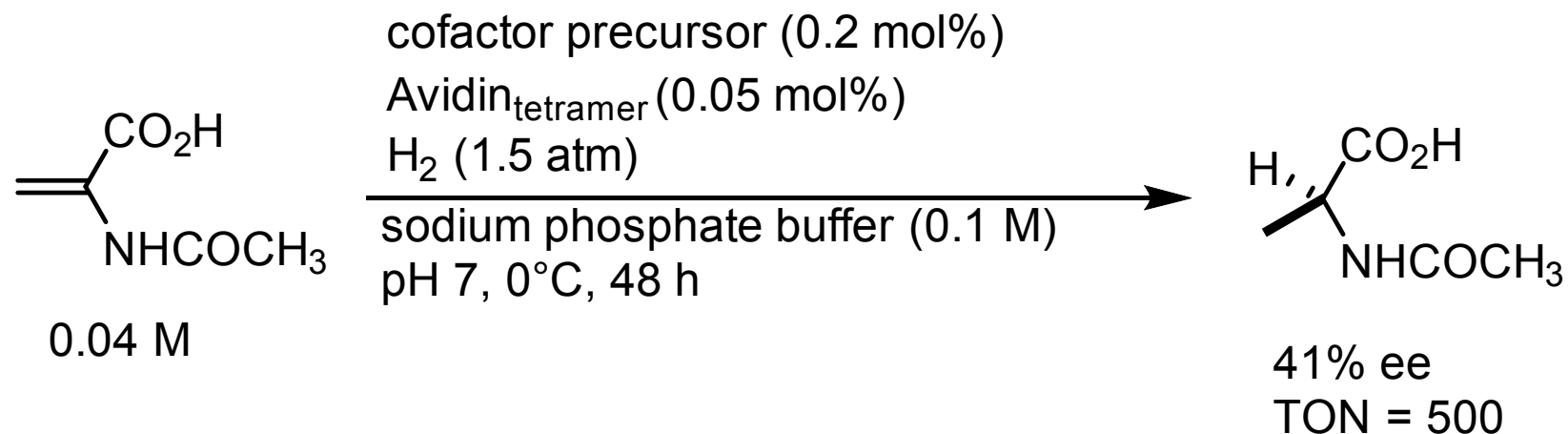
$$k_{\text{cat}} = 6 \text{ min}^{-1} \text{ and } K_{\text{m}} = 0.24 \text{ mM}$$

metal exchange:

dialysis
against:

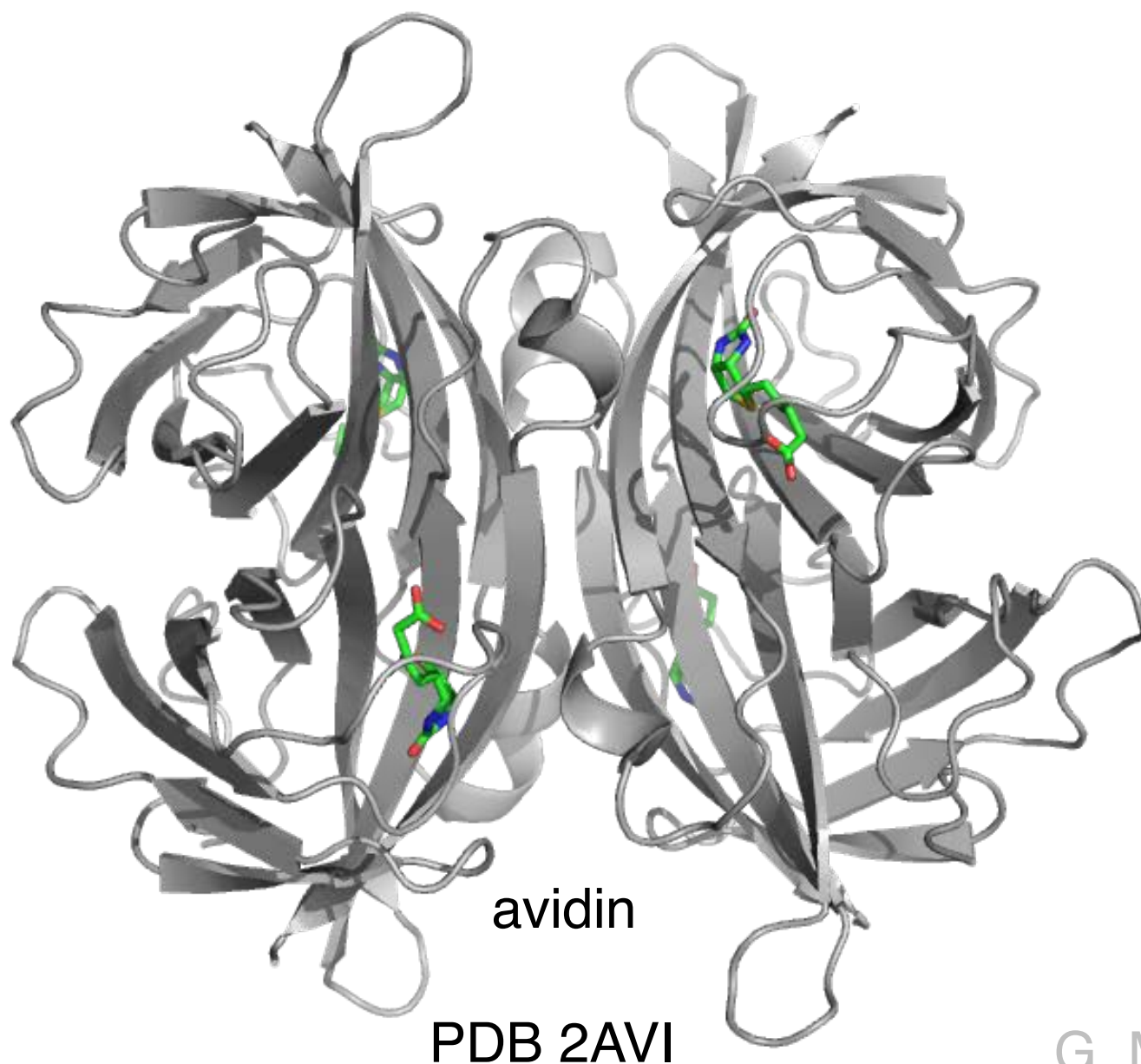


hydrogenation - Whitesides (1978)

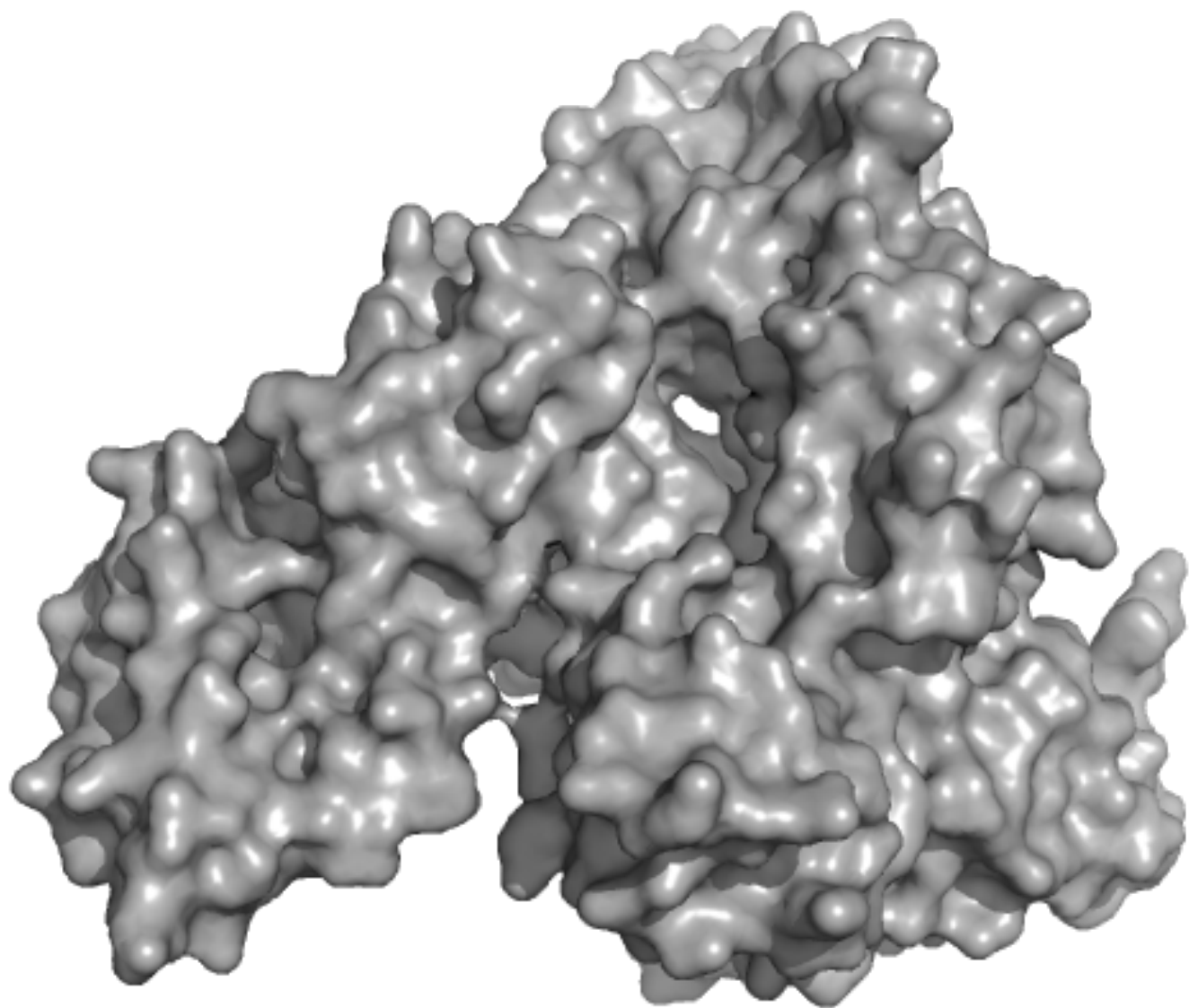
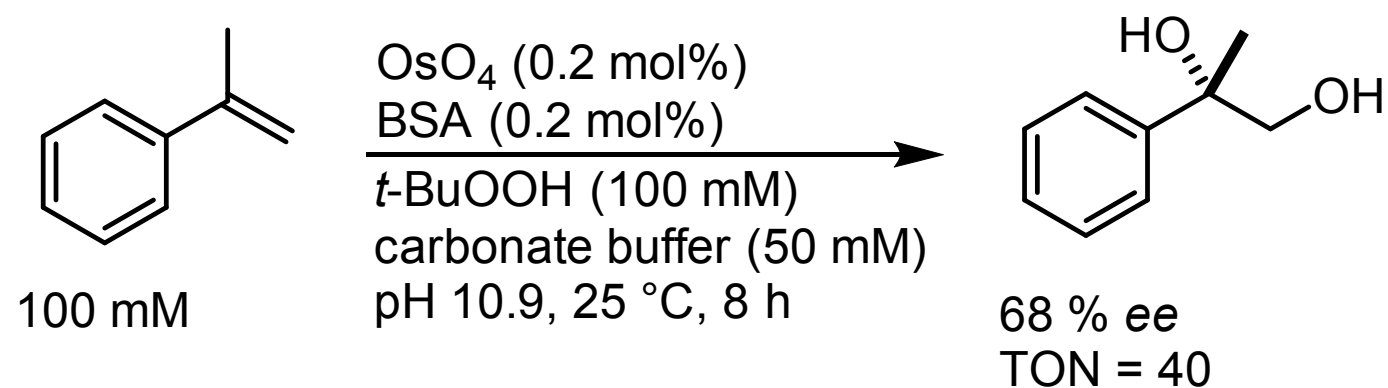


$$K_D(\text{biotin-avidin}) \approx 10^{-15} \text{ M}$$

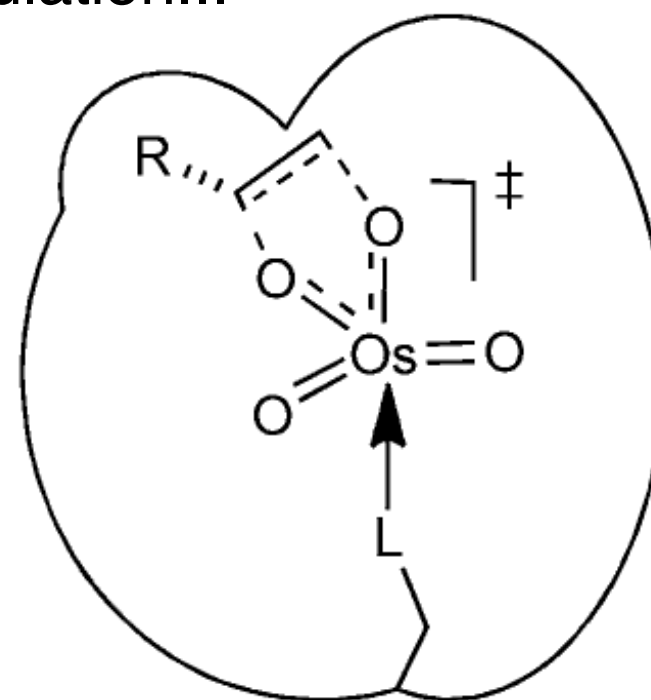
of the catalyst (CA by approximately a factor of 10). The presence of 1 equiv of avidin in solution (assuming each avidin subunit to be associated with 1 equiv of 1) resulted in a definite increase in activity, and in the production of 3 with ~40% *S* enantiomeric excess. When the ability of avidin to bind 1 was blocked by prior exposure to ether a 10% excess or a tenfold



alkene dihydroxylation. Kokubo et al. (1983)

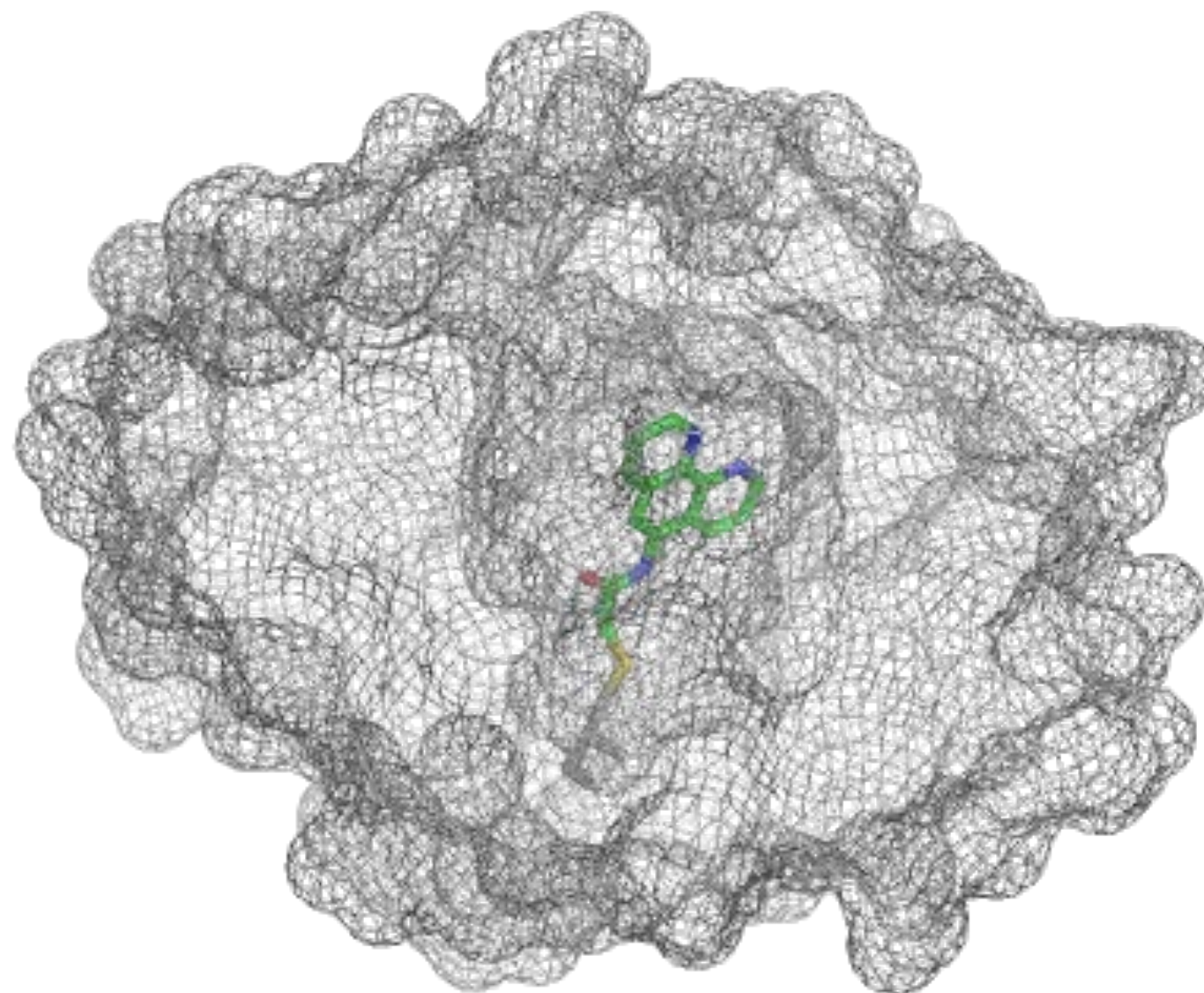
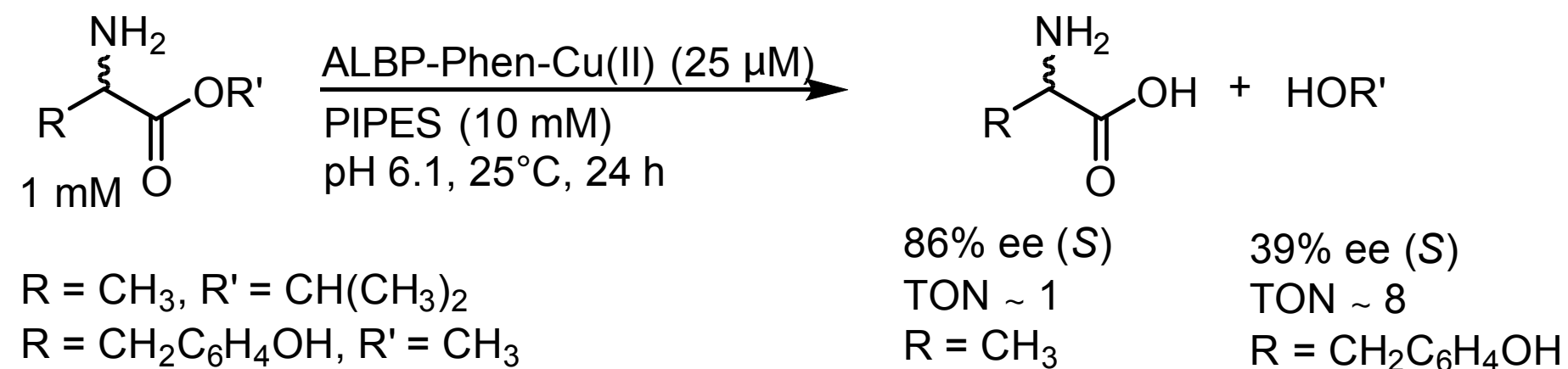


speculation...

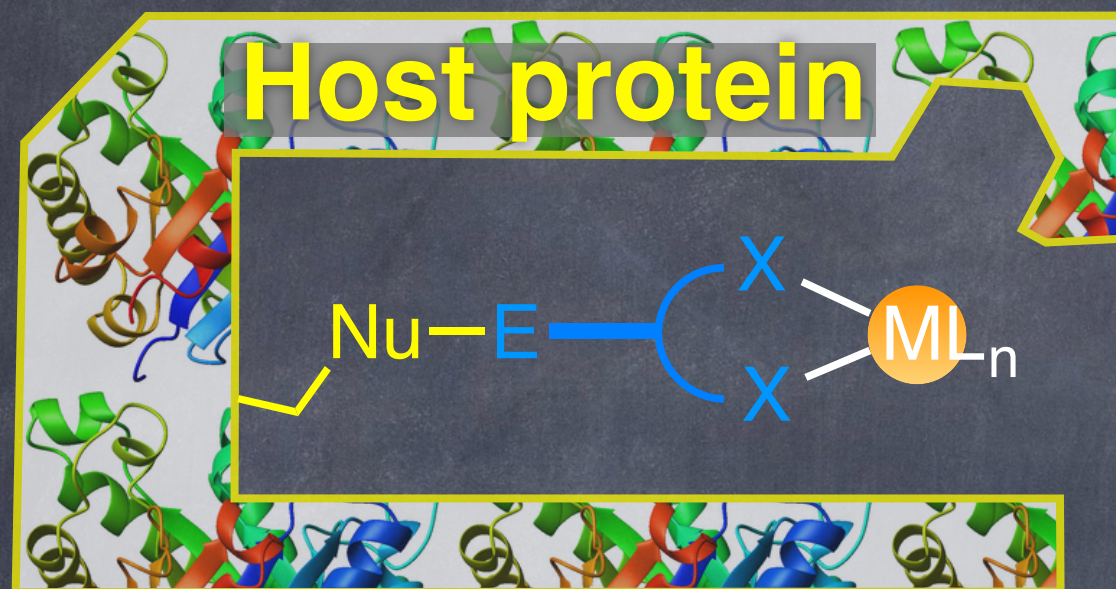


bovine serum albumin (BSA)
PDB ID 4FS5

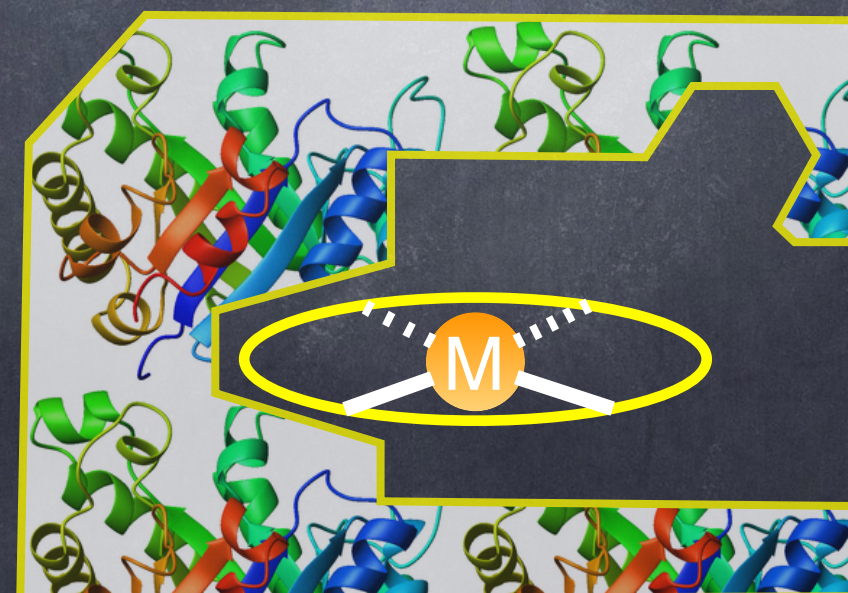
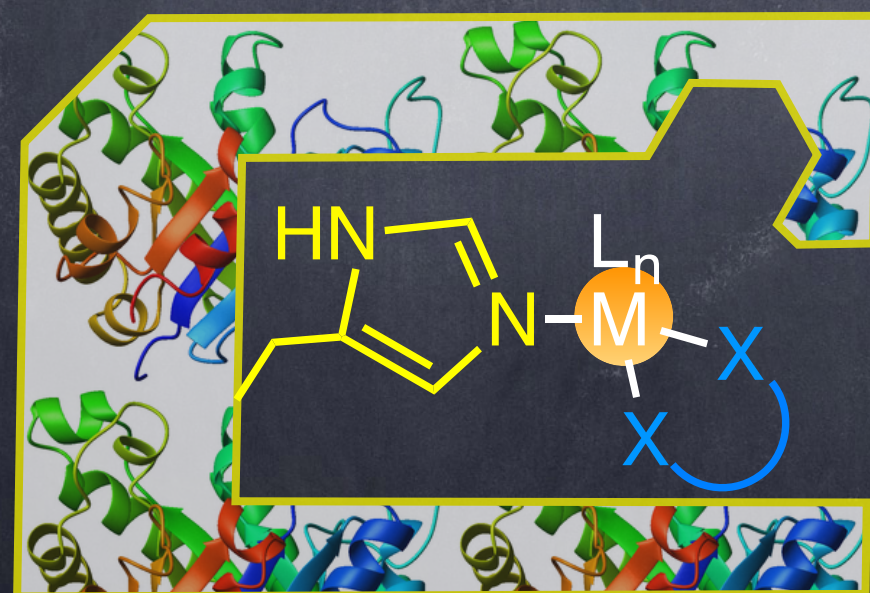
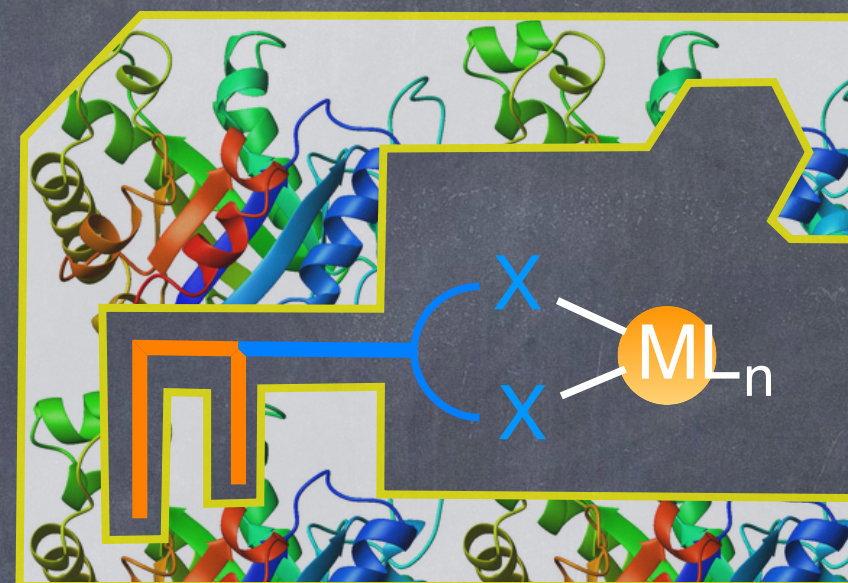
ester hydrolysis. Distefano (1997)



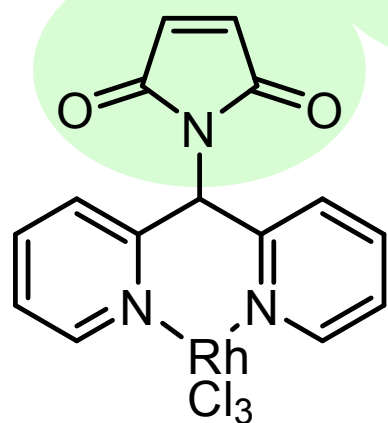
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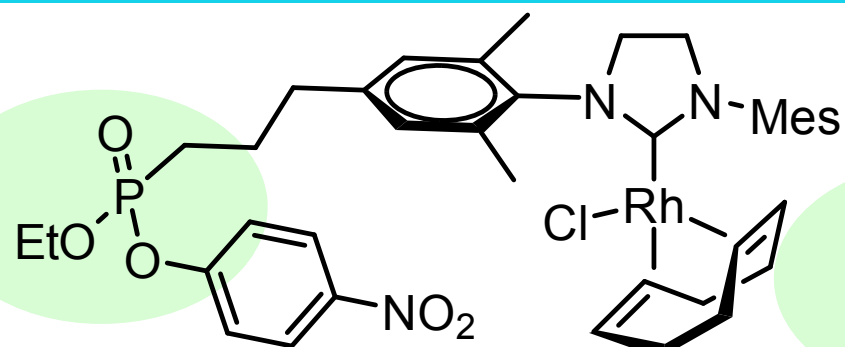
Covalent anchoring



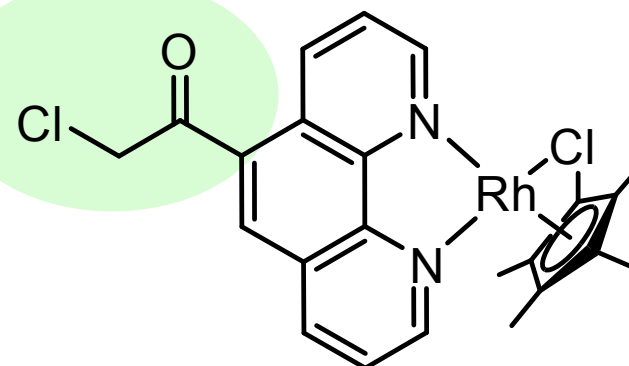
anchoring strategies – covalent anchoring



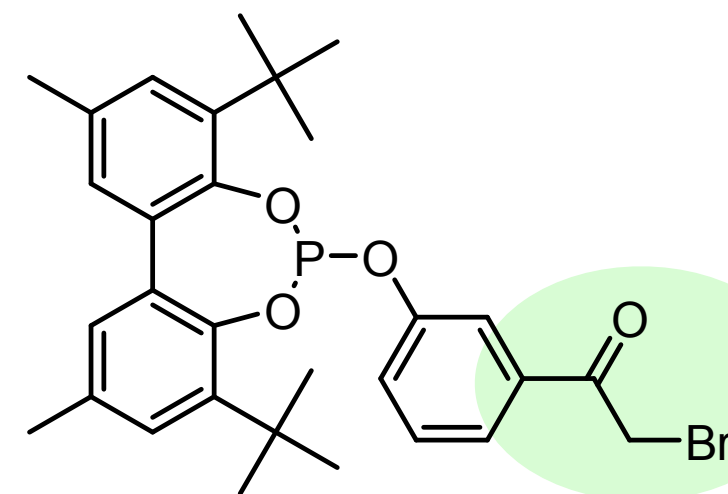
Reetz 2002



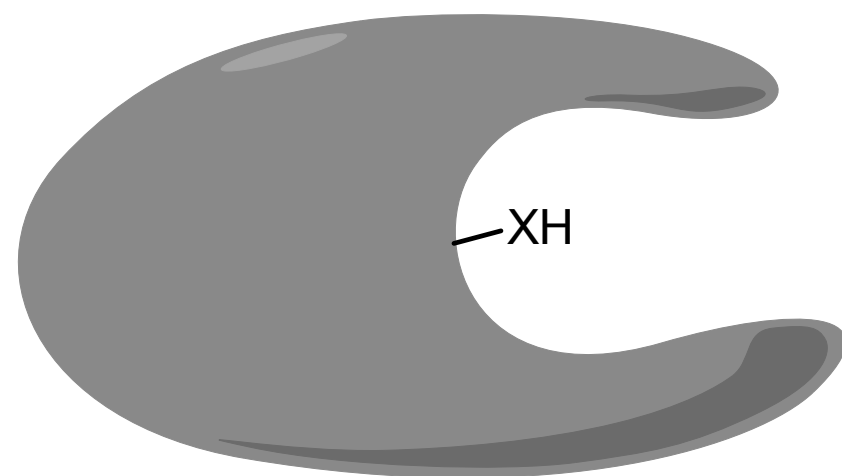
Klein Gebbink 2015



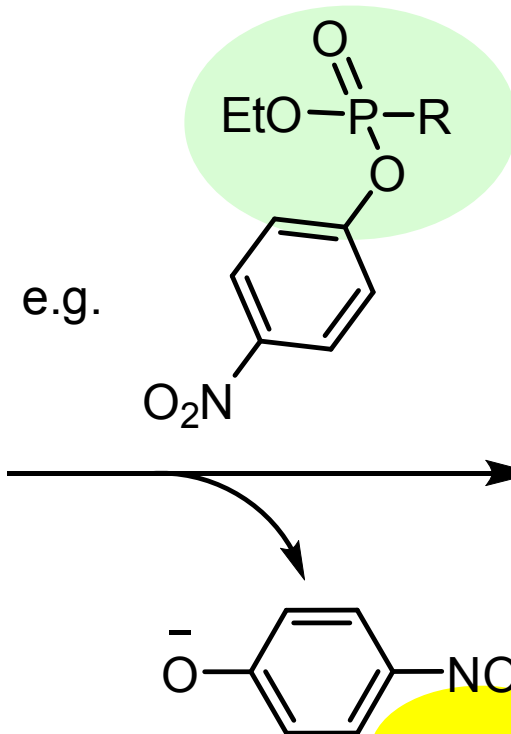
Salmain 2015



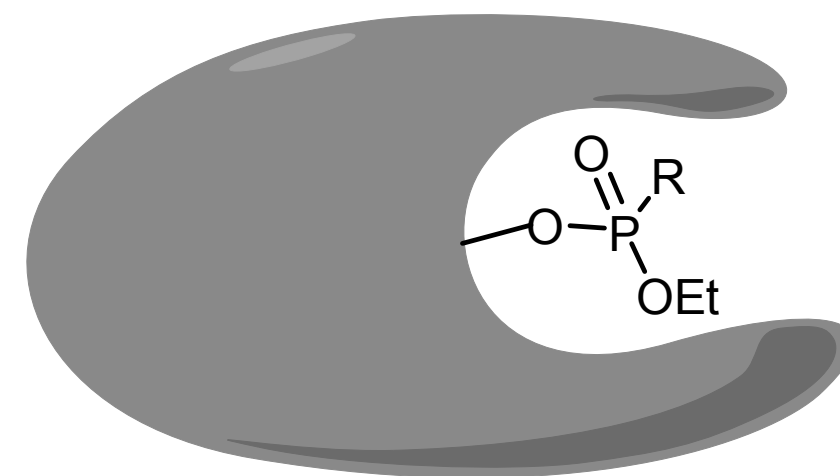
de Vries 2005



X = S, O nucleophilic group
hydrolytic enzyme
e.g. papain, cutinase,
CalB



for X = O
R = linked metal complex

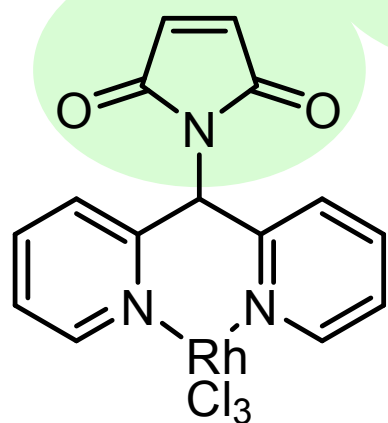


test for successful
immobilisation: inhibited
hydrolytic activity

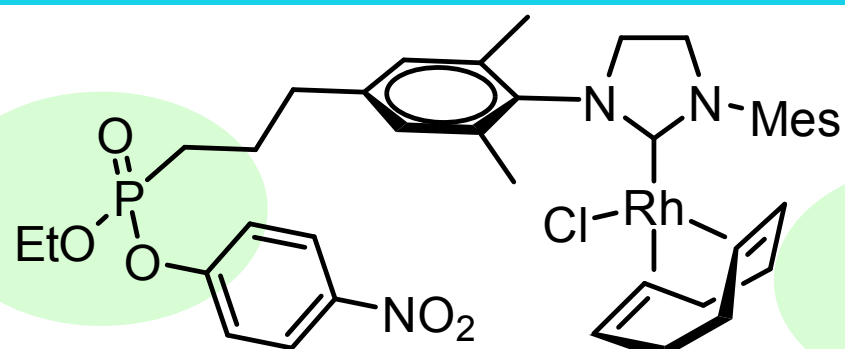
E. T. Kaiser, Biochem. Biophys. Res. Commun. 1977, 76, 64; M. T. Reetz, Chimia, 2002, 56, 721;

R. J. M. Klein Gebbink, Chem. Commun. 2015, 51, 6792; M. Salmain, J. Mol. Catal. B, 2015, 100, 214; de Vries, J. Mol. Catal. B, 2005, 5050

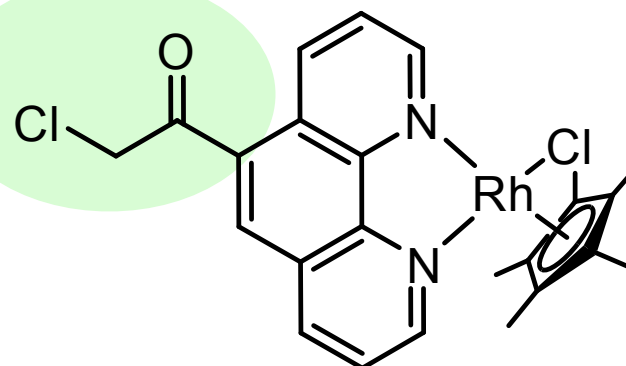
anchoring strategies – covalent anchoring



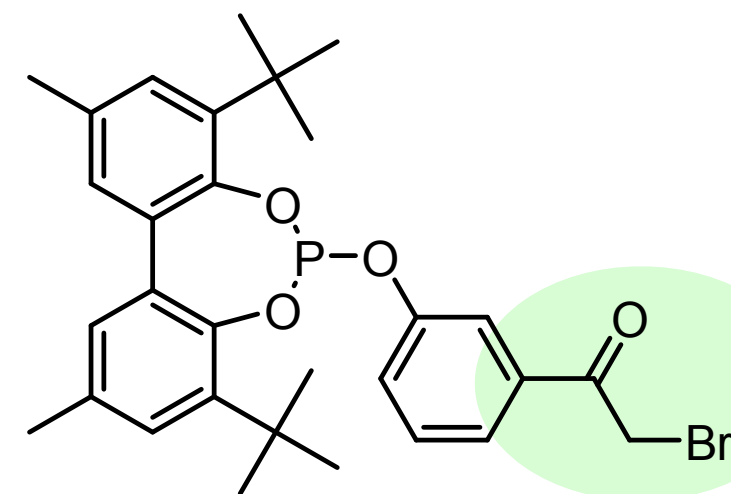
Reetz 2002



Klein Gebbink 2015



Salmain 2015



de Vries 2005

only modest enantioselectivities observed in hydrogenation and transfer hydrogenation reactions –

poorly defined catalyst environment? Too much flexibility?

E. T. Kaiser, Biochem. Biophys. Res. Commun. 1977, 76, 64; M. T. Reetz, Chimia, 2002, 56, 721;

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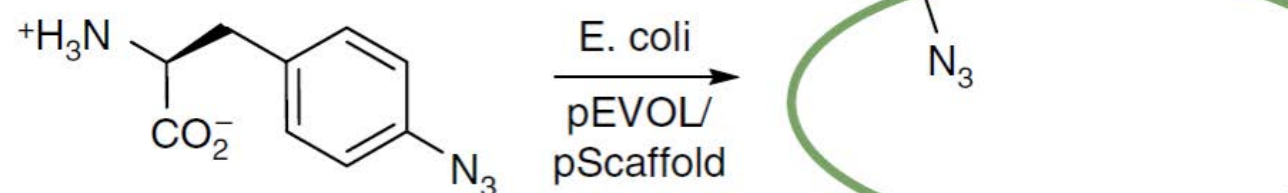
'biocompatible cofactors' – more options

purified protein

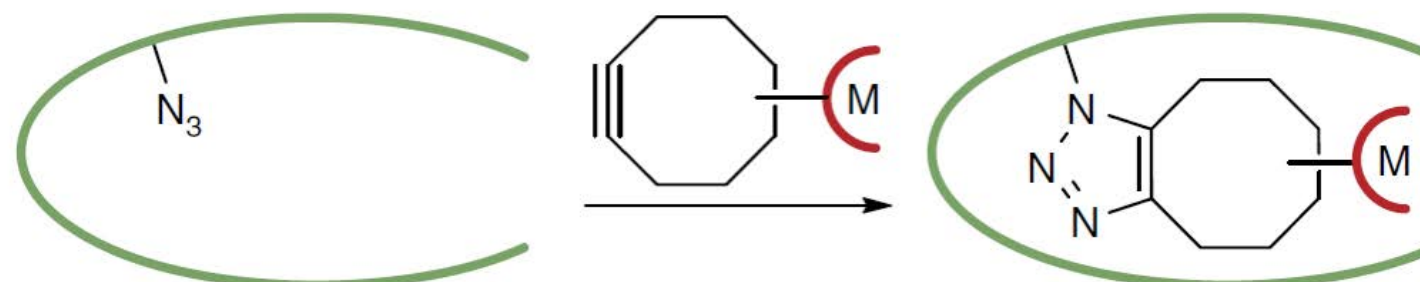
1. installing the cofactor

a

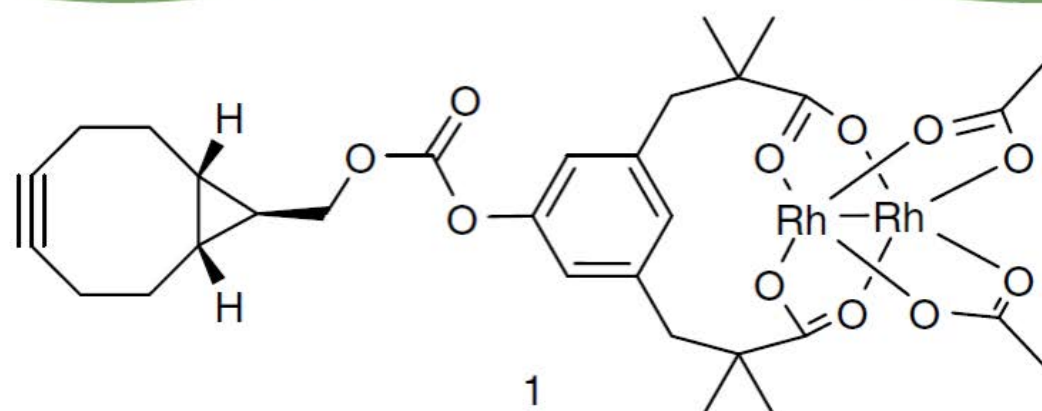
1. Scaffold expression:



2. Cofactor bioconjugation:



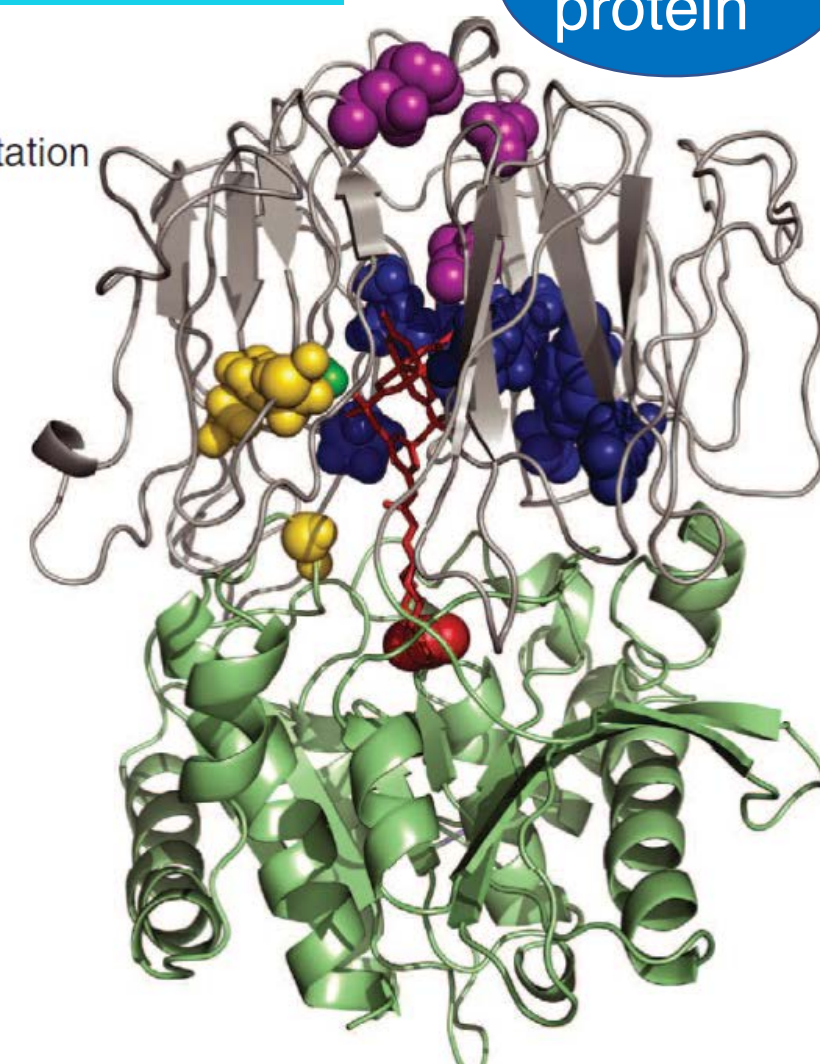
b



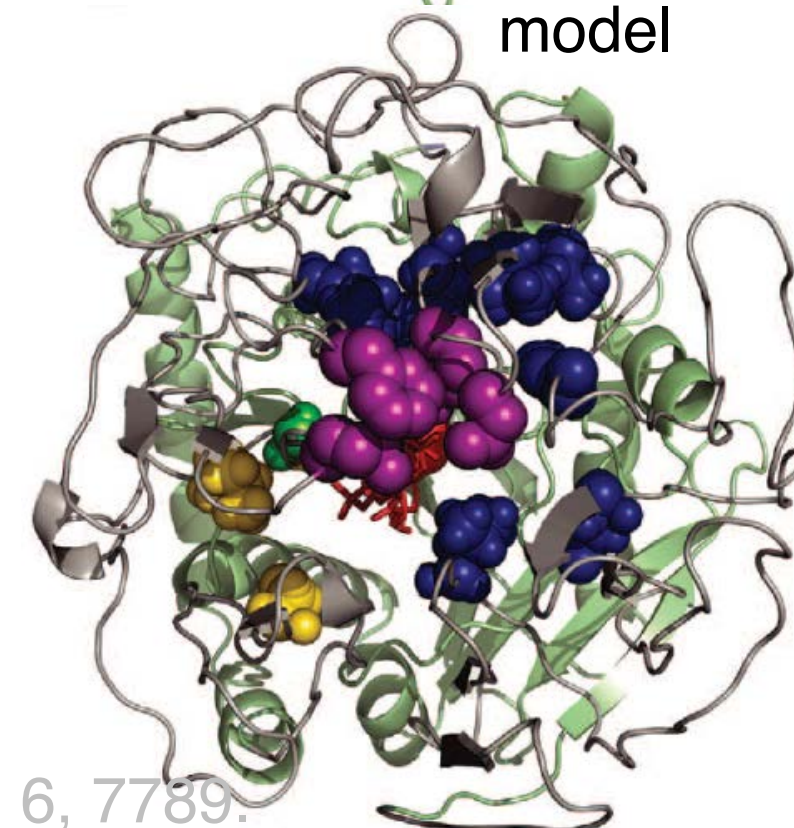
mutant identified which can bind the cofactor (4 Alanine mutations) and is active and selective (1 His mutation) in cyclopropanation

0-ZA₄

- Alanine mutations (A₄)
- Histidine mutations
- Histidine/phenylalanine mutation
- Phenylalanine mutations



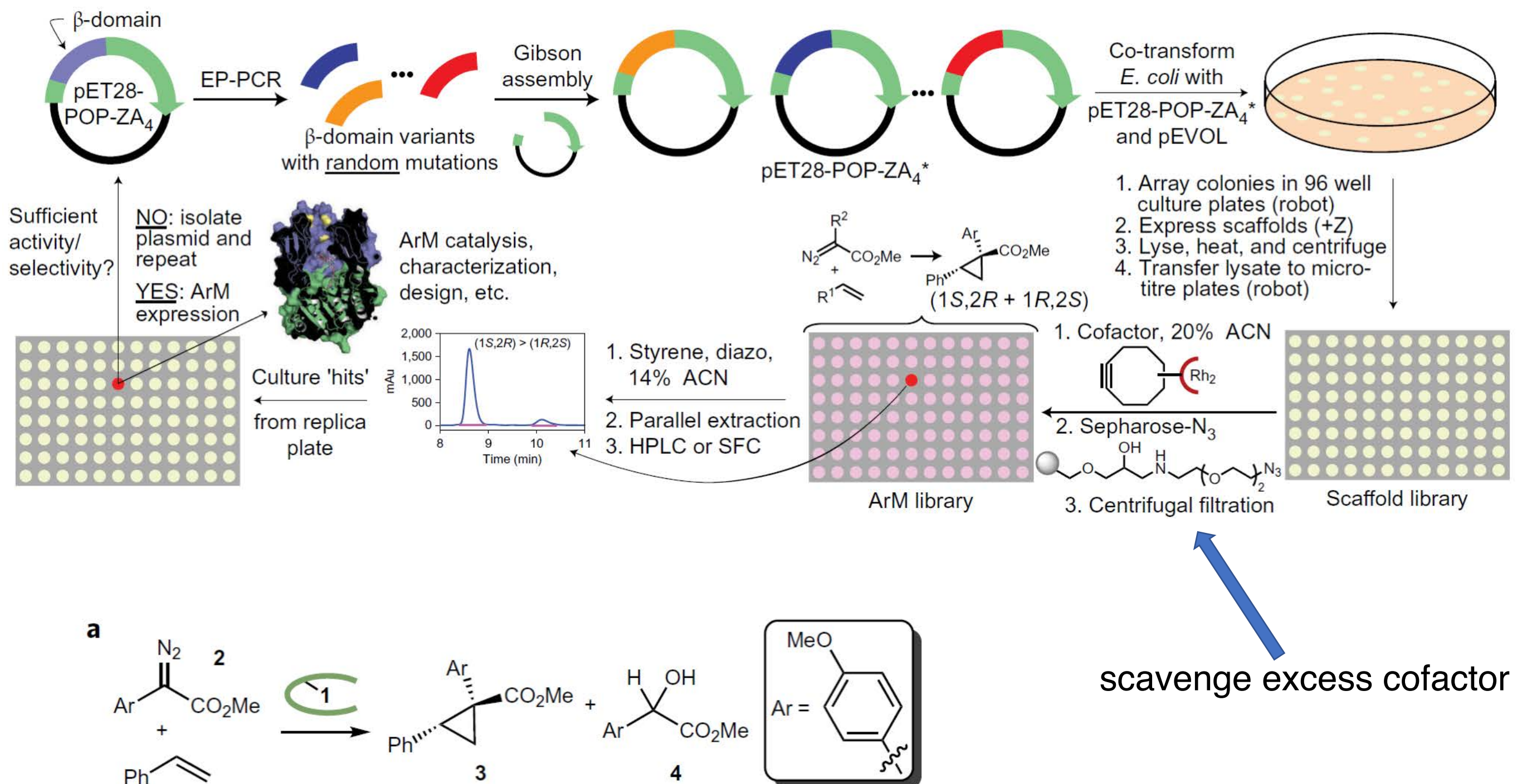
model



'biocompatible cofactor' – random mutagenesis

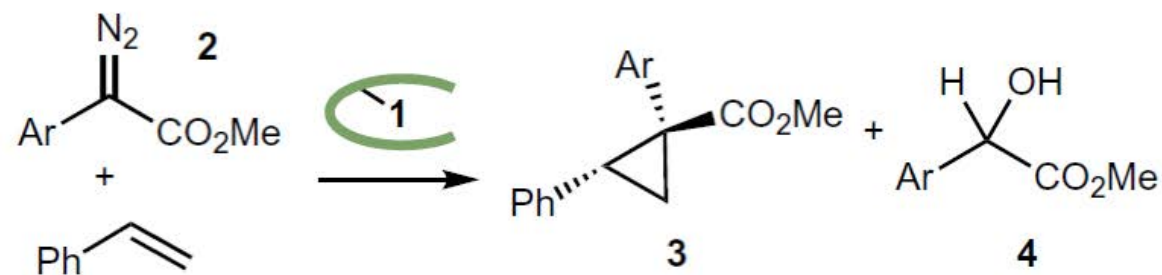
cfe heat treated

2. library generation and screening

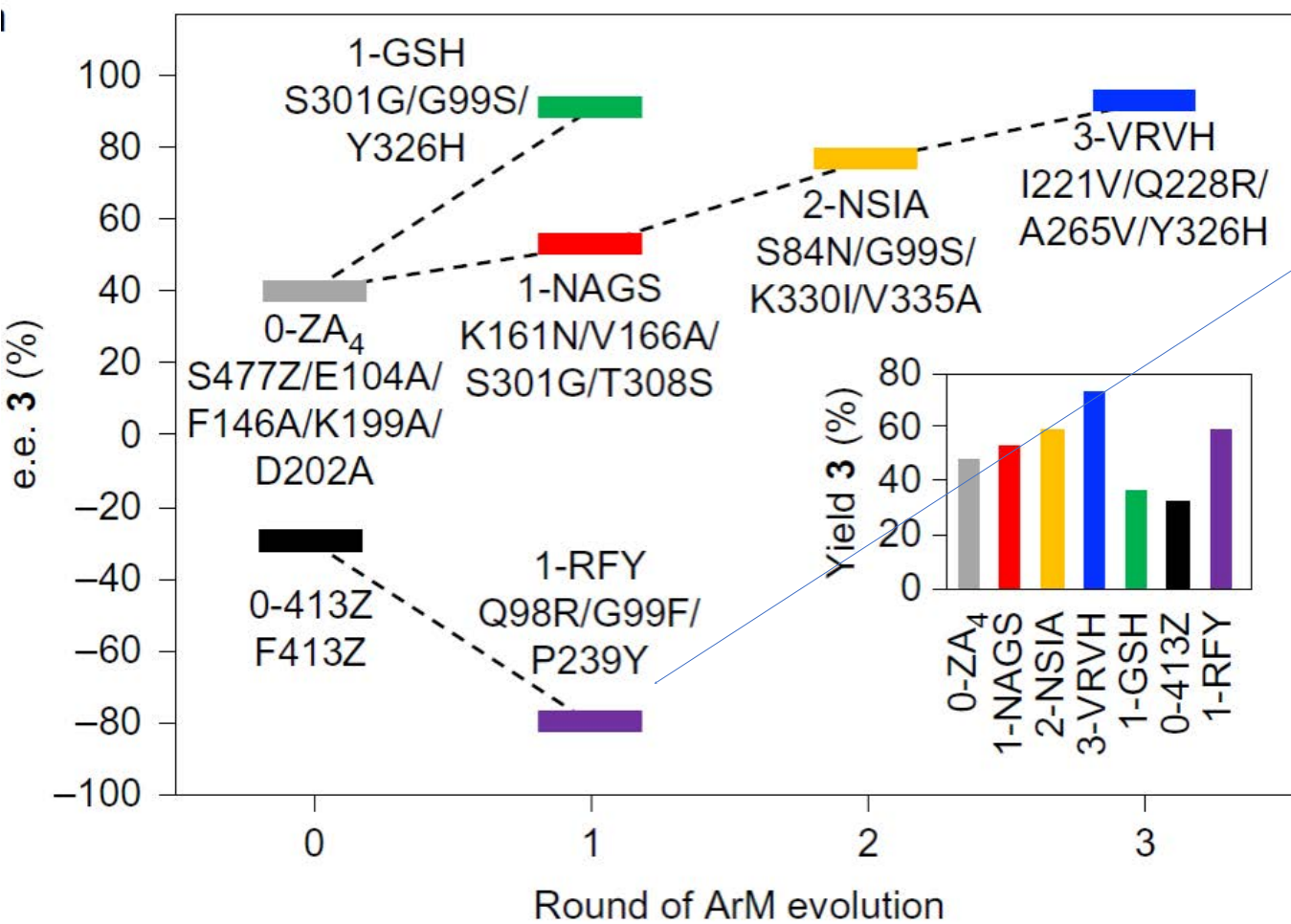


'biocompatible cofactor' – random mutagenesis

a



TON up to 66



random mutagenesis of selected sites based on crystal structure inspection with degenerate codons

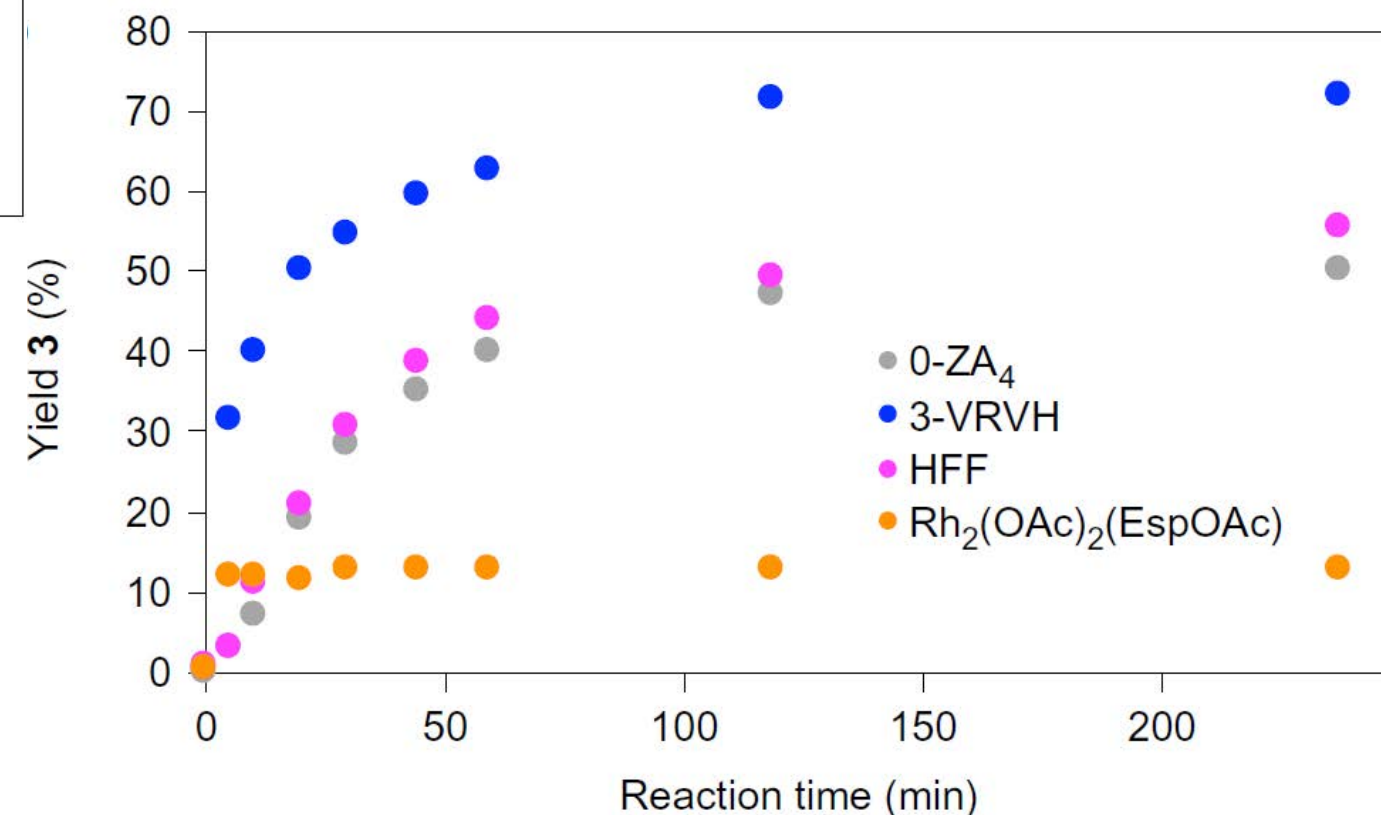
variants/
round

96

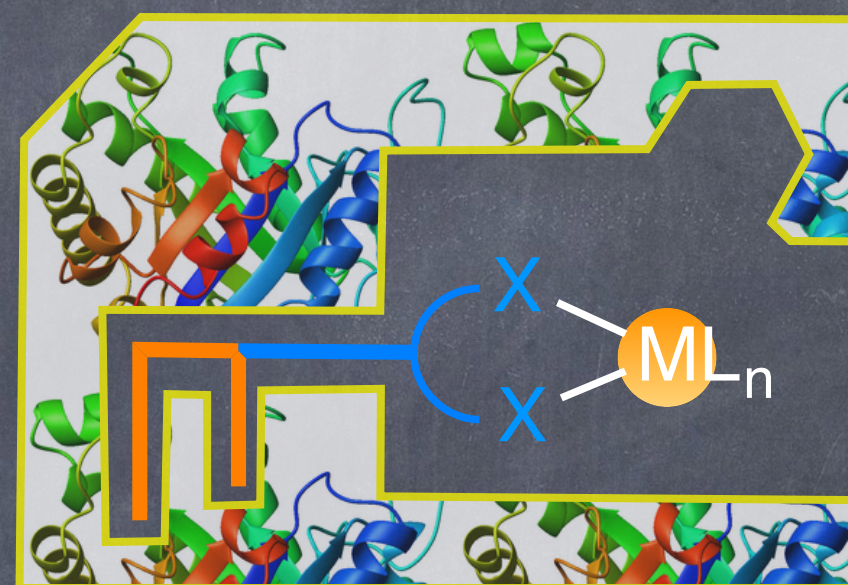
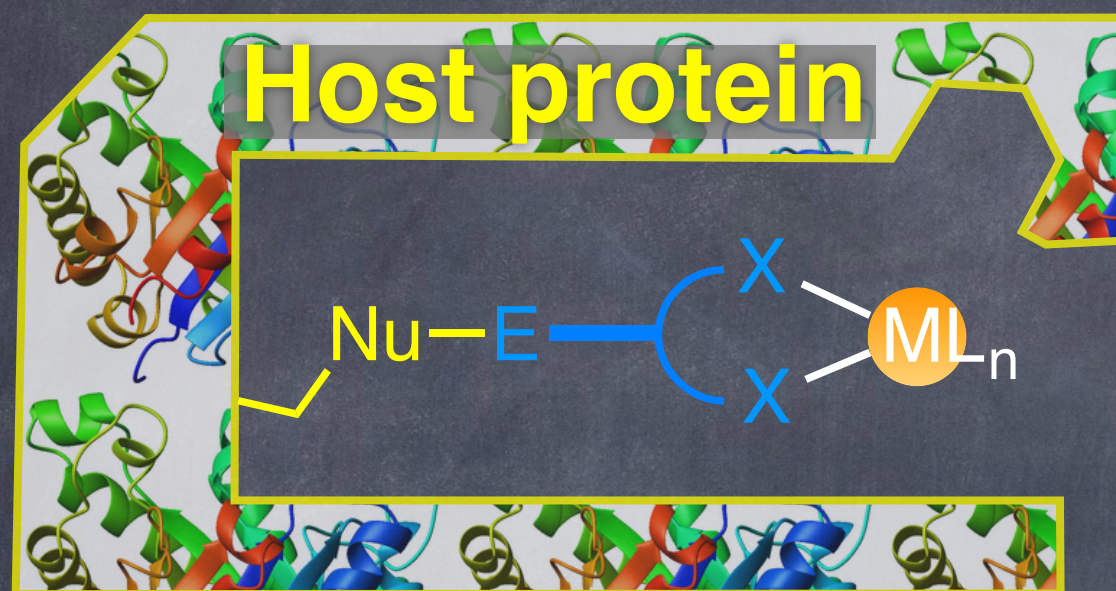
48

576

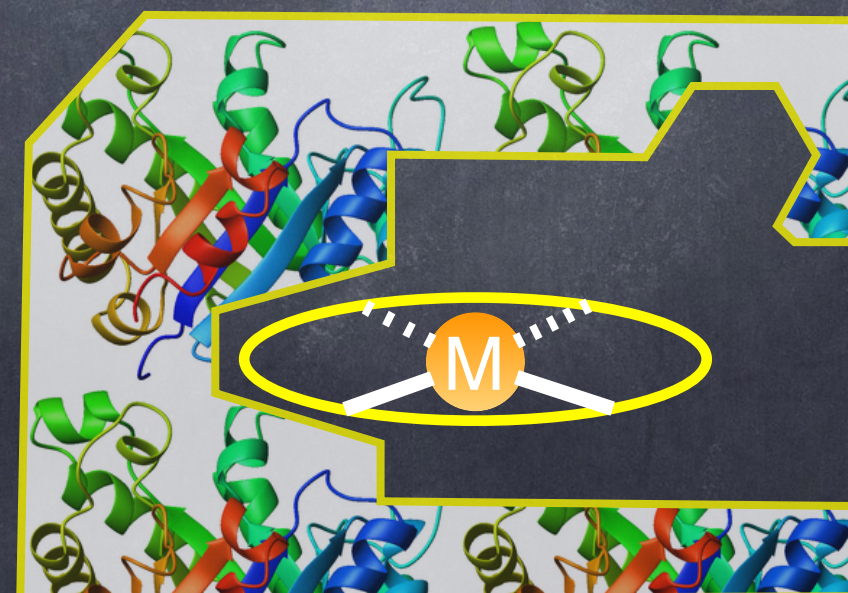
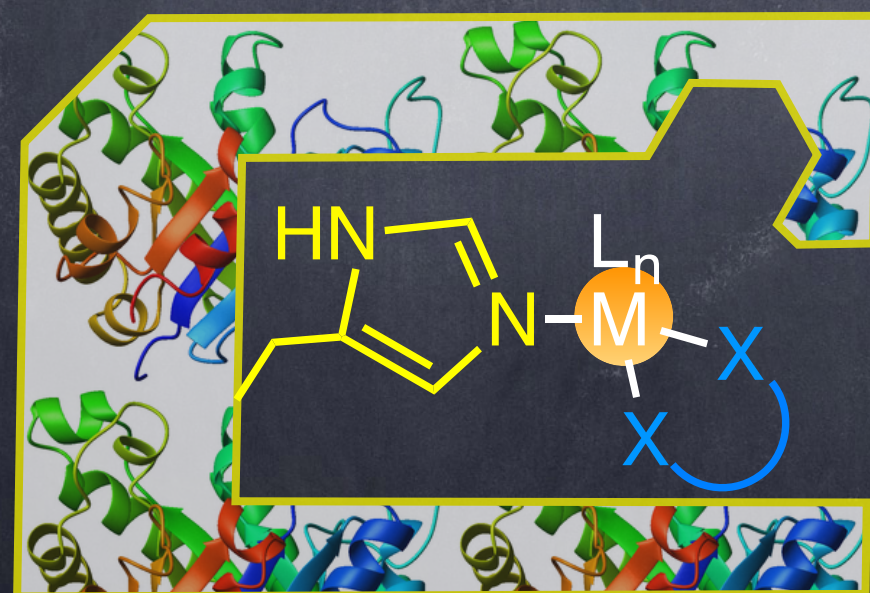
high mutation frequency



Anchoring of the Catalyst: Four Alternatives to Ensure Localization

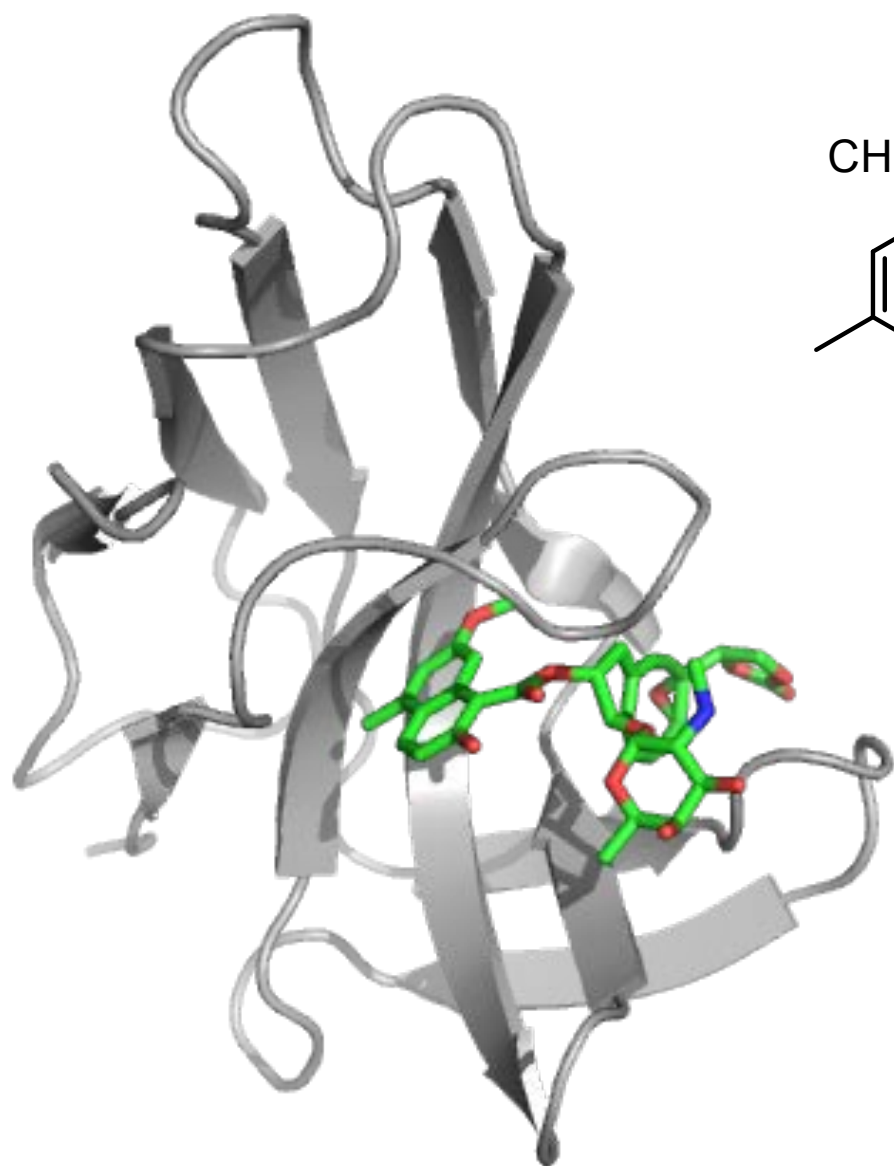


Supramolecular anchoring

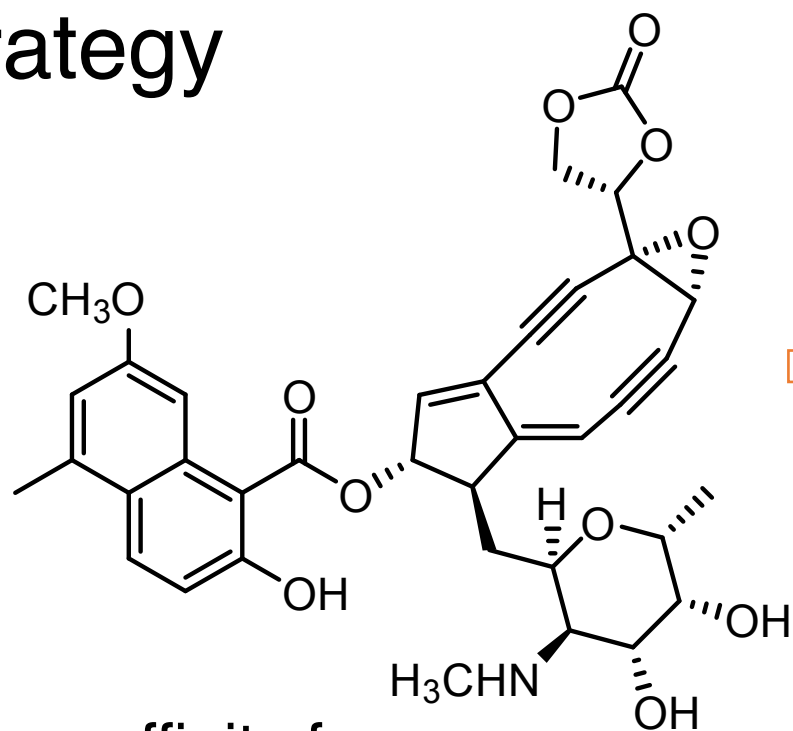


anchoring strategies - supramolecular

the Trojan horse strategy



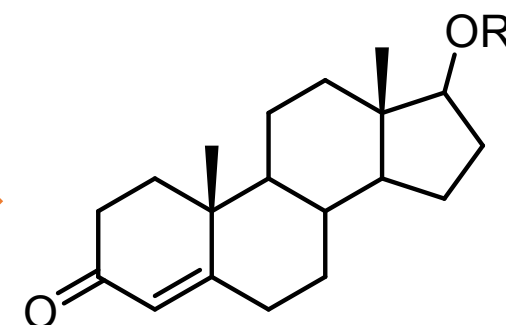
neocarzinostatin with natural ligand bound
PDB 1NCO



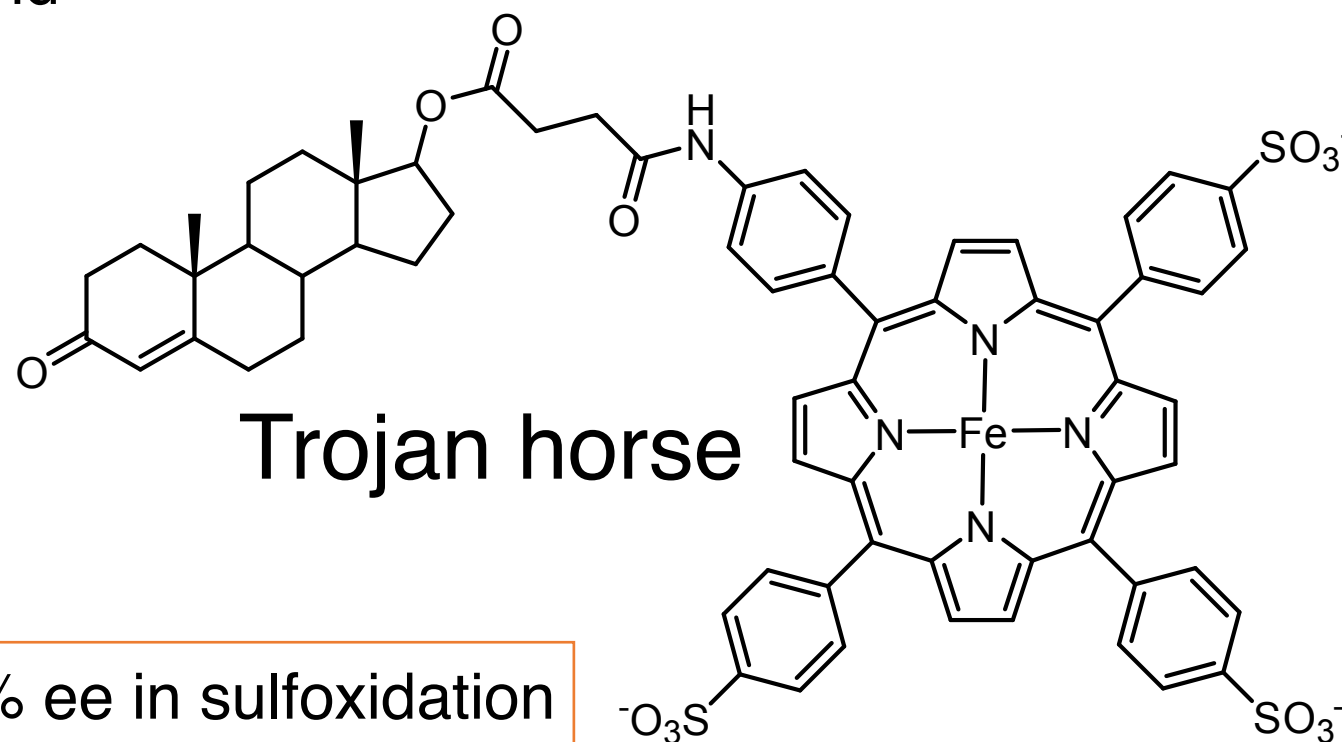
affinity for
natural ligand

analysis
directed evolution
(phage display)

7 mutations



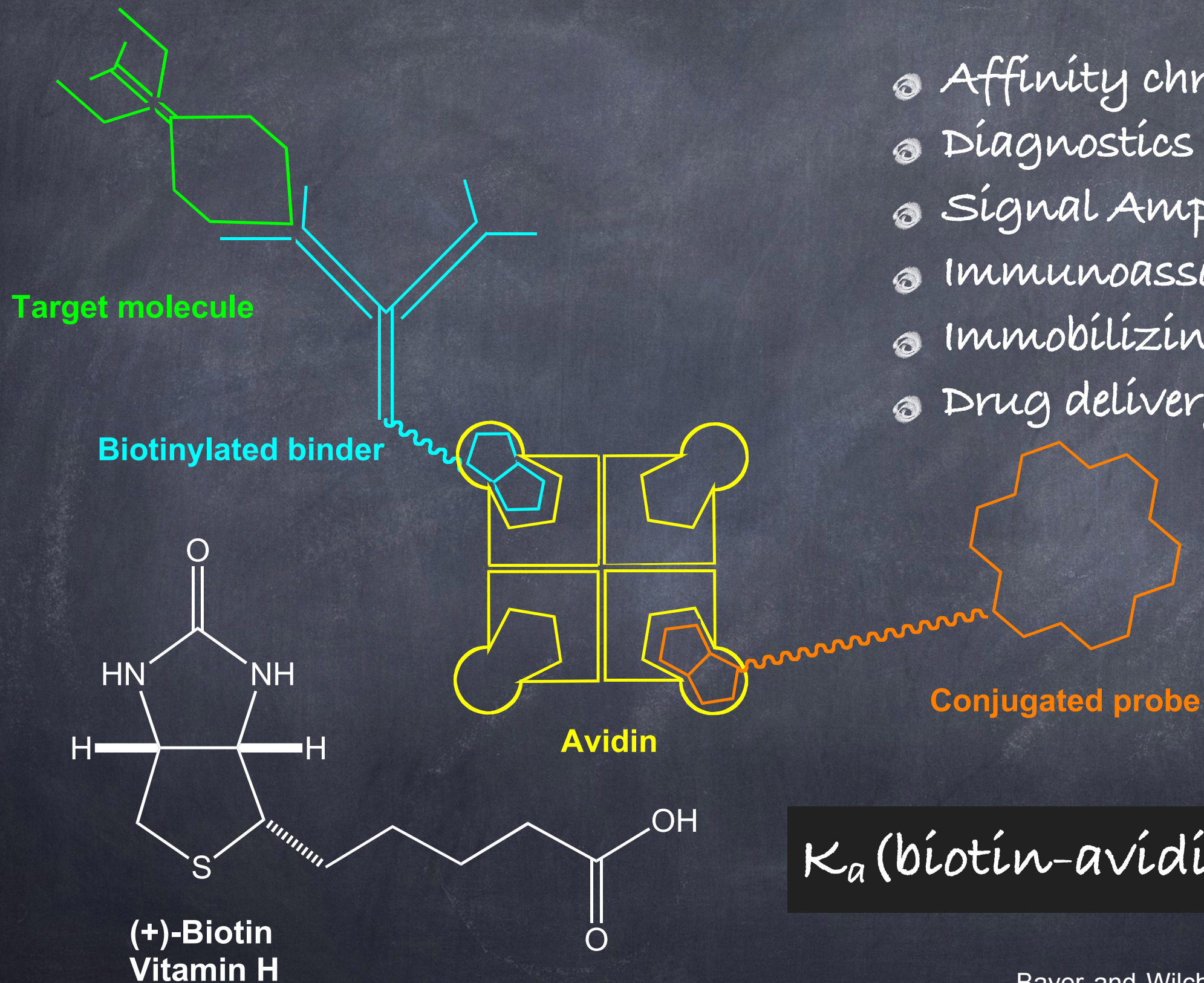
affinity for new ligand
 $K_d = 4 \times 10^{-5} \text{ M}$



Trojan horse

13% ee in sulfoxidation

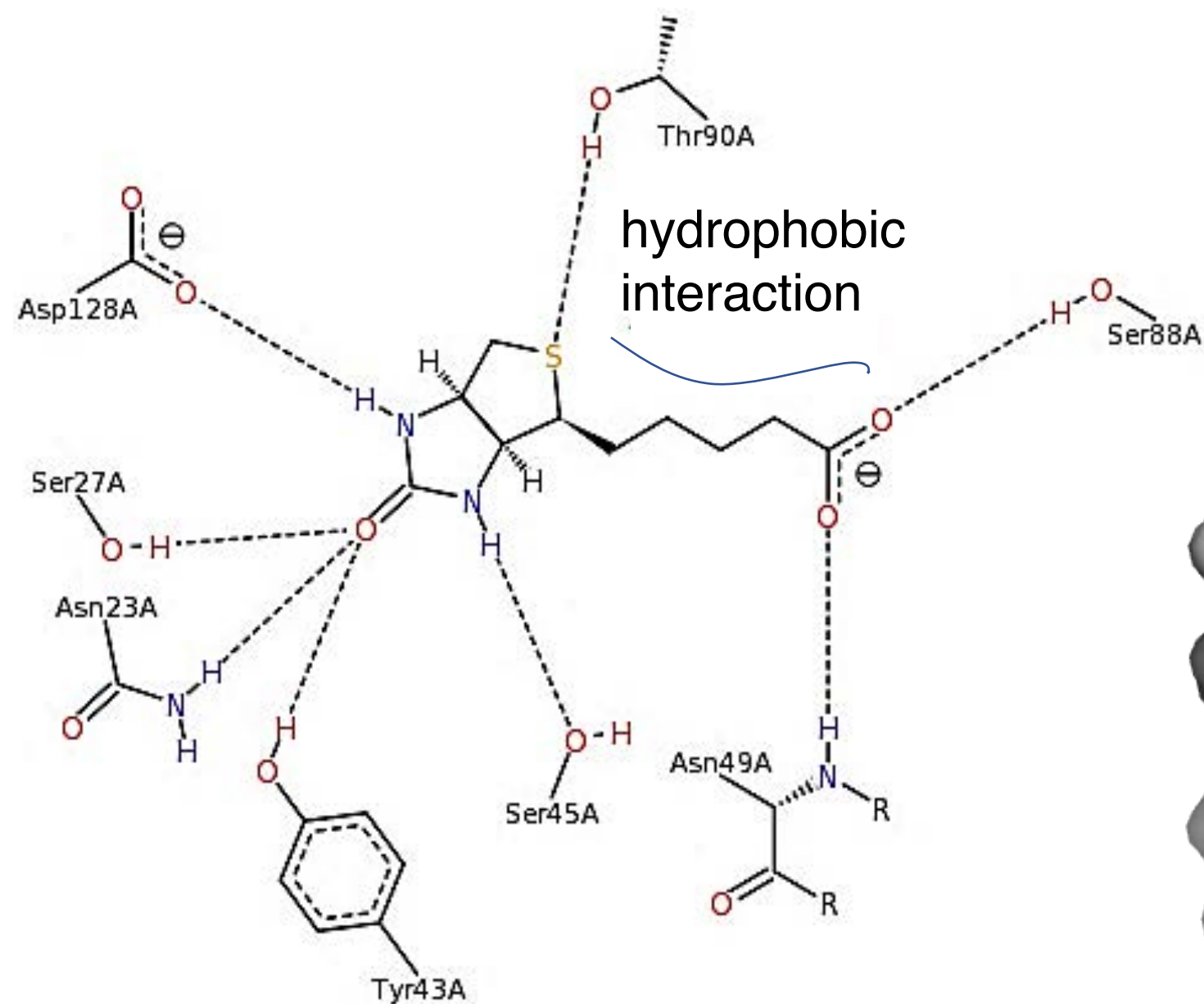
Biotin-Avidin Technology: Molecular Velcro



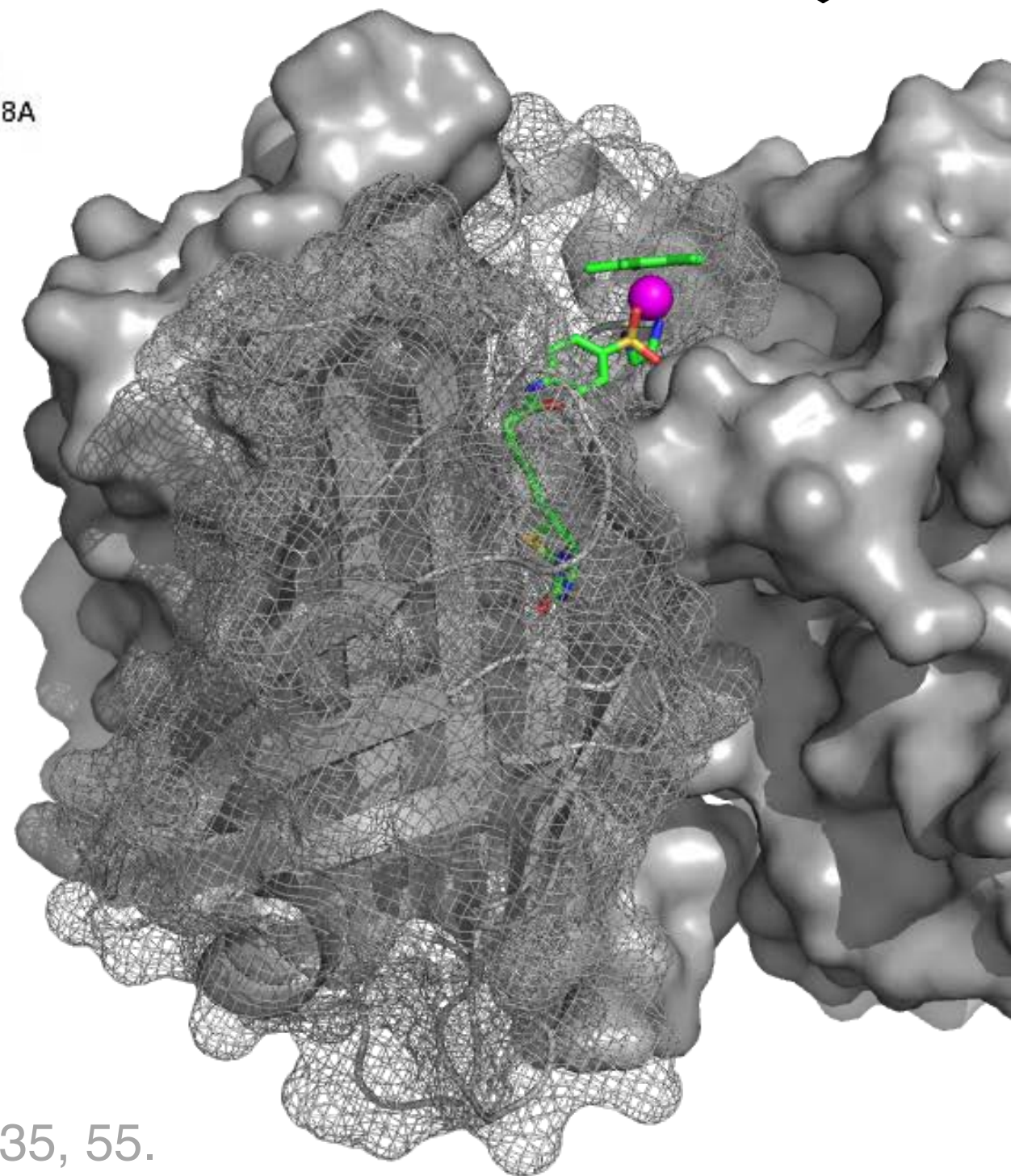
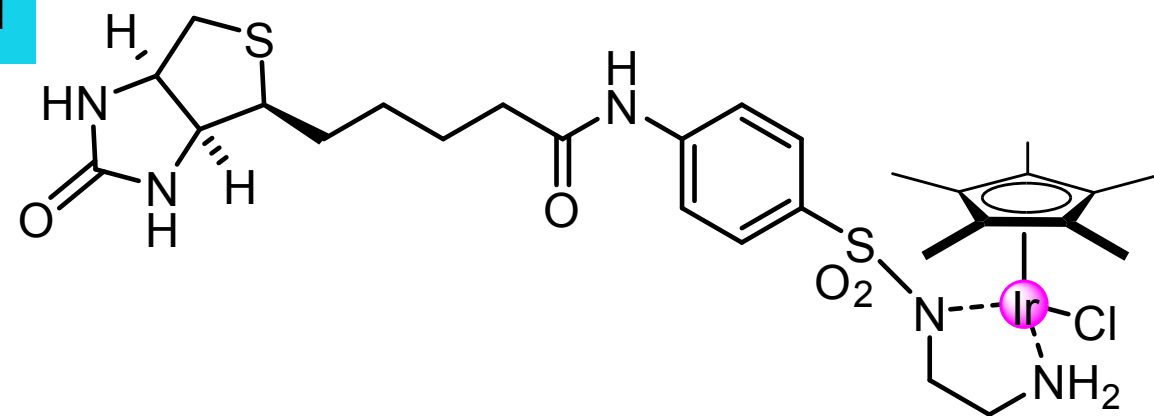
- Affinity chromatography
- Diagnostics
- Signal Amplification
- Immunoassay
- Immobilizing agent
- Drug delivery

$$K_a(\text{biotin-avidin}) 10^{14} \text{M}^{-1}$$

anchoring strategies - supramolecular



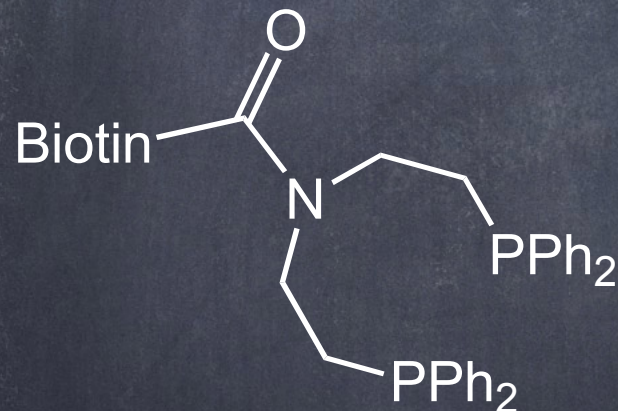
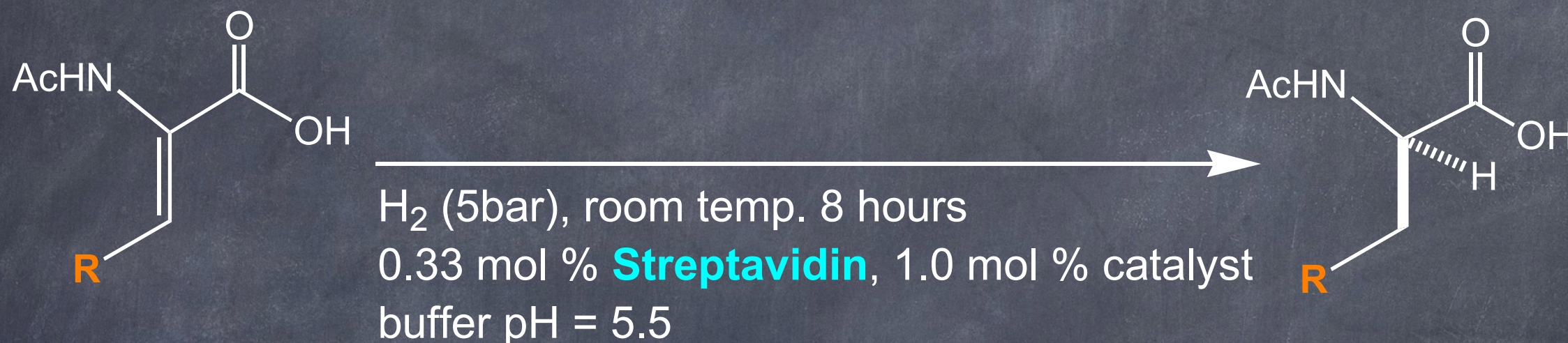
biotin-streptavidin interactions; $K_D \approx 10^{-14}$ M



affinity engineering: M. Howarth, Biochem J. 2011, 435, 55.

Ir^{cp}* catalysis: T. R. Ward, Angew. Chem. Int. Ed. 2011, 50, 3026.

Substrate Specificity: Enzyme-Like or Homogeneous Catalyst-Like?



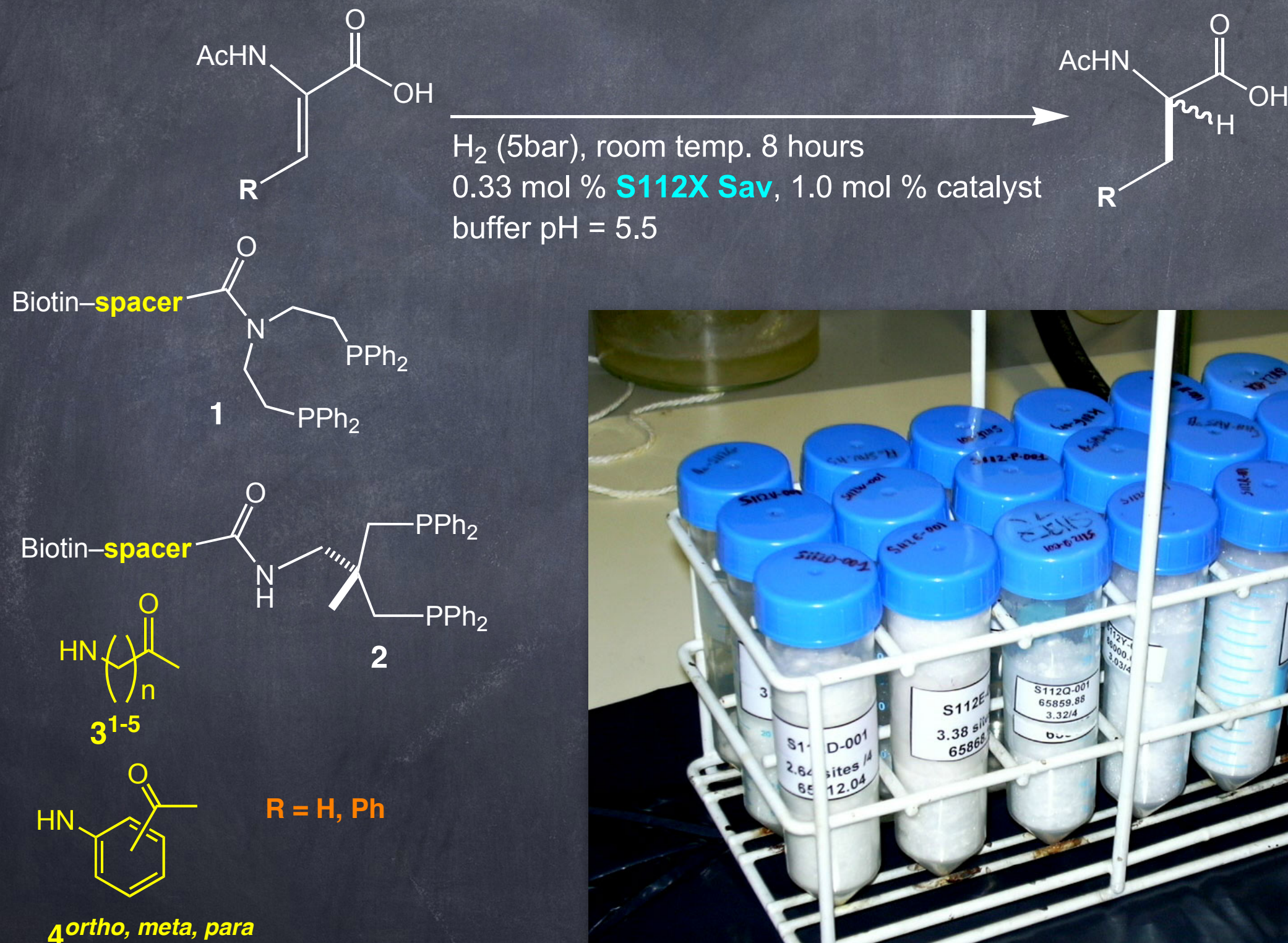
$\text{R} = \text{H}$

94 % ee (R)
quant.

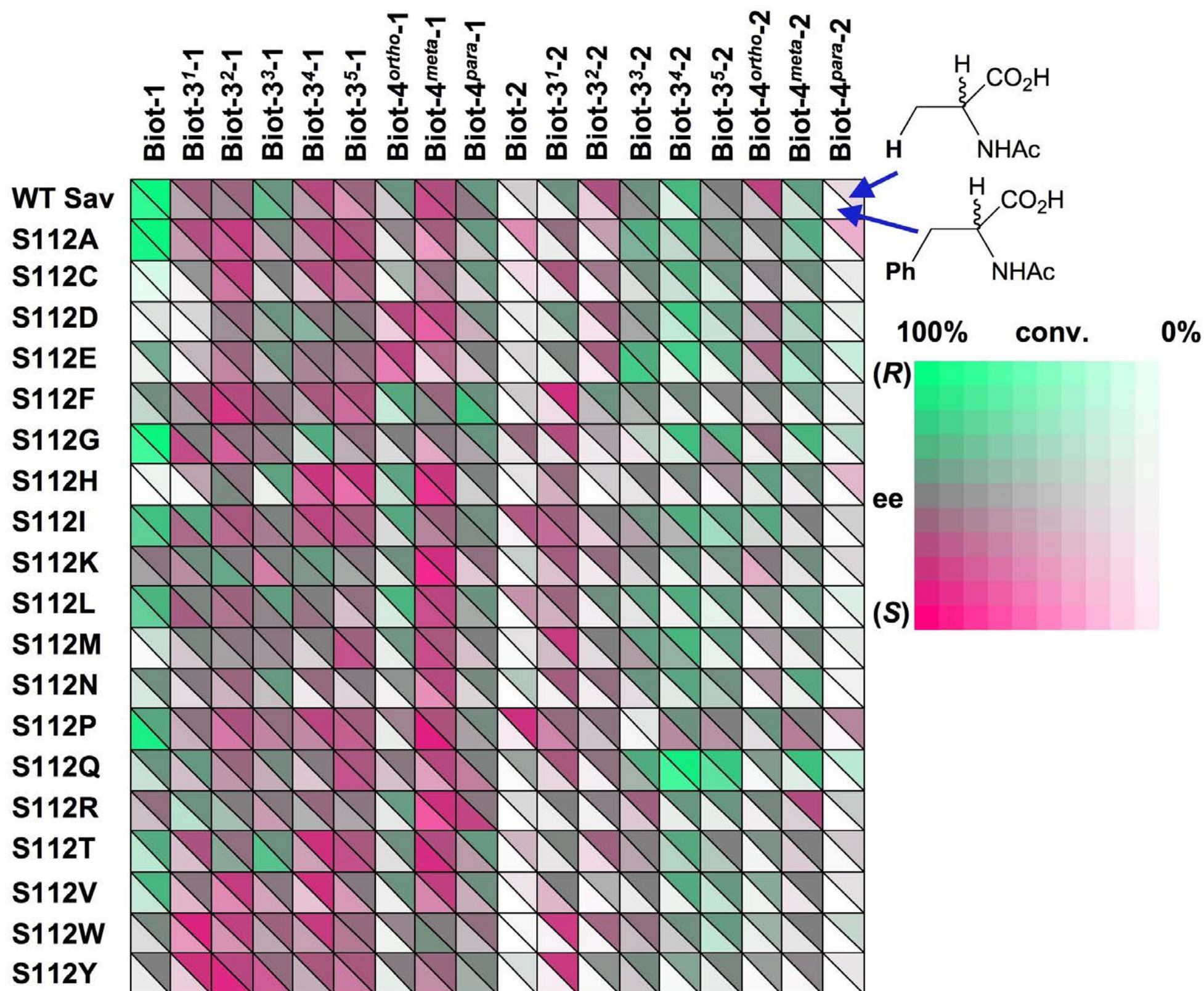
$\text{R} = \text{Ph}$

93 % ee (R)
quant.

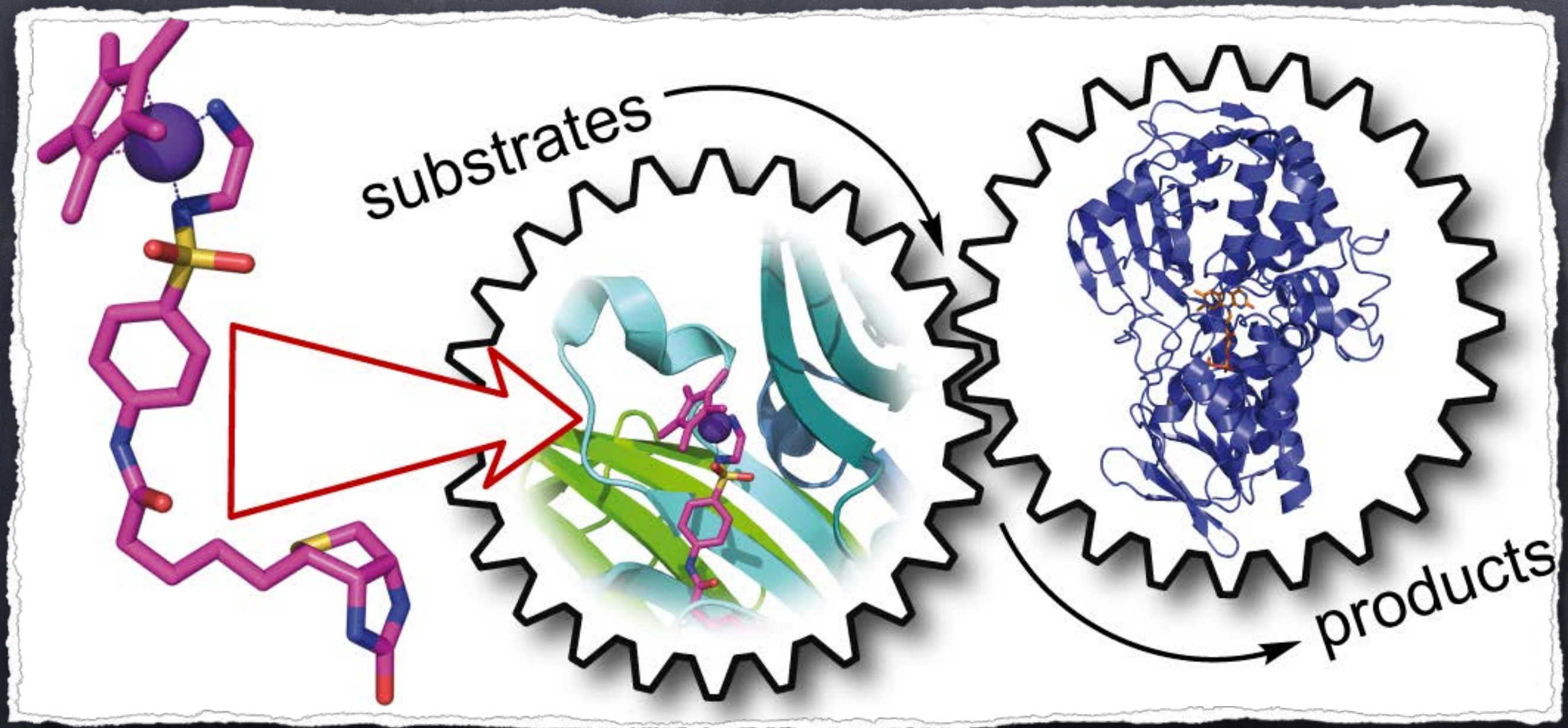
Chemo-Genetic Optimization: 18 Ligands Combined with 20 Proteins



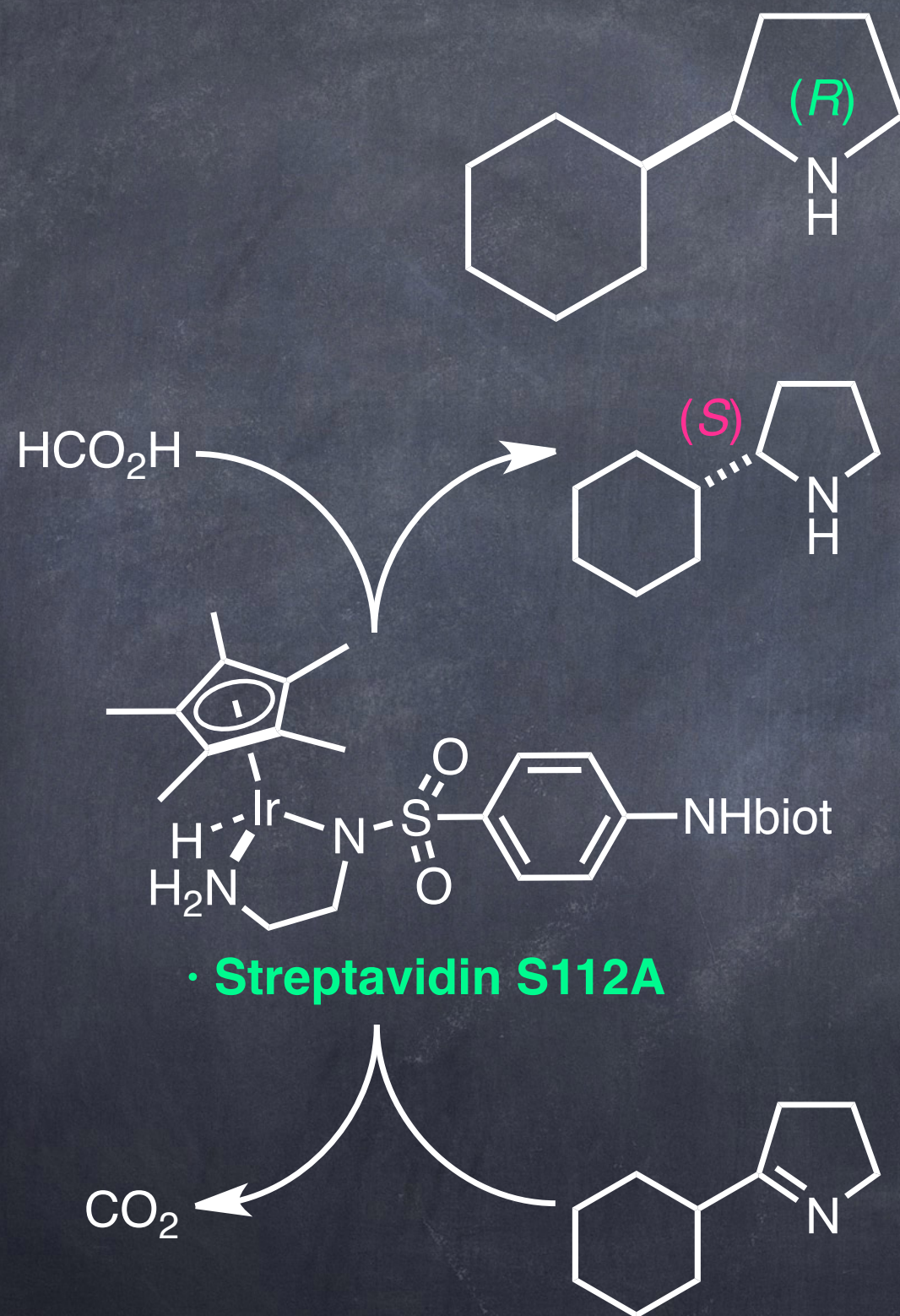
Chemo-Genetic Optimization: Fingerprints



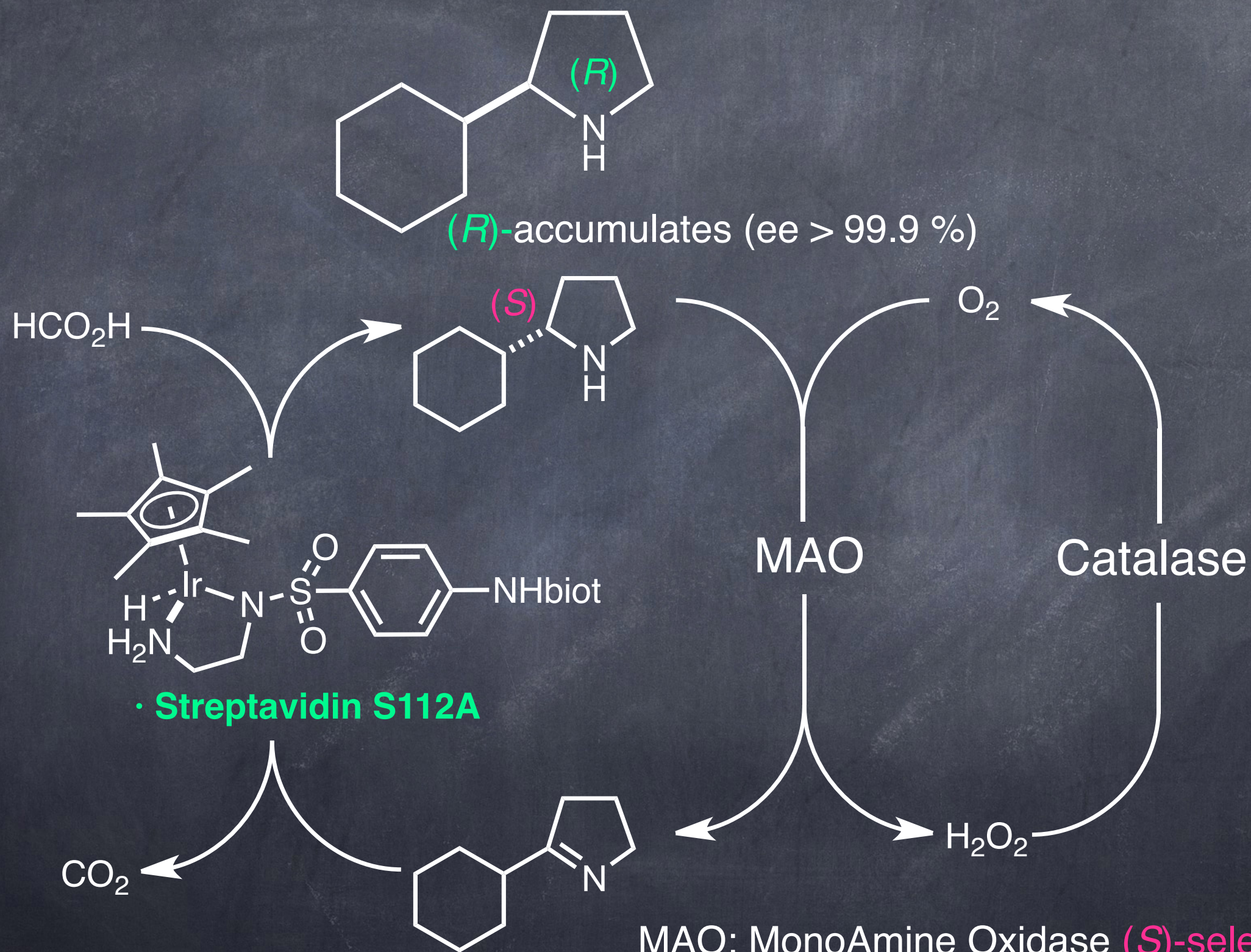
Enzyme Cascades...



Orthogonal Redox Cascades



Orthogonal Redox Cascades

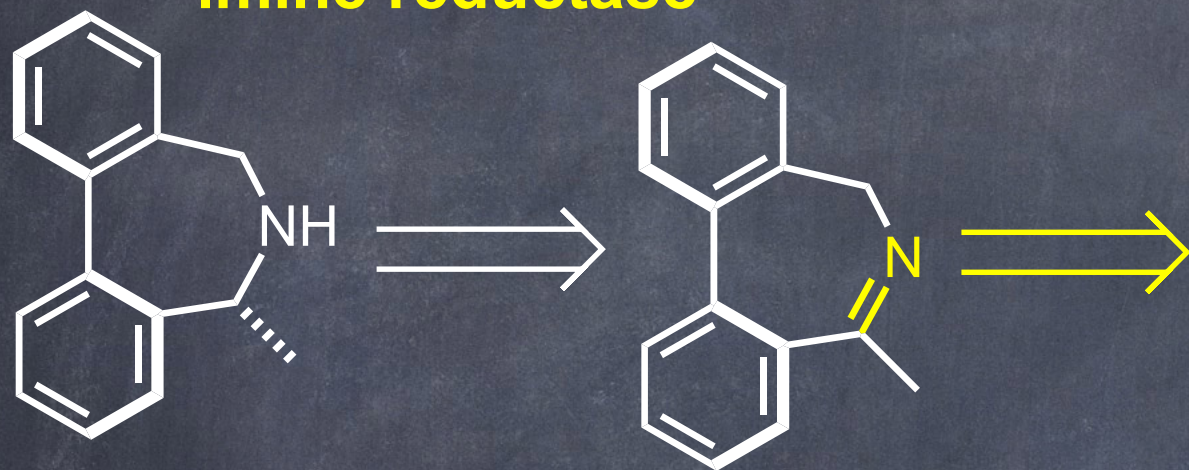


MAO: MonoAmine Oxidase (**S**)-selective

M. Dürrenberger, V. Köhler
with N. Turner (U. Manchester)

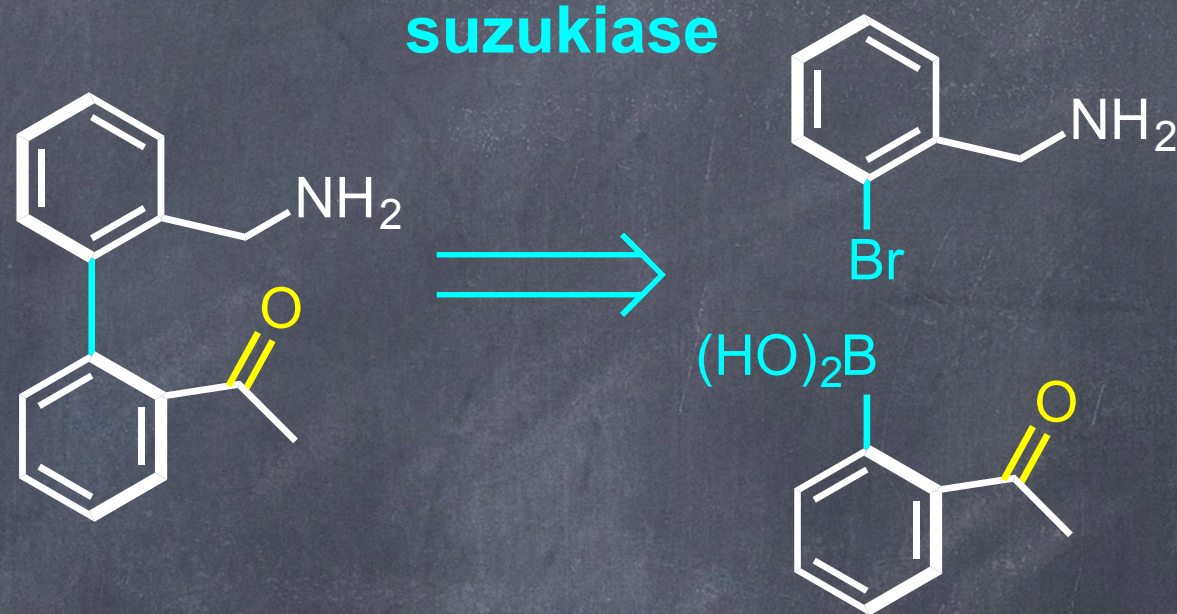
Can Artificial Metalloenzymes Complement Natural Enzymes?

imine reductase



Dibenzo-azepine

suzukiase



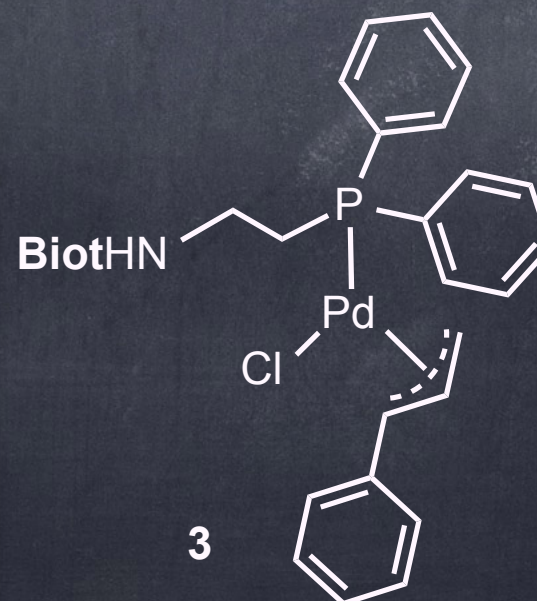
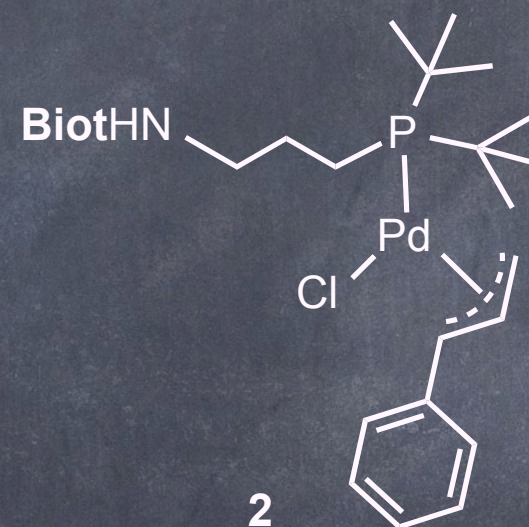
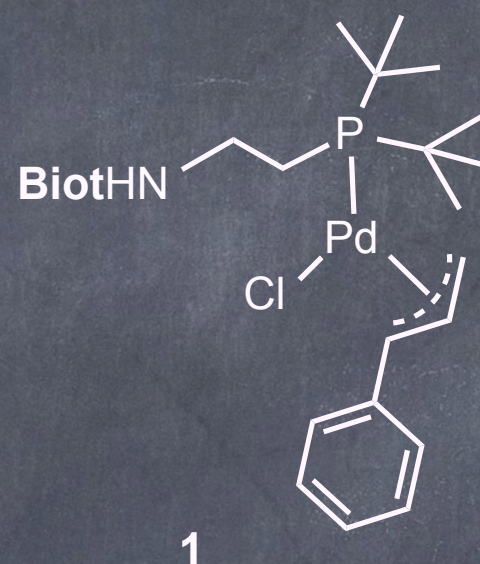
Artificial Suzukiase: Chemo-genetic Optimization

Complex 1 2 3

WT	●	●	●
S112G	●	●	●
S112A	●	●	●
S112V	●	●	●
S112L	●	●	●
S112C	●	●	●
S112M	●	●	●
S112T	●	●	●
S112F	●	●	●
S112Y	●	●	●
S112Q	●	●	●
S112R	●	●	●
S112H	●	●	●
K121A	●	●	●
K121C	●	●	●
K121M	●	●	●
K121D	●	●	●
K121E	●	●	●
K121F	●	●	●
K121Y	●	●	●
K121N	●	●	●
K121H	●	●	●
K121R	●	●	●

Complex 1

S112A-K121E	●
S112V-K121E	●
S112M-K121E	●
S112Y-K121E	●
S112W-K121E	●
S112E-K121E	●
S112N-K121E	●
S112Q-K121E	●
S112R-K121E	●
S112H-K121E	●
N118L-K121E	●
N118S-K121E	●
N118E-K121E	●
N118K-K121E	●



complex 1

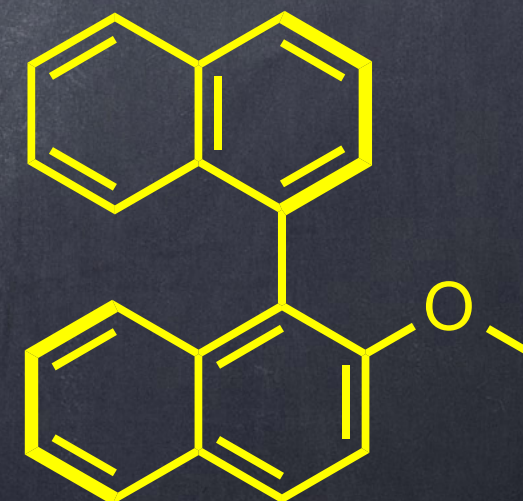
no Sav *rac.*, 10 TON

WT 58 (*R*), 78
 K121M 67, 59
 K121E 76, 50
 S112A 60, 58
 S112Y 16 (*S*), 14
 S112A-K121E 70, 80
 N118L-K121E 72, 86
 S112Y-K121E 90, 50

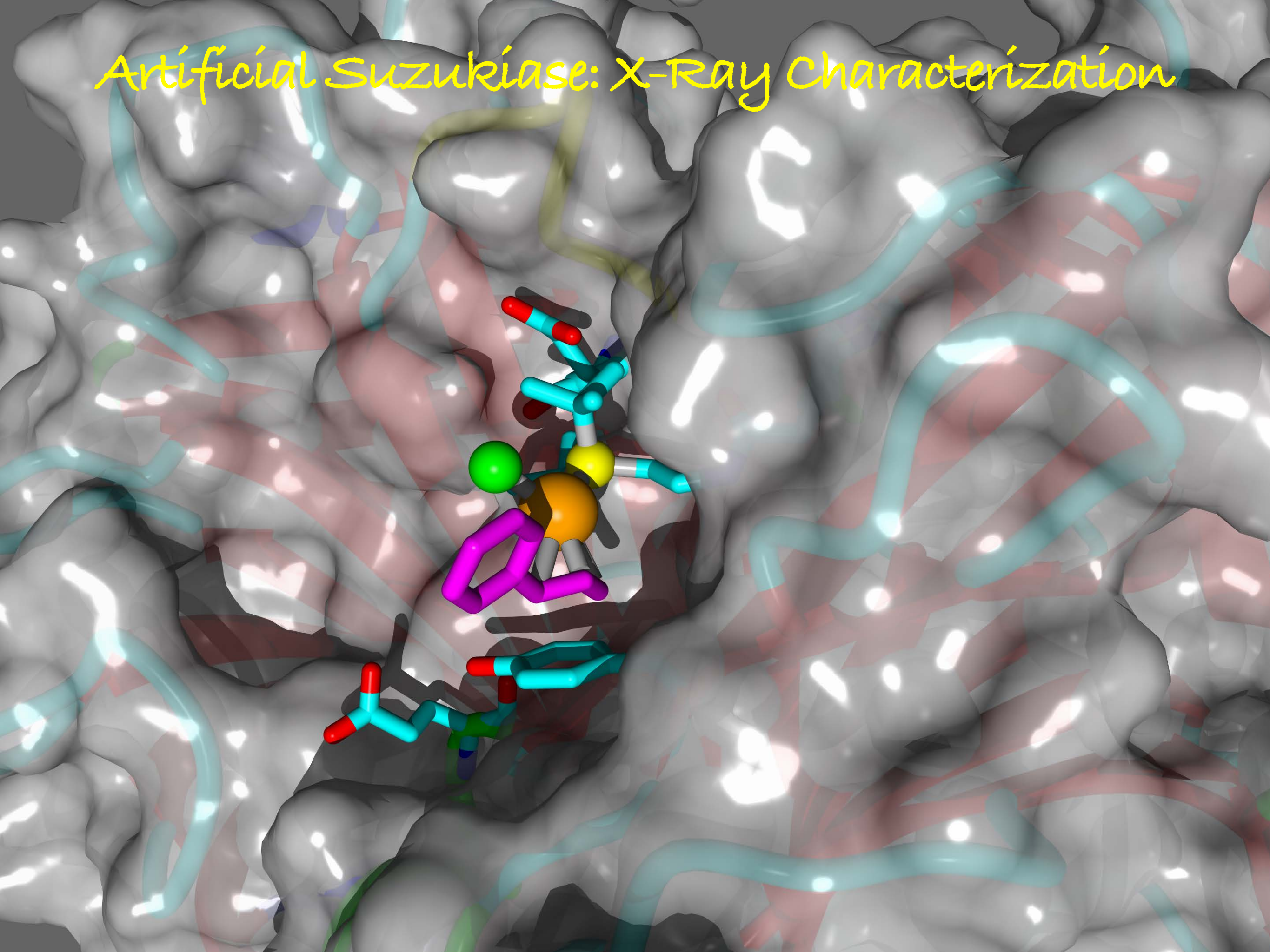
complex 2

WT 10 (*S*), 73
 S112L 51, 45
 S112M 44, 53
 K121A 47, 32

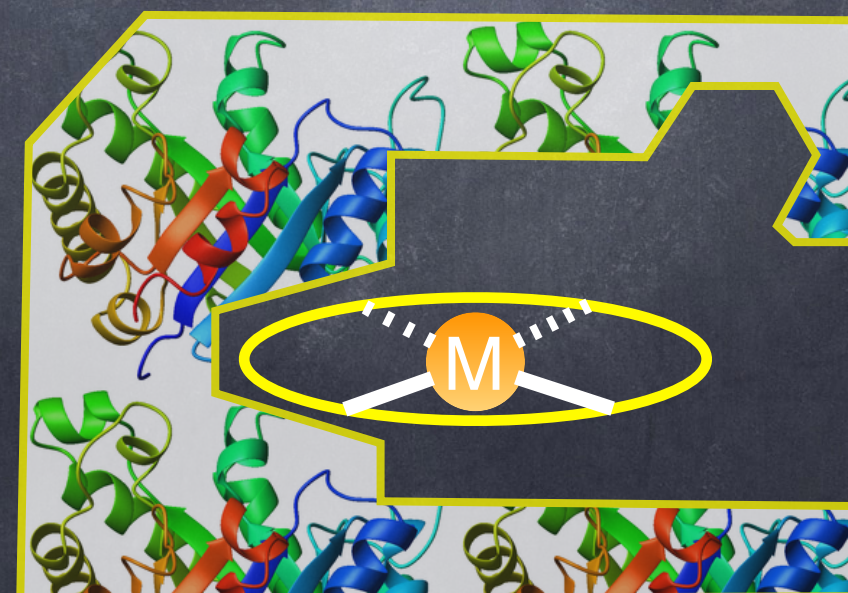
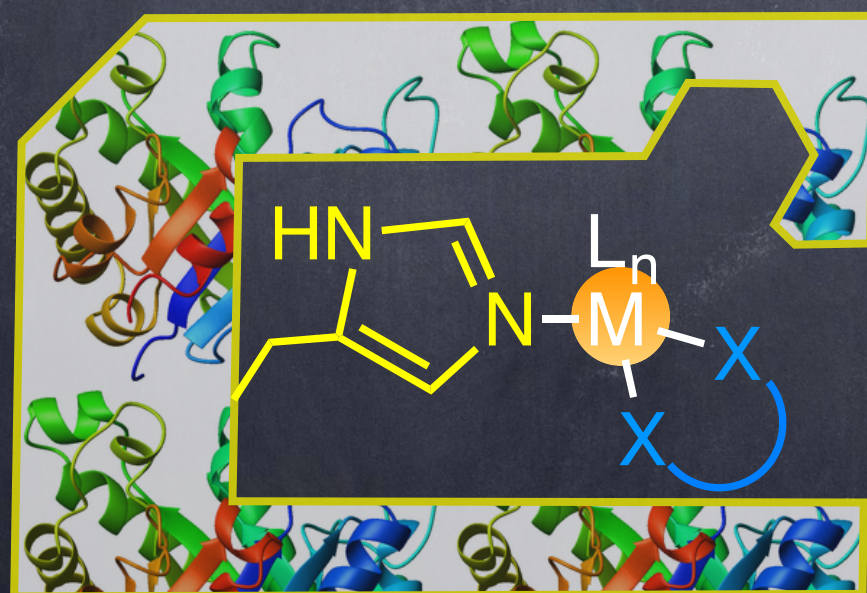
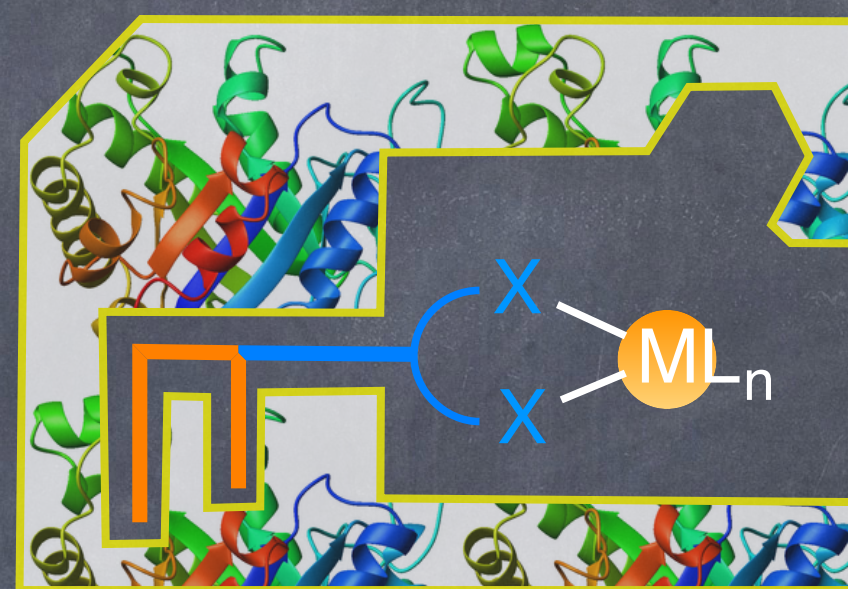
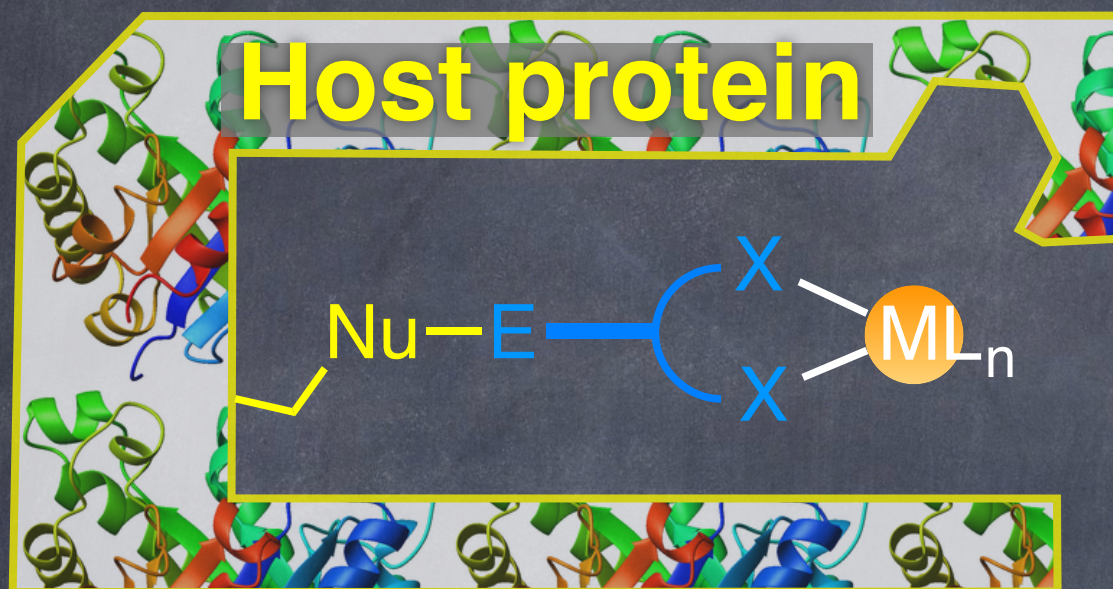
90	●	80	●	●	47
75	●	64	●	●	38
60	●	48	●	●	28
40	●	32	●	●	19
20	●	16	●	●	9
0	●	0	●	●	0
conv. [%]		(<i>R</i>)	ee [%]	(<i>S</i>)	



Artificial Suzukiase: X-Ray Characterization



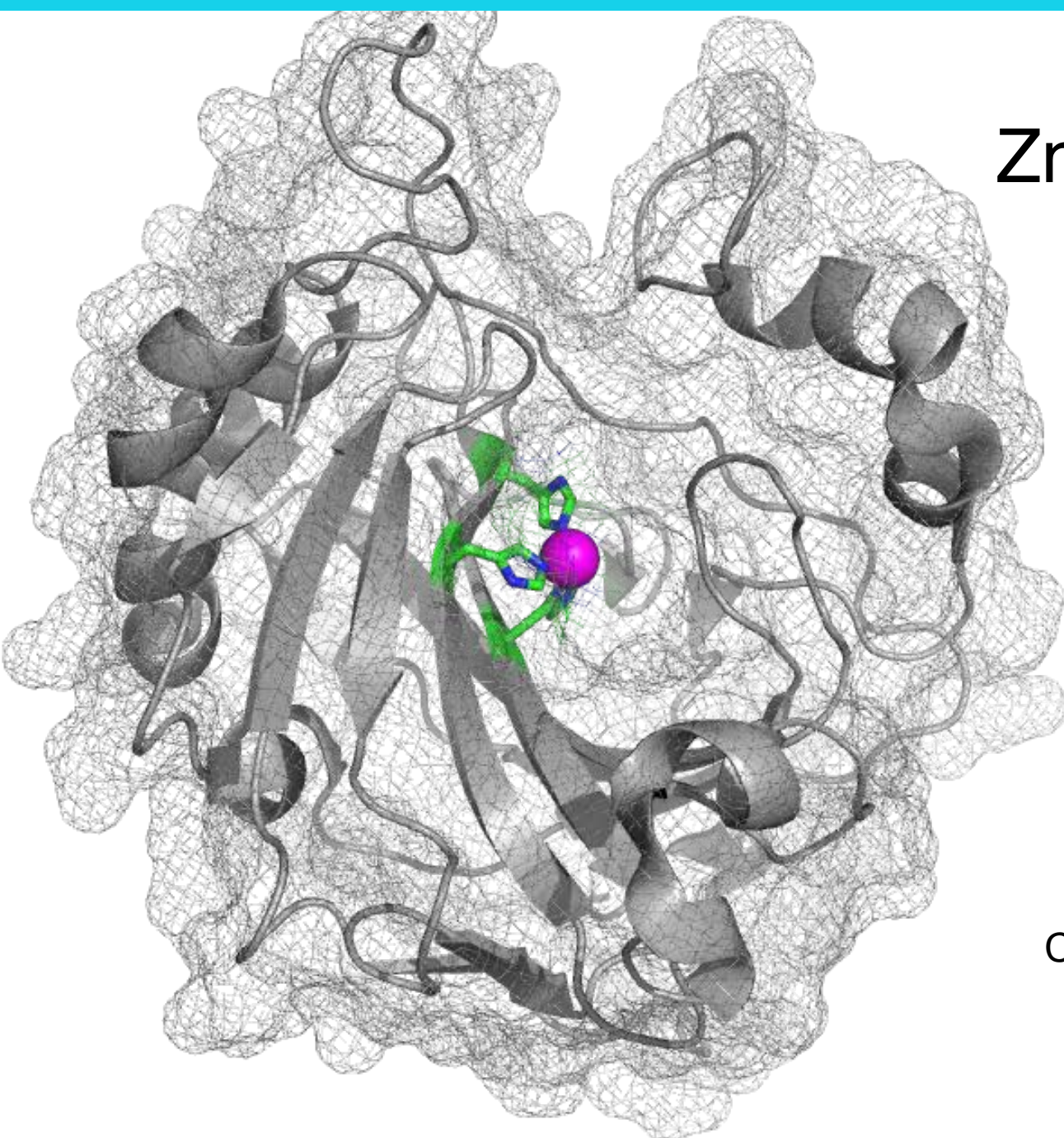
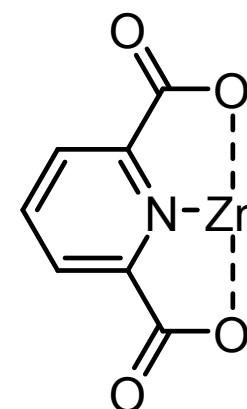
Anchoring of the Catalyst: Four Alternatives to Ensure Localization



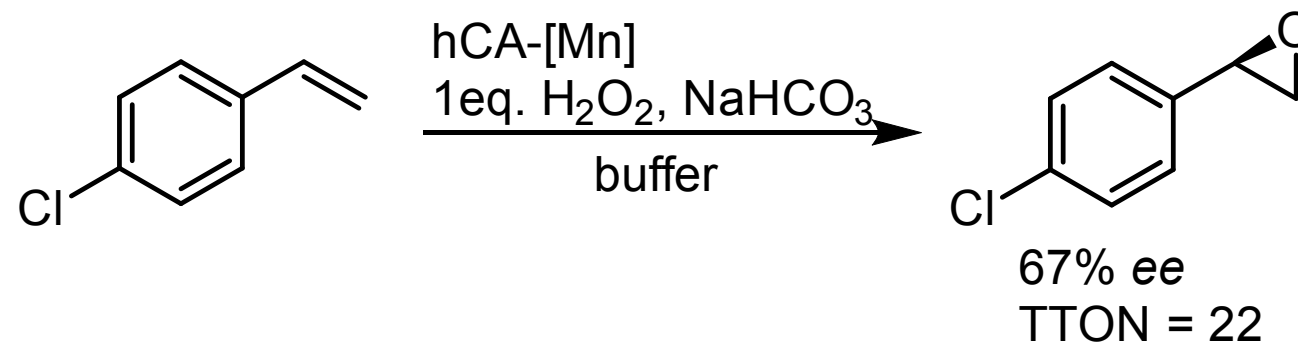
Dative anchoring

anchoring strategies – dative – metal coordination by aa-side chains

1. dialysis against chelating ligand
2. incubation with Mn^{2+}



human carbonic anhydrase
PDB 1G1D



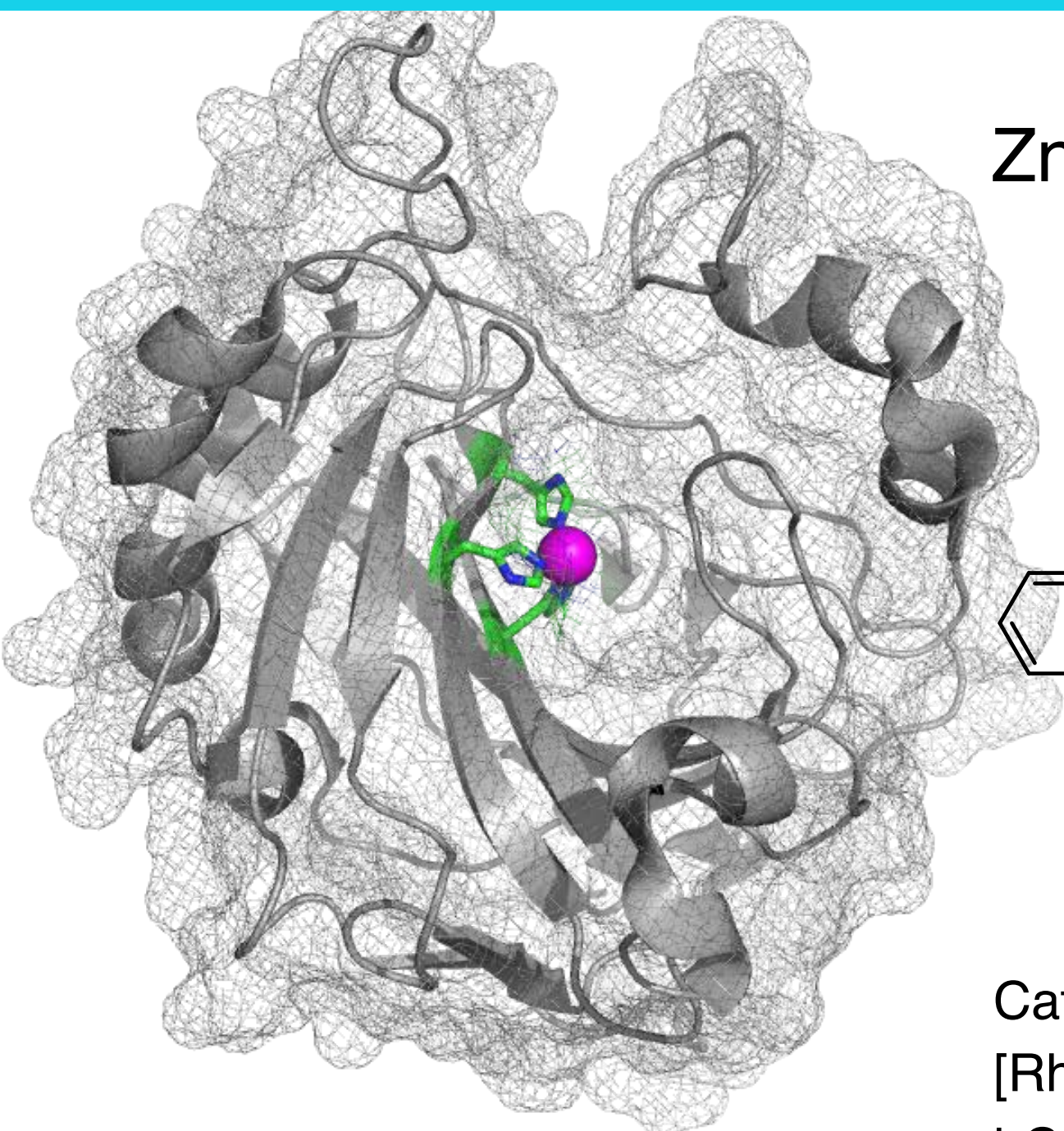
anchoring strategies – dative – metal coordination by aa-side chains

1. dialysis against chelating ligand
2. incubation with Rh⁺

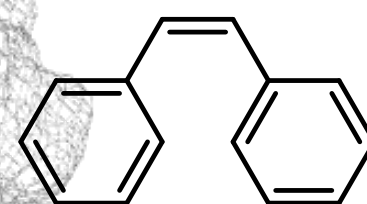
Zn²⁺



Rh⁺

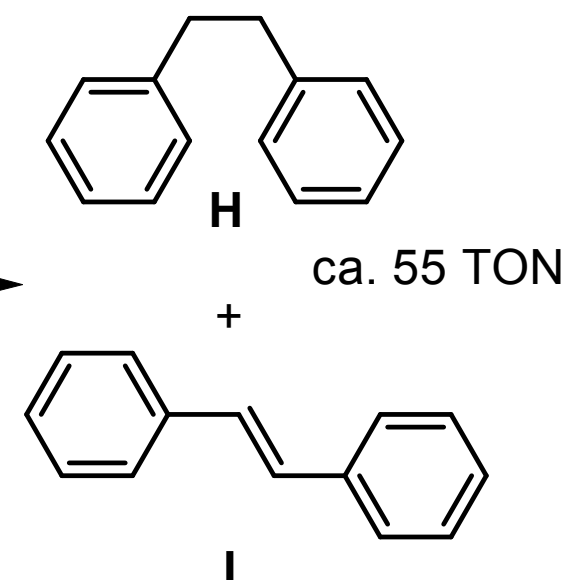


human carbonic anhydrase
PDB 1G1D



hCAII-[Rh] (1 mol%)

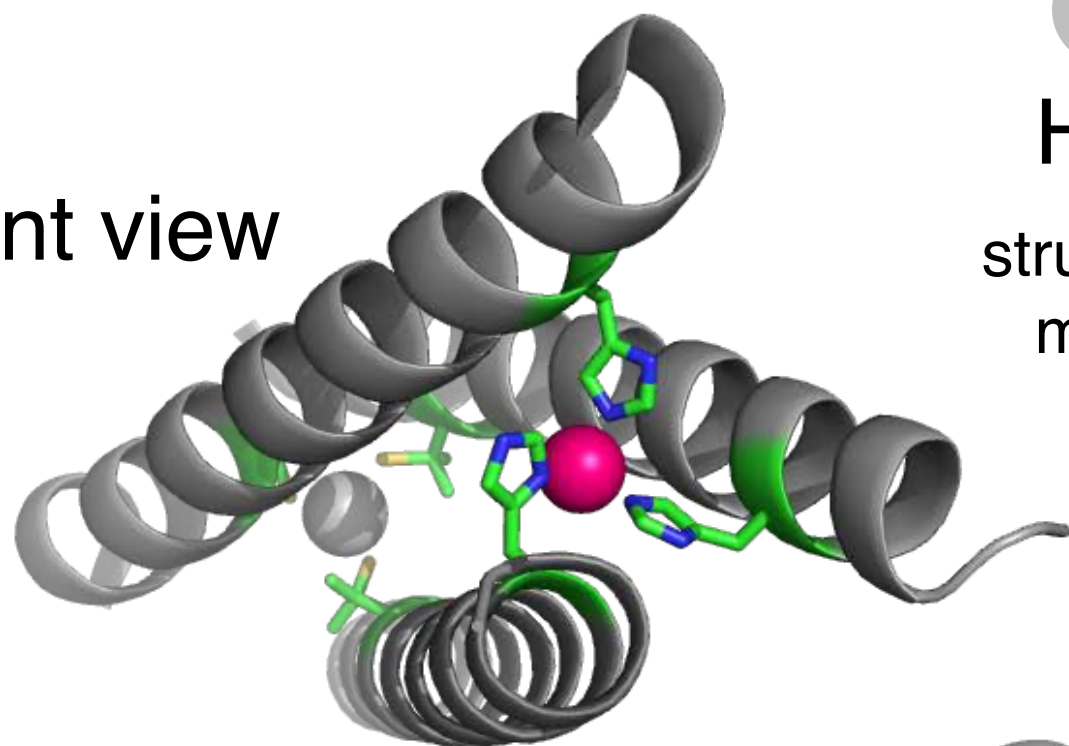
H₂ (5 atm)
MES buffer (0.1 M)
RT, 12 h



Catalyst	Rh / hCAII	H / I
[Rh(cod) ₂] ₂ BF ₄	-	6.3
hCAII-[Rh]	6.5	3.5
9*His-hCAII-[Rh]	1.8	20.6

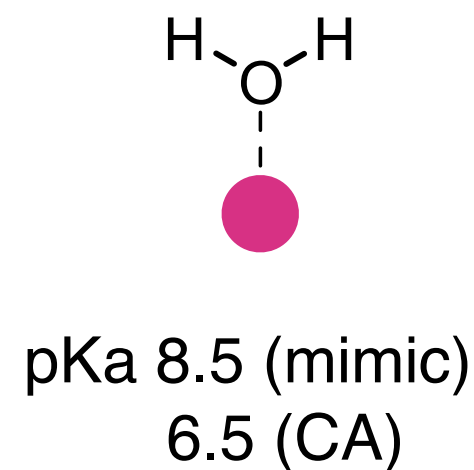
anchoring strategies – dative – in de novo design

front view



Hg^{2+}
structural
metal

Zn^{2+}
catalytic
center



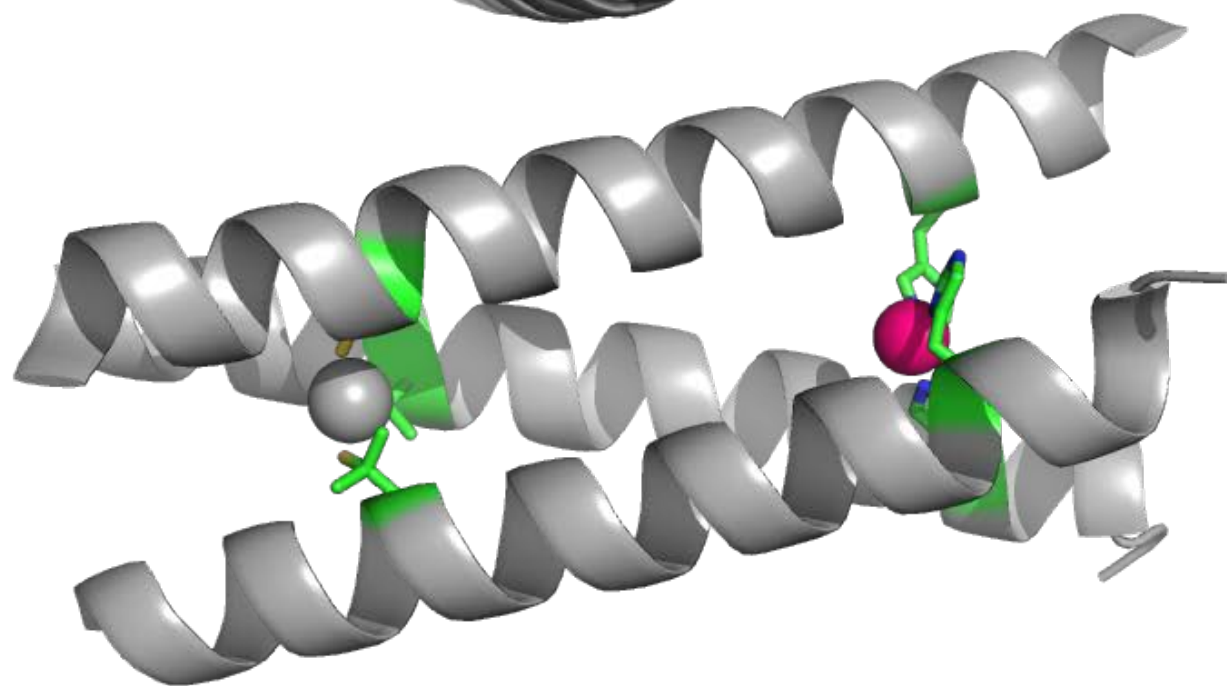
a carbonic anhydrase (CA)



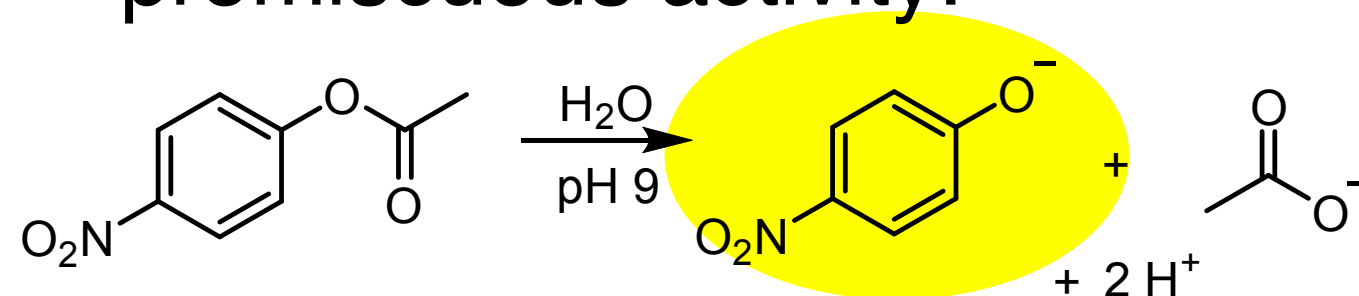
	CA (phys. pH)	mimic (pH 9.5)
$k_{\text{cat}} (\text{s}^{-1})$	8.2×10^5	$1.8 (\pm 0.4) \times 10^3$
$K_{\text{M}} (\text{mM})$	89	10.0 ± 2.4

side view

crystal structure of analogue
PDB 3PBJ



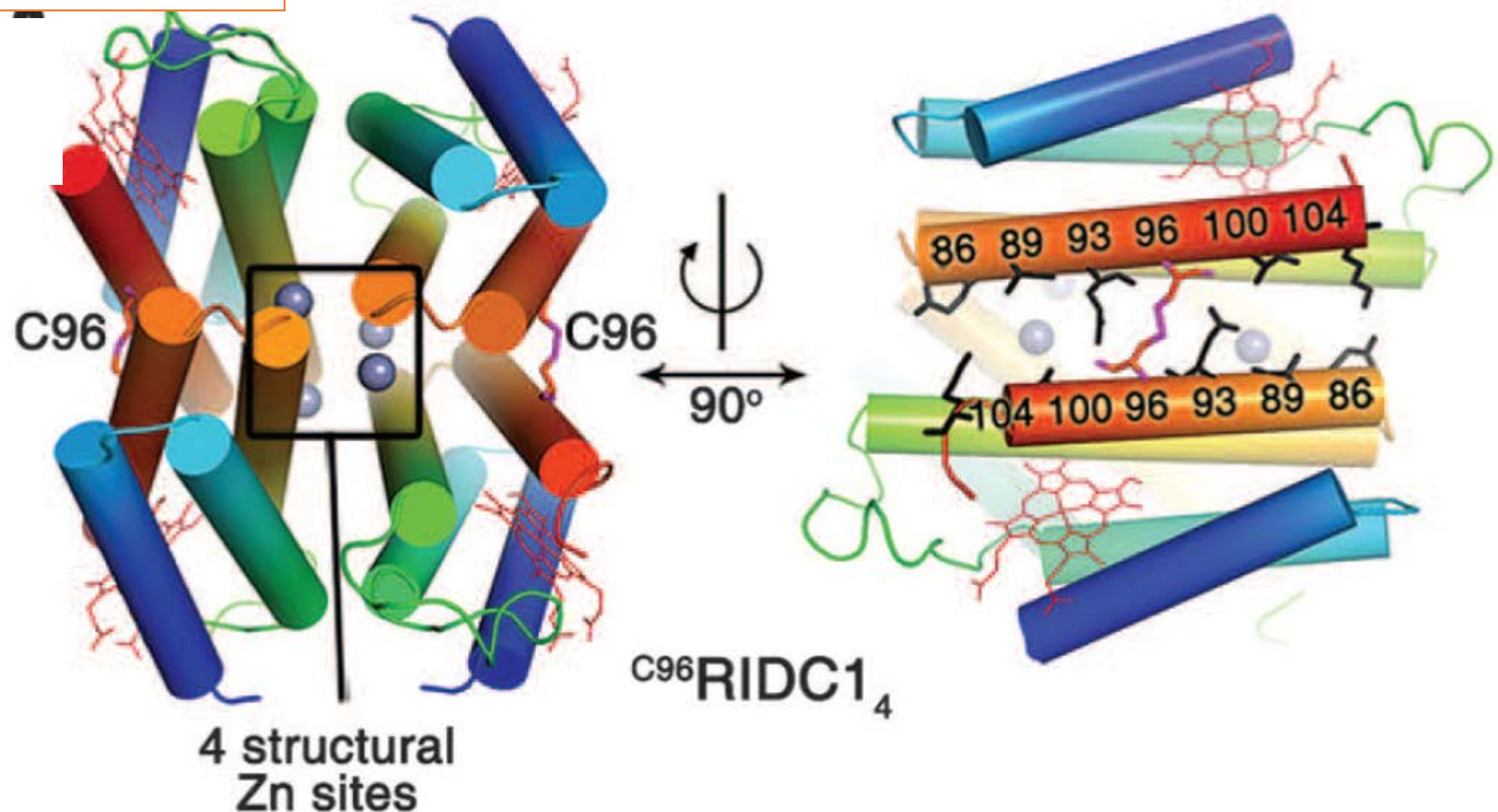
promiscuous activity:



	CA	mimic
$k_{\text{cat}} (\text{s}^{-1})$	56 ± 10	$38 (\pm 10) \times 10^{-3}$
$K_{\text{M}} (\text{mM})$	23.9	2.1 ± 0.6

'bio-available cofactor' – applying selection

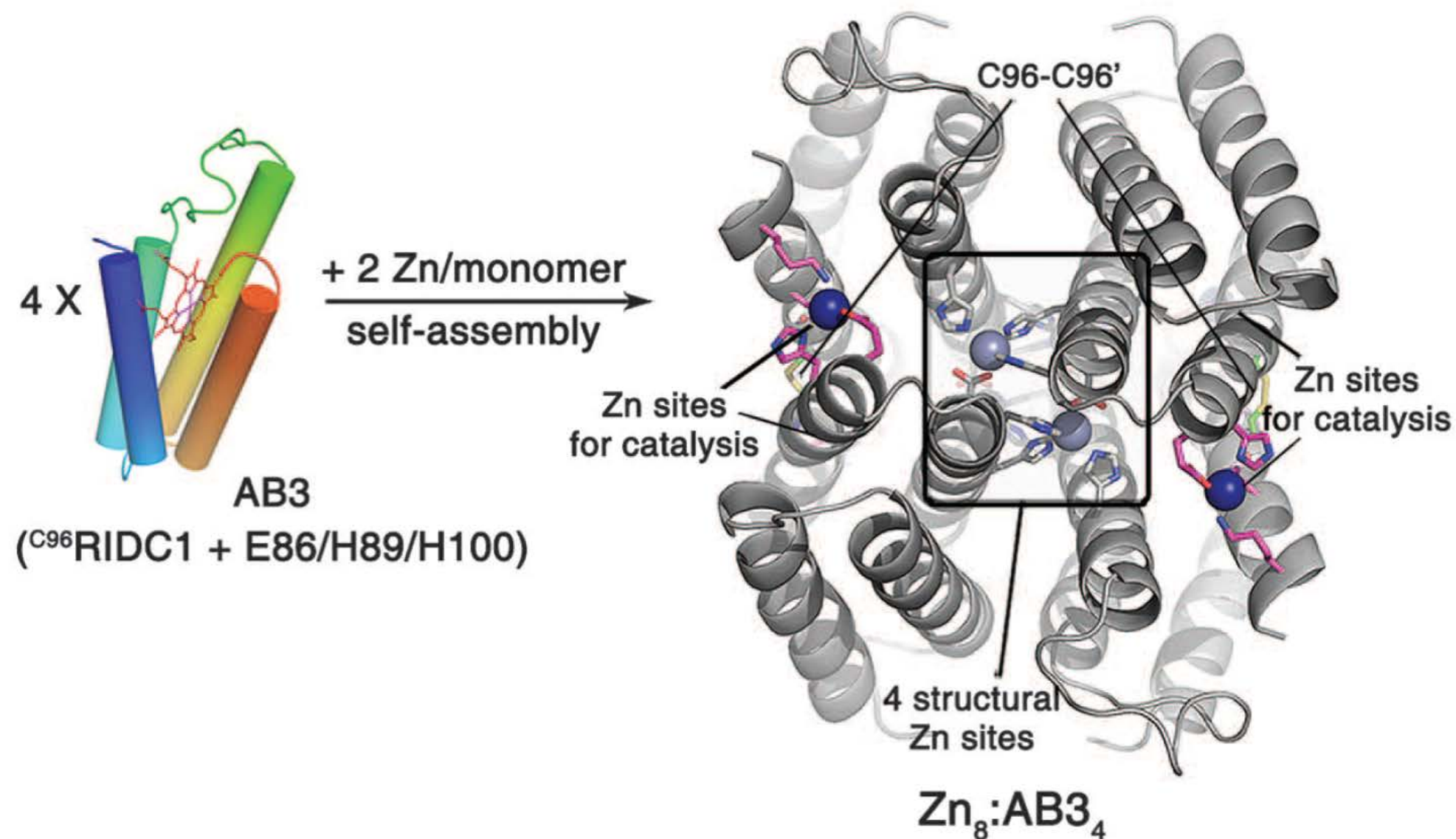
1. starting point



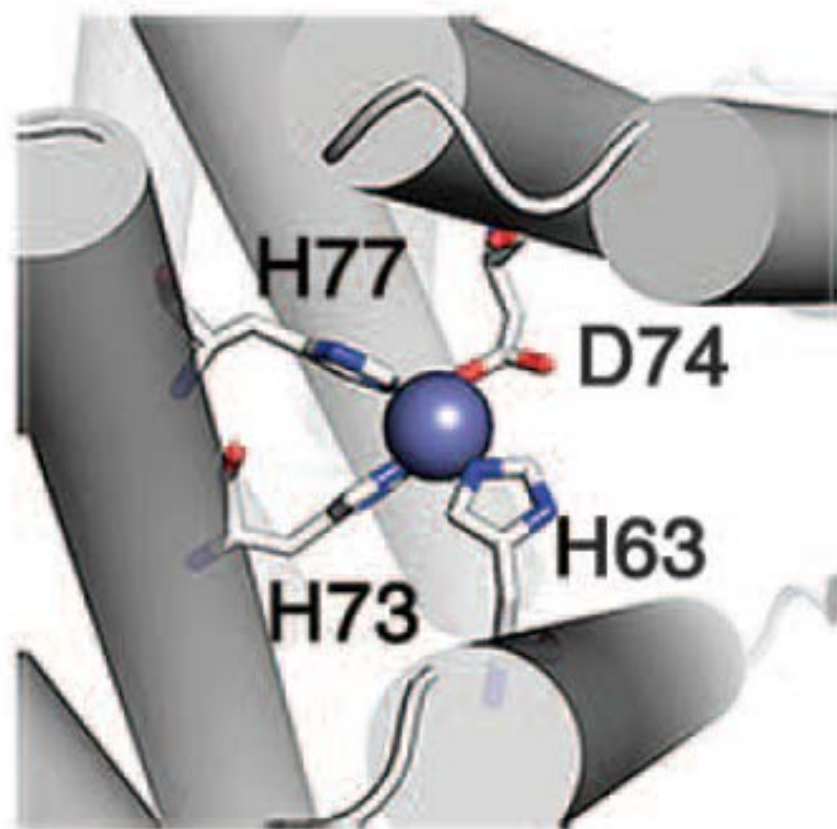
- cytochrome cb₅₆₂ matures in the periplasm
- an engineered mutant of cytochrome cb₅₆₂ self-assembles into a tetramer upon Zn²⁺ coordination
- the assembly is stabilized by S-S bridges

'bio-available cofactor' – applying selection

2. installing additional Zn-binding sites

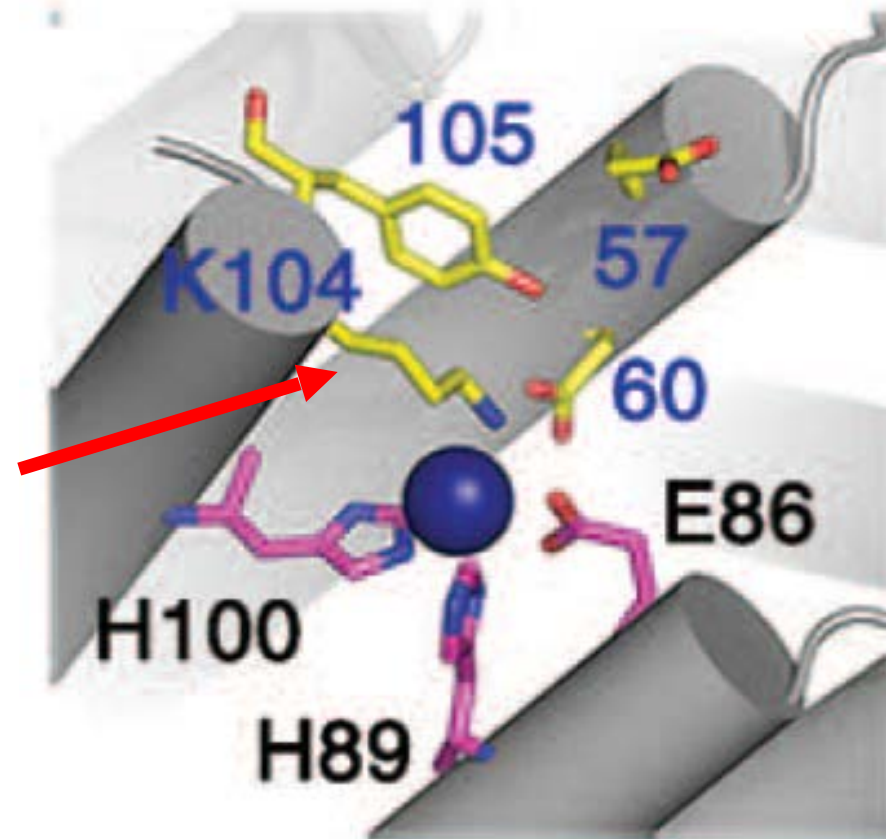


'bio-available cofactor' – applying selection



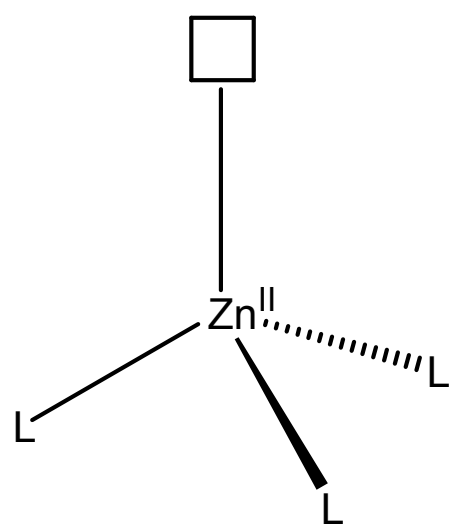
structural Zn site

unexpected
coordination
of K104
→ mutate to
A104

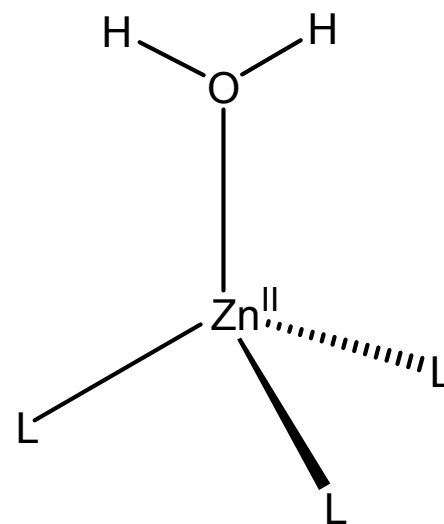


Zn site for catalysis

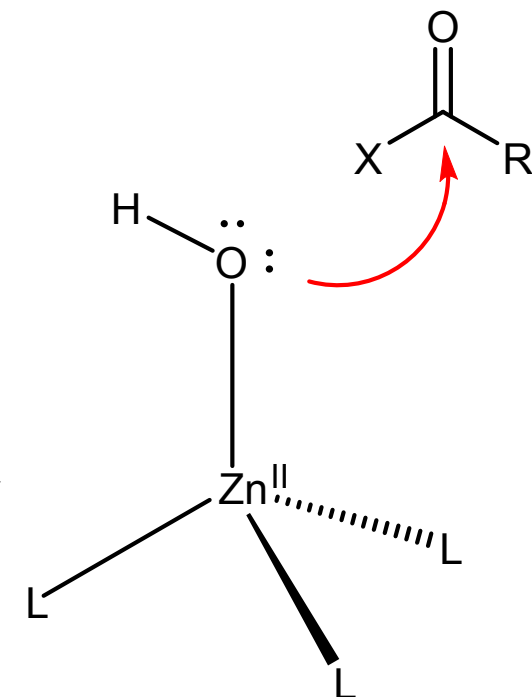
active site in zinc-hydrolases:



free coordination site



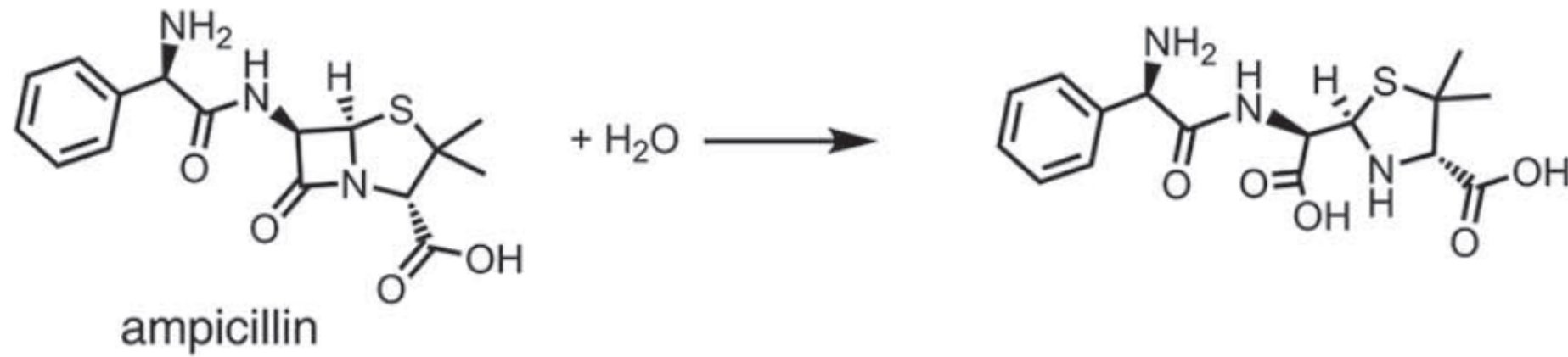
water bound and acidified



good nucleophile

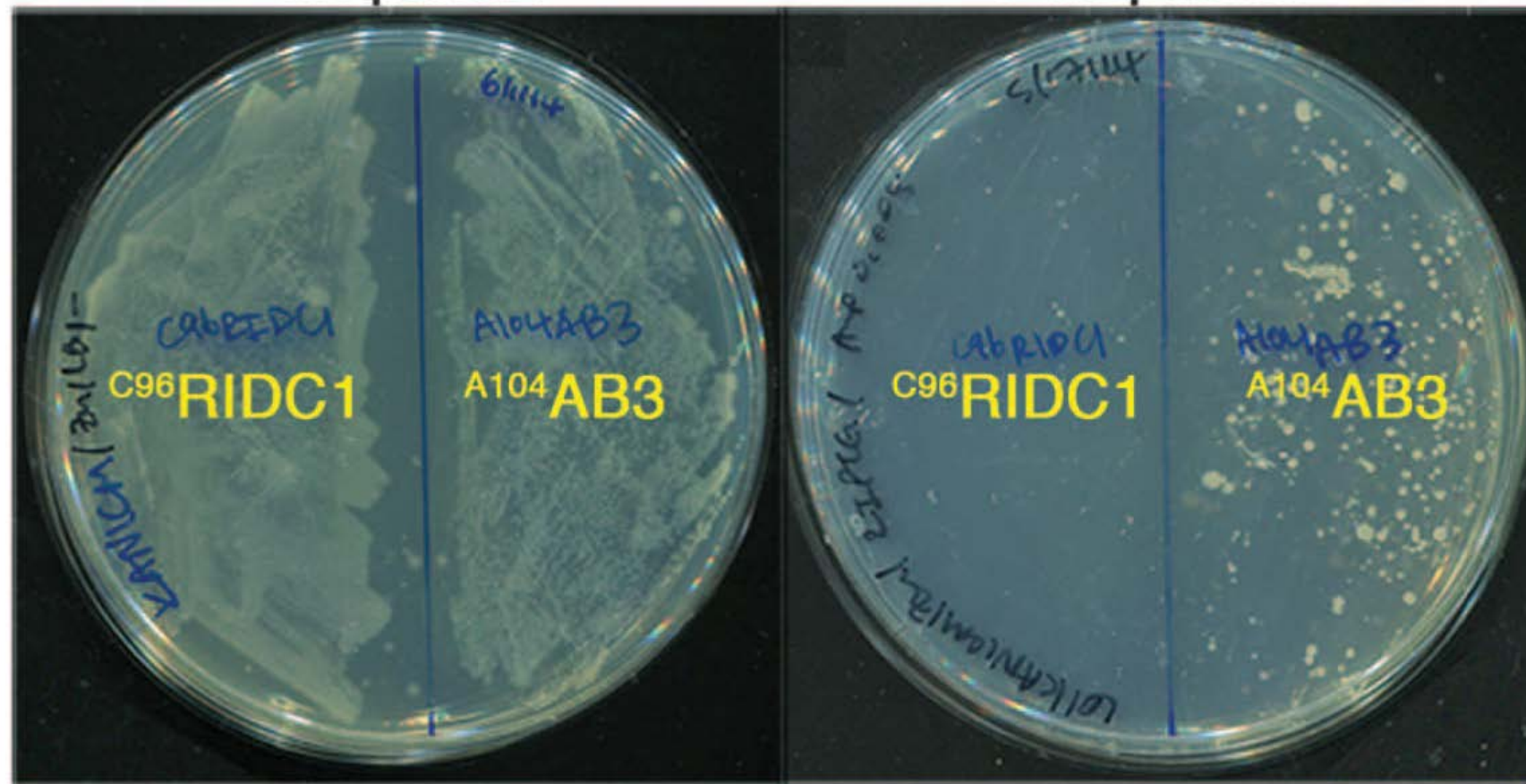
'bio-available cofactor' – applying selection

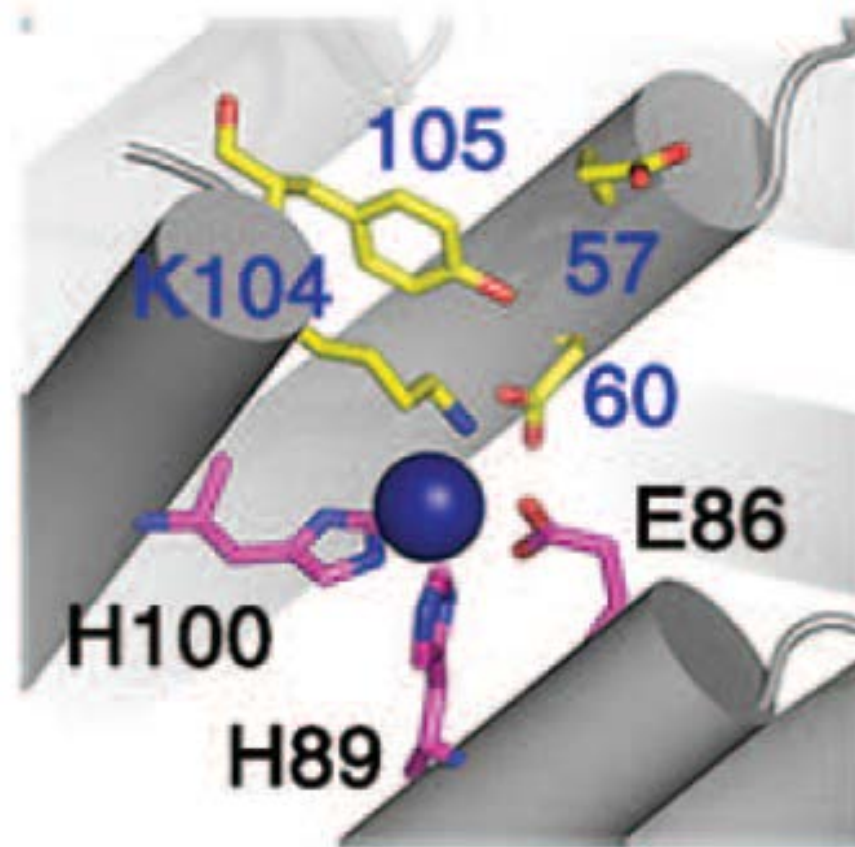
in vivo



– ampicillin

+ ampicillin





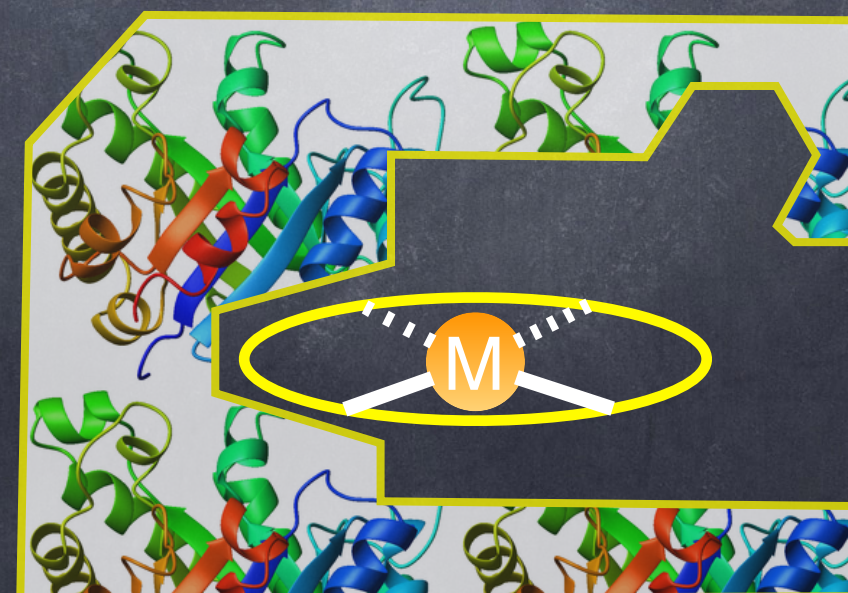
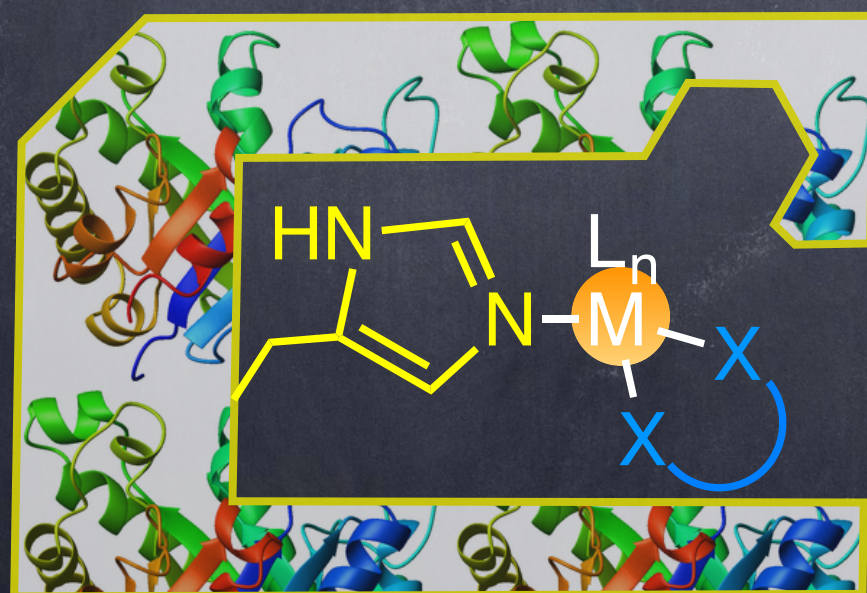
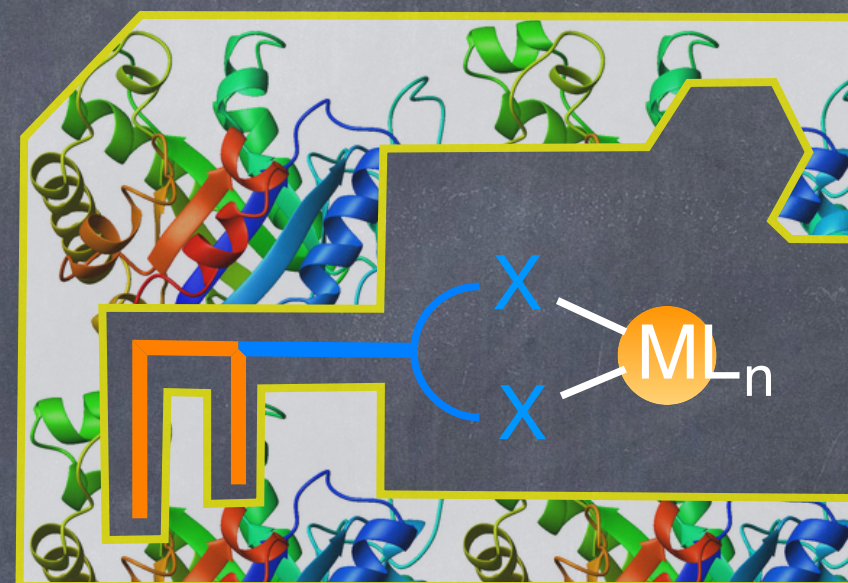
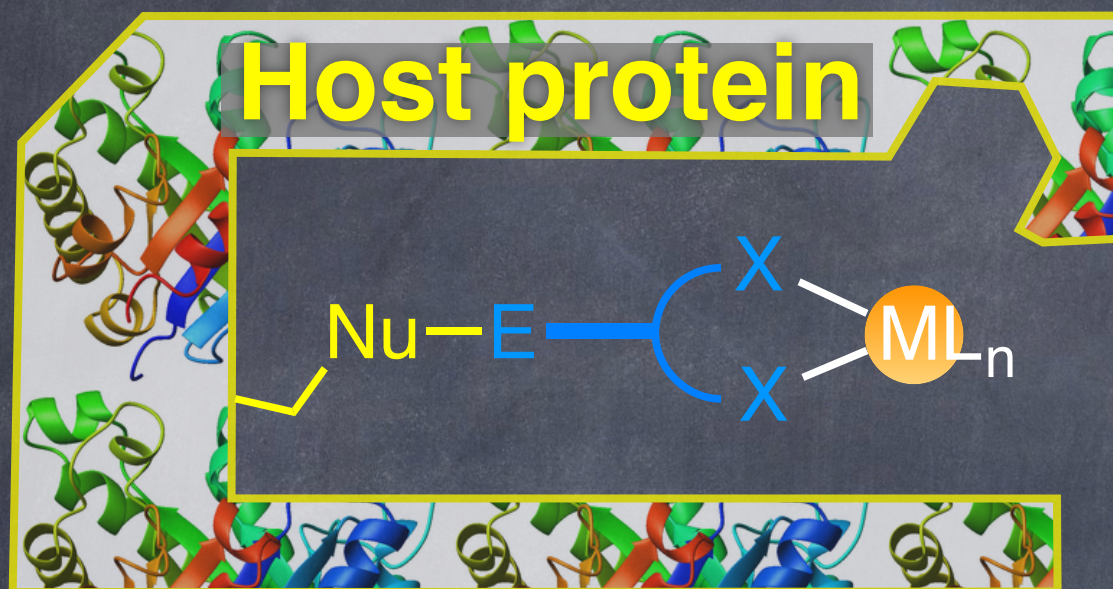
Zn site for catalysis

randomisation of 4 active site residues Glu57, Asp60, Lys104, Tyr105

weak correlation with survival frequency, but the two mutants with highest survival frequency were also the most active in vitro
A104-G57 and A104-T105

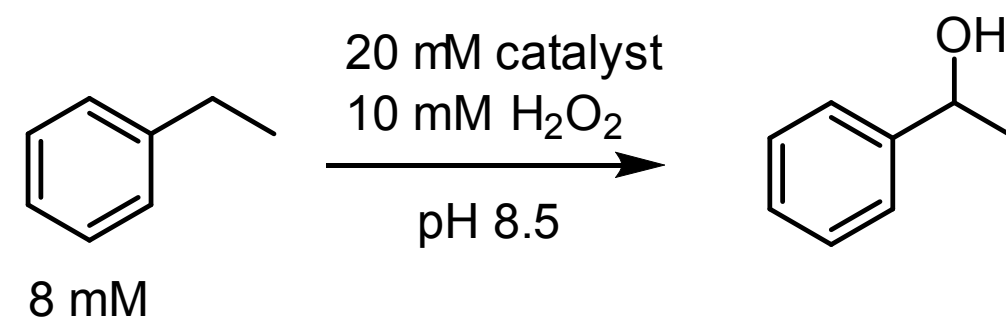
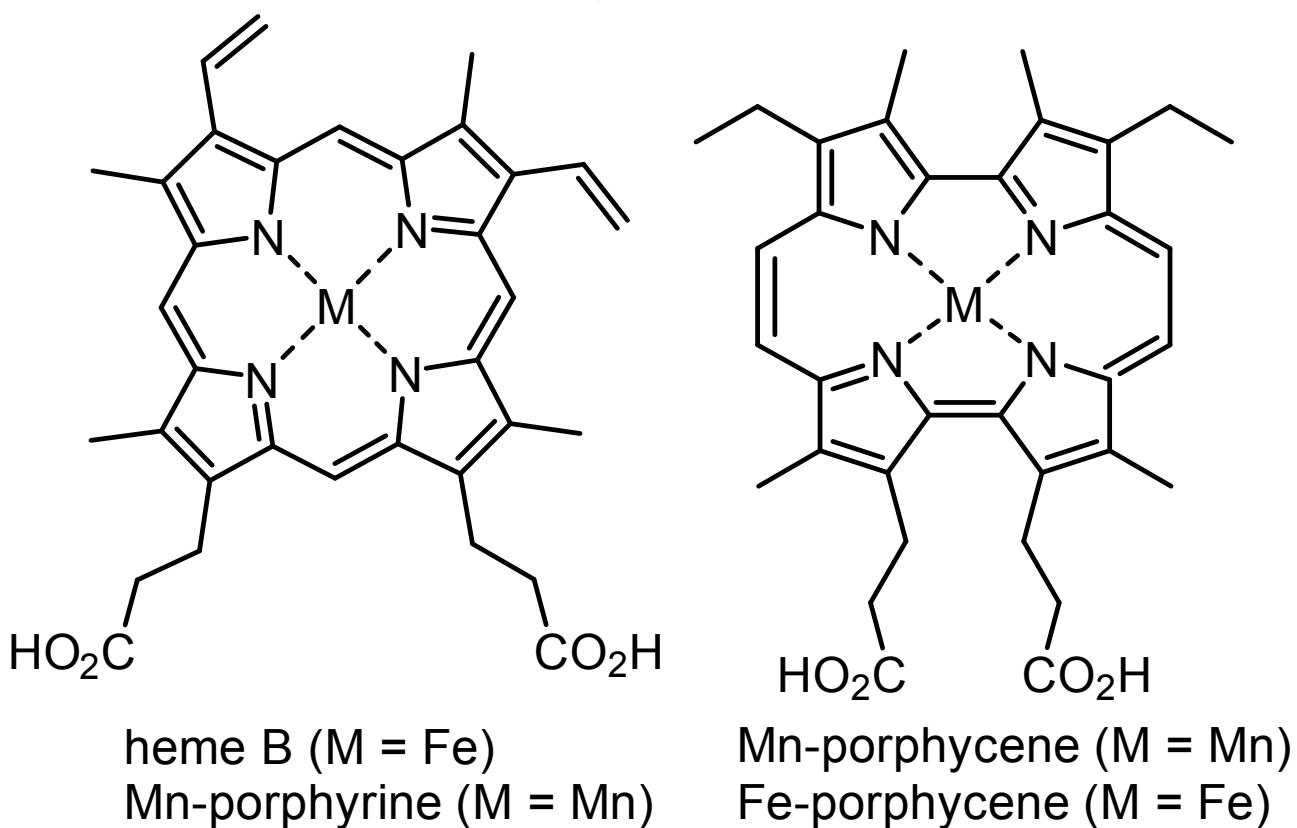
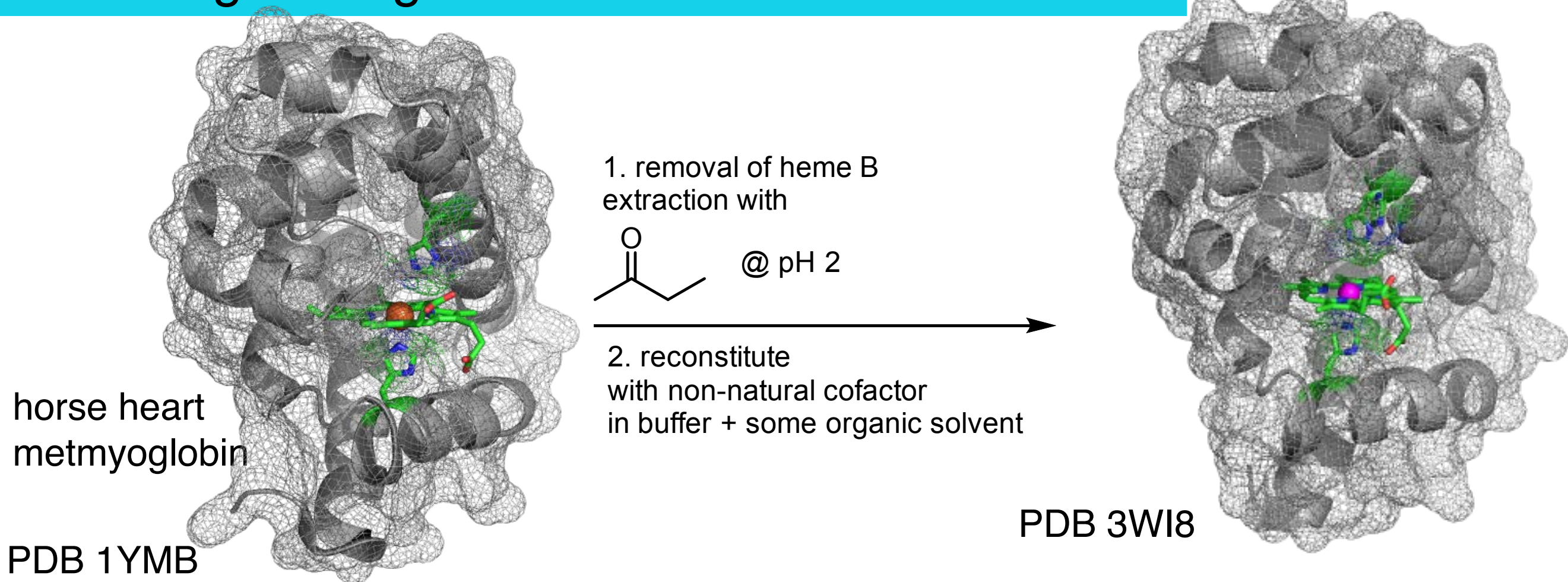
mutant A104-G57 displays Michaelis-Menten kinetics

Anchoring of the Catalyst: Four Alternatives to Ensure Localization



Metal/cofactor substitution

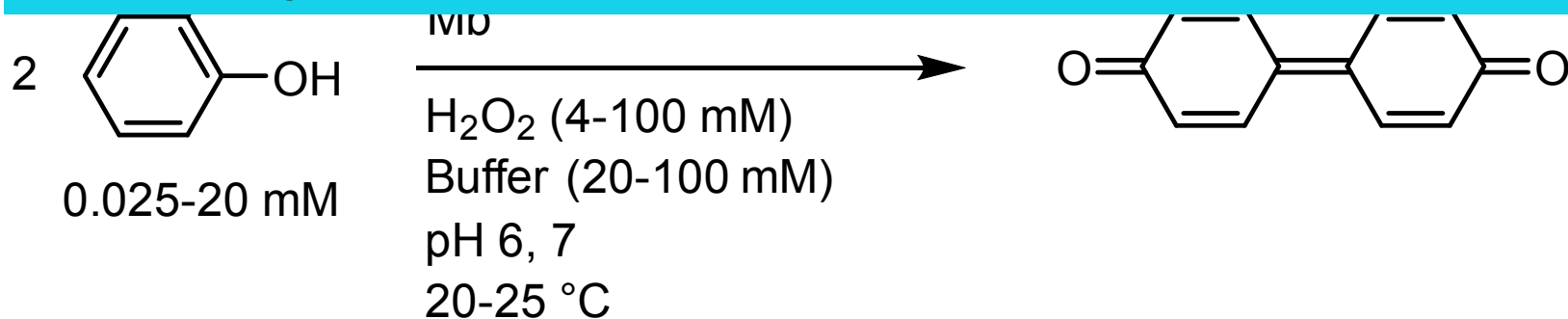
anchoring strategies – metal/cofactor substitution



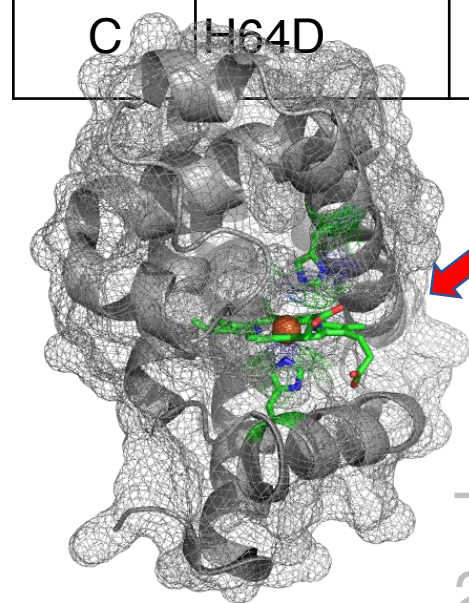
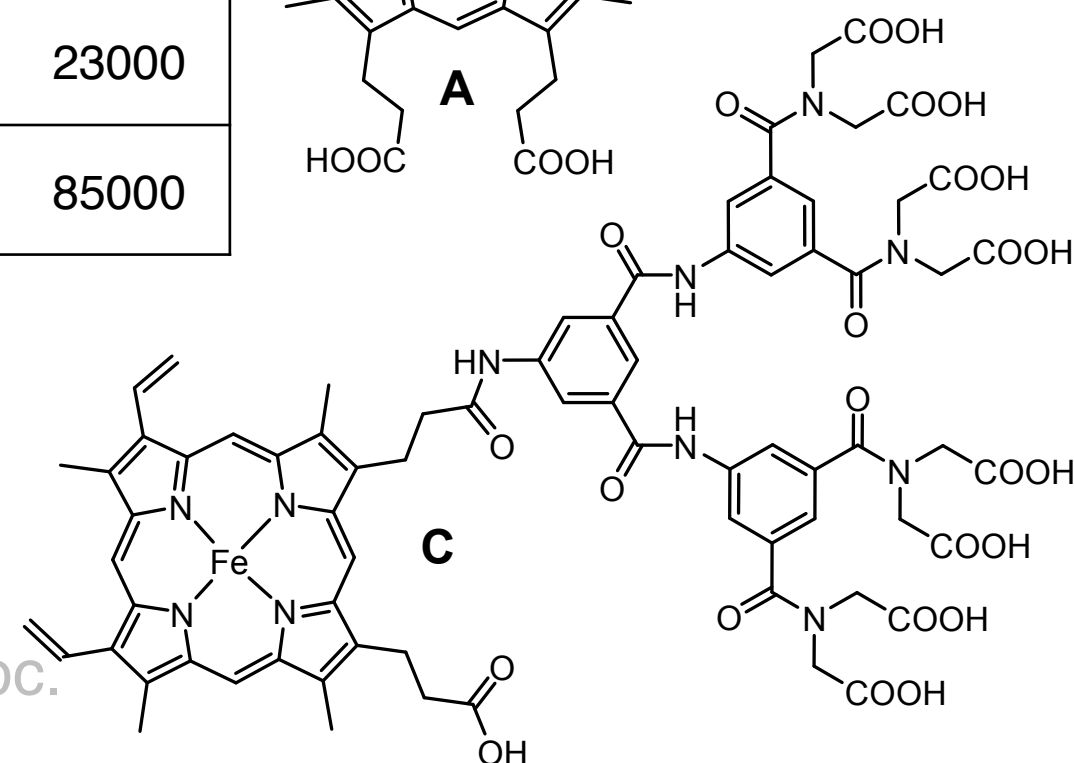
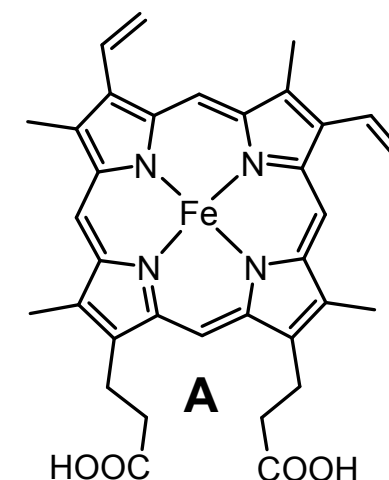
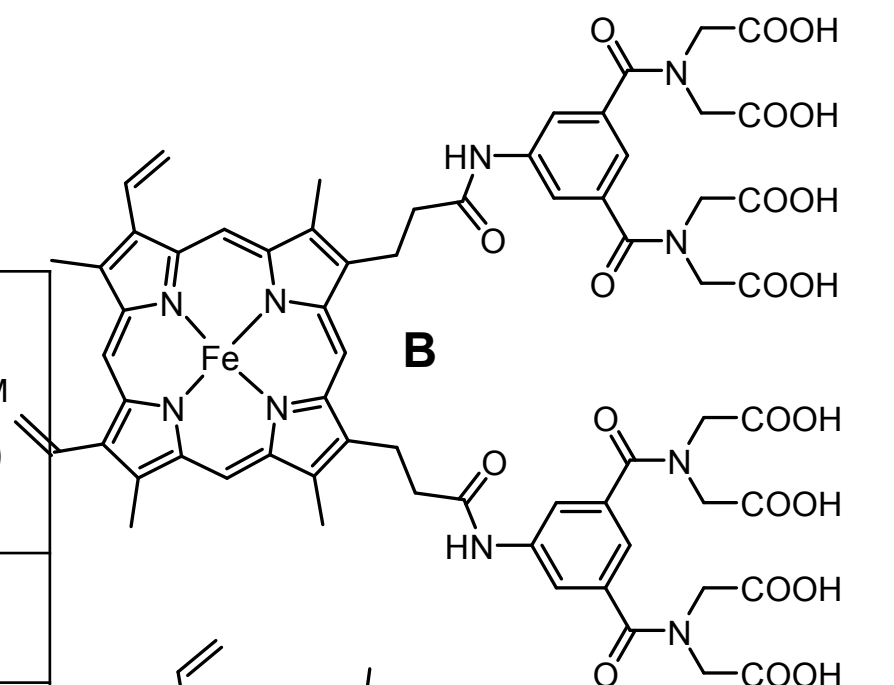
cofactor of protein	TTON	TOF (h ⁻¹)
heme B	0	0
Mn-porphyrine	0	0
Fe-porphycene	0	0
Mn-porphycene	13	33

improving catalytic efficiency through effective molarity?

purified protein



Cof.	Protein	pH	T (°C)	[Cof] (μM)	[H ₂ O ₂] (mM)	K _M (mM ⁻¹)	k _{cat} (s ⁻¹)	k _{cat} /K _M (s ⁻¹ M ⁻¹)
A	WT	7.0	20	1	9.7	32	0.36	11
B	WT	7.0	20	1	9.7	7.4	1.1	149
A	WT	6.0	25	2	100	54	2.8	53
B	WT	6.0	25	2	100	3.4	6.2	1800
A	H64D	6.0	25	2	15	1.8	9.0	5100
B	H64D	6.0	25	2	15	0.052	1.2	23000
C	H64D	6.0	25	4	100	0.29	24	85000



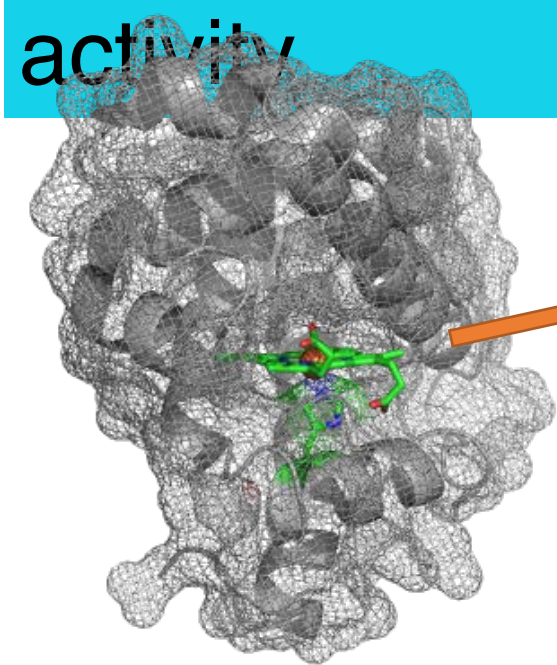
A ↓ B/C

T. Hayashi, Y. Watanabe, J. Am. Chem. Soc.
2004, 126, 436.

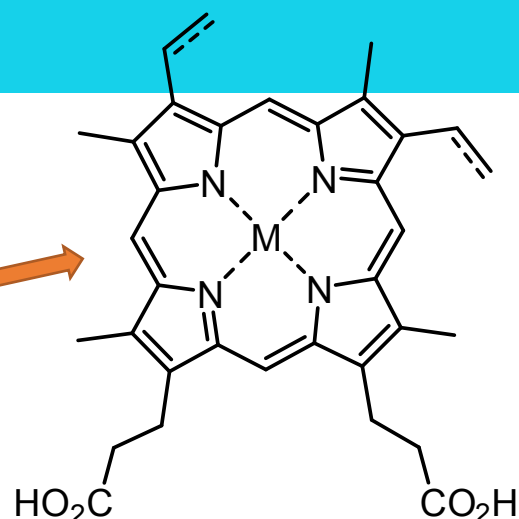
chemogenetic optimization – screening for activity

purified protein

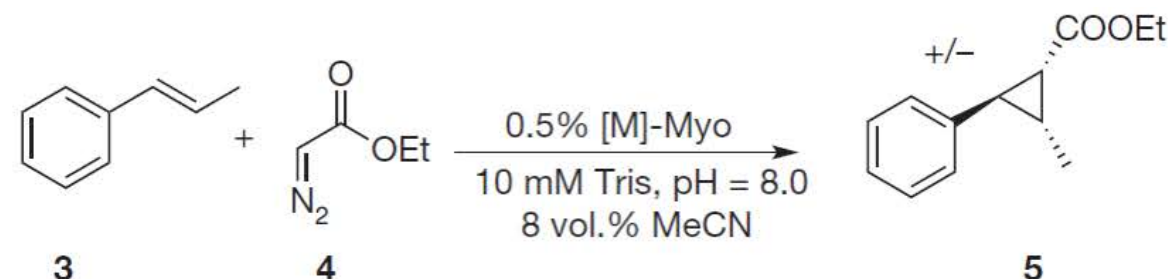
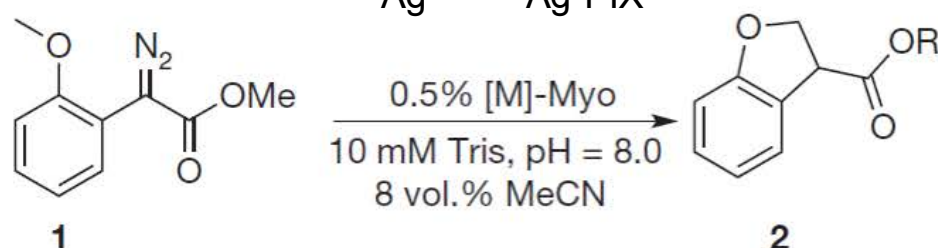
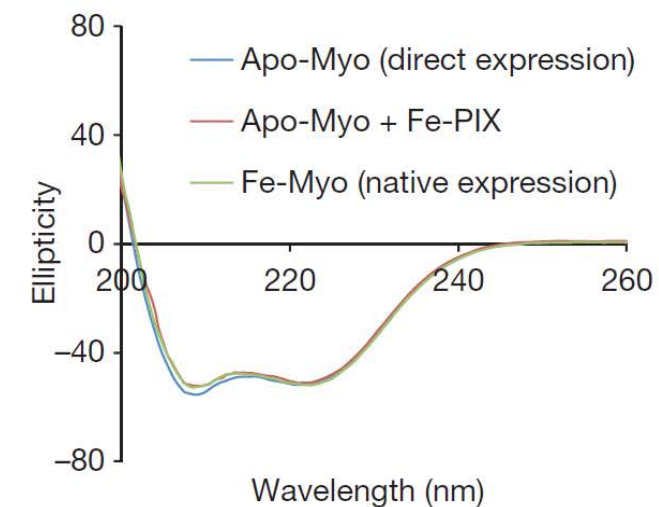
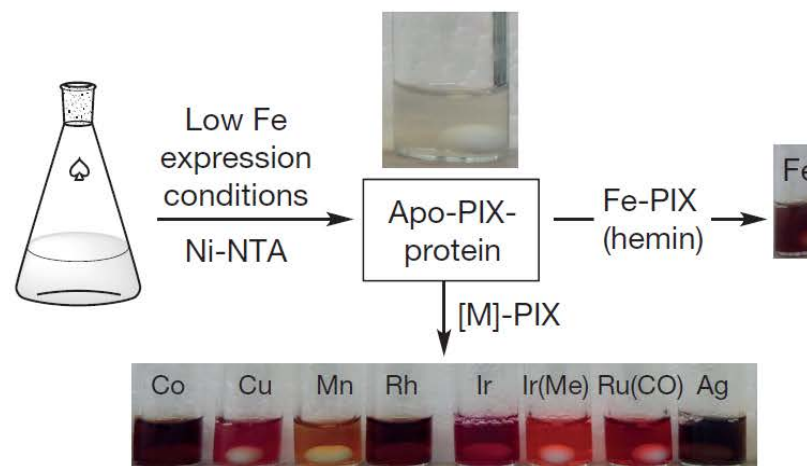
cofactor-free expression



myoglobin



M = Fe^{III}Cl → Fe(Cl)-PIX
 Co^{III}Cl Co(Cl)-PIX
 Cu^{II} Cu-PIX
 Mn^{III}Cl Mn(Cl)-PIX
 Rh[?] Rh-PIX
 Ir^{III}Cl Ir(Cl)-PIX
 Ir^{III}Me Ir(Me)-PIX
 Ru^{II}CO Ru(CO)-PIX
 Ag[?] Ag-PIX



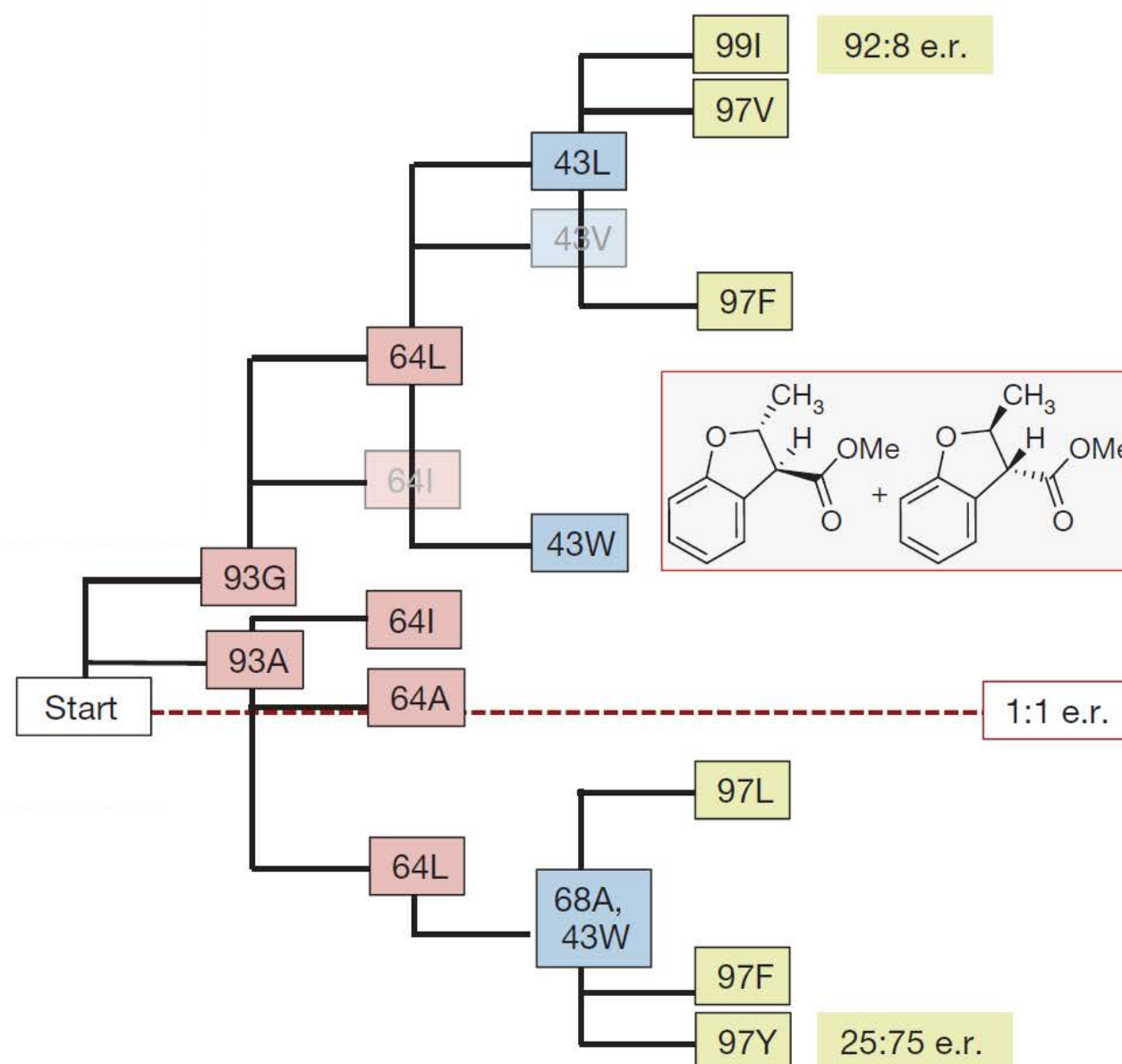
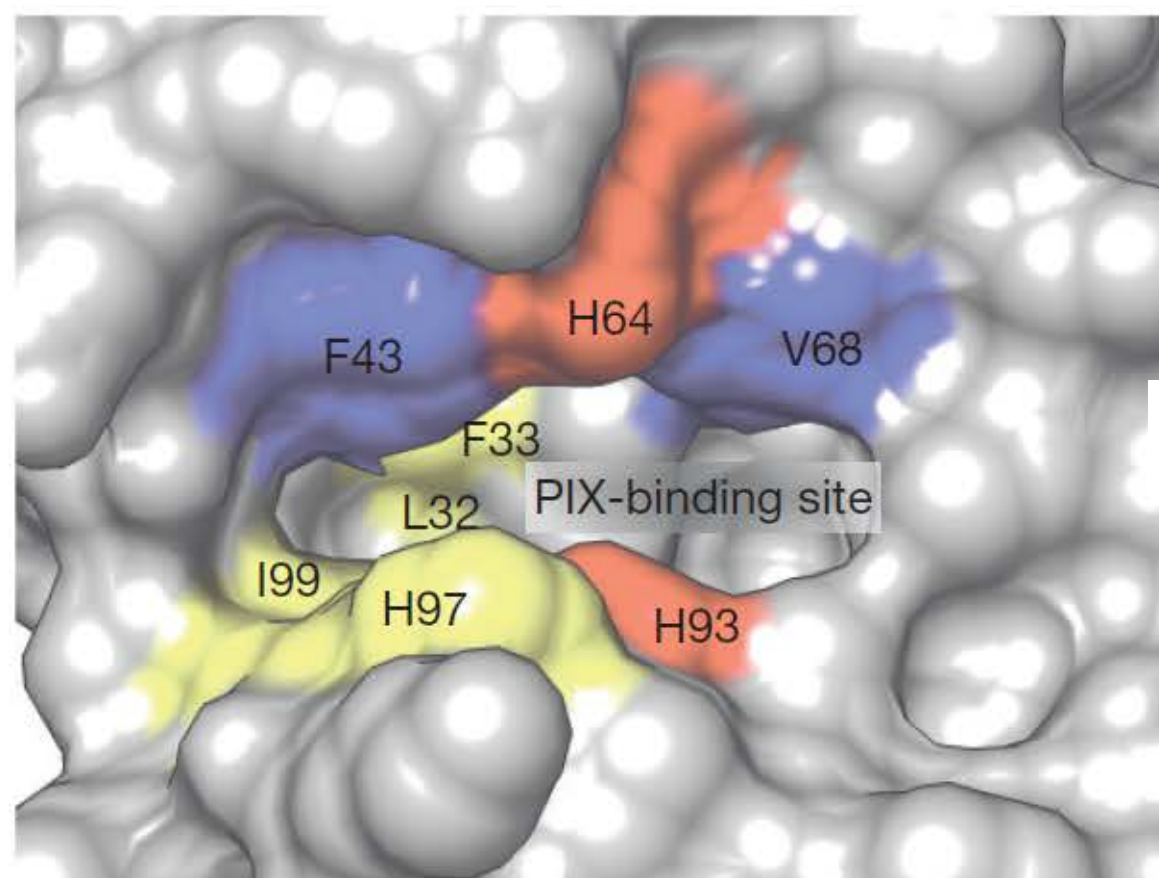
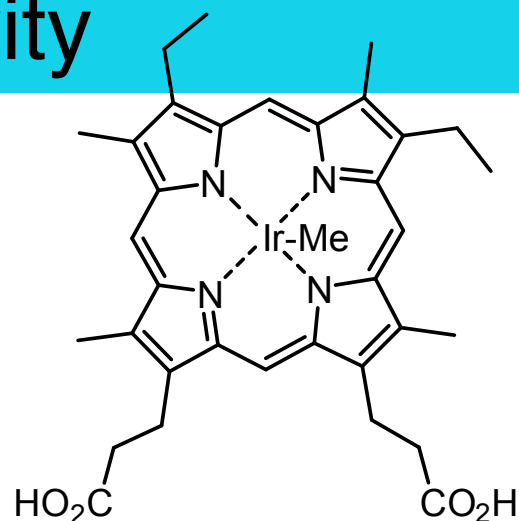
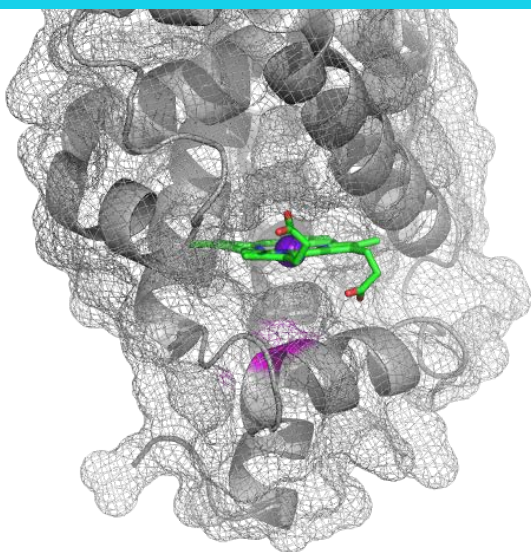
C-H insertion	93H	93C	93D	93E	93M	93S	93A	93G	
Fe(Cl)-PIX									
Co(Cl)-PIX									
Cu-PIX									
Mn(Cl)-PIX									TON
Rh-PIX									<4
Ir(Cl)-PIX									4–10
Ir(Me)-PIX									11–30
Ru(CO)-PIX									31–60
Ag-PIX									>60

Cyclopropanation	93H	93C	93D	93E	93M	93S	93A	93G	
Fe(Cl)-PIX									
Co(Cl)-PIX									
Cu-PIX									
Mn(Cl)-PIX									
Rh-PIX									
Ir(Cl)-PIX									
Ir(Me)-PIX									
Ru(CO)-PIX									
Ag-PIX									

chemogenetic optimization – improving enantioselectivity

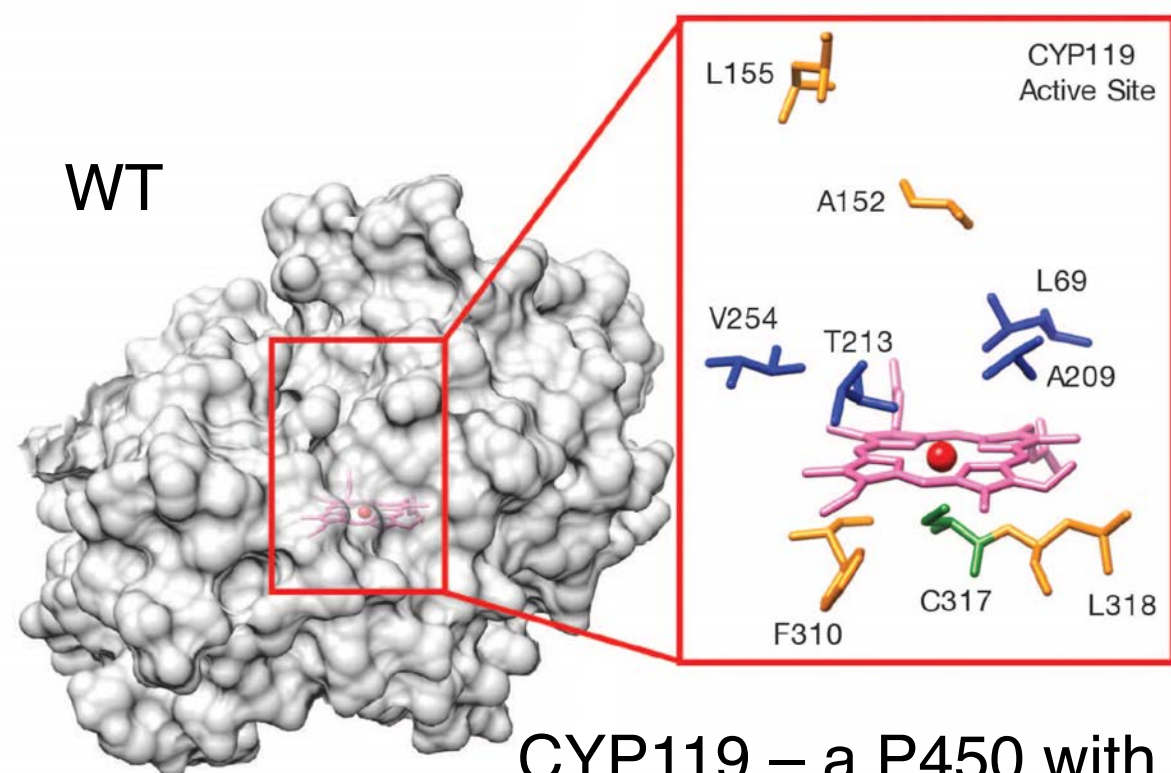
purified protein

ca. 500 mutants screened
degenerate primers

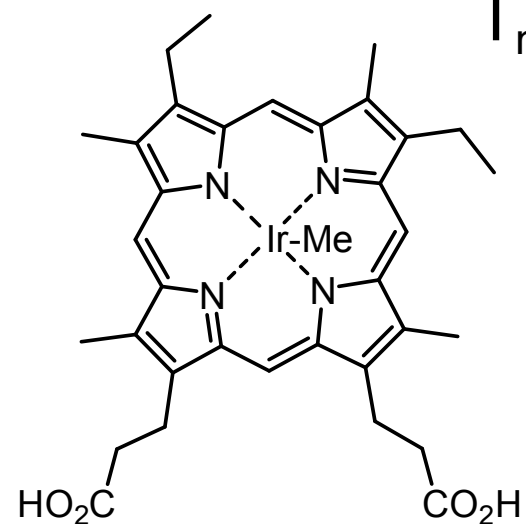


chemogenetic optimization – change scaffold - repeat

purified
protein



CYP119 – a P450 with
high thermostability
 $T_m = 69^\circ\text{C}$

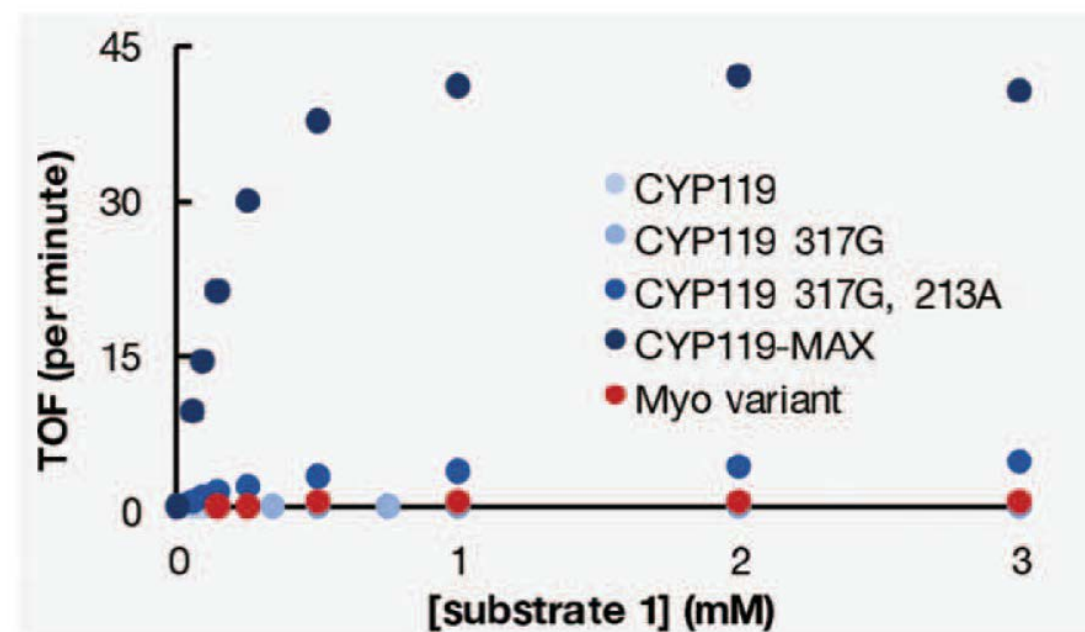
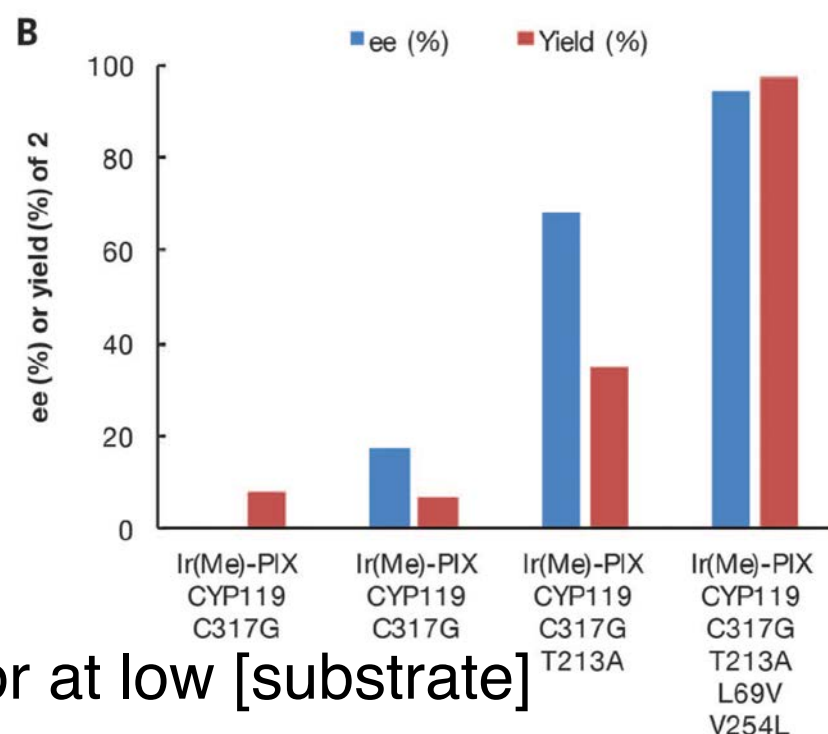
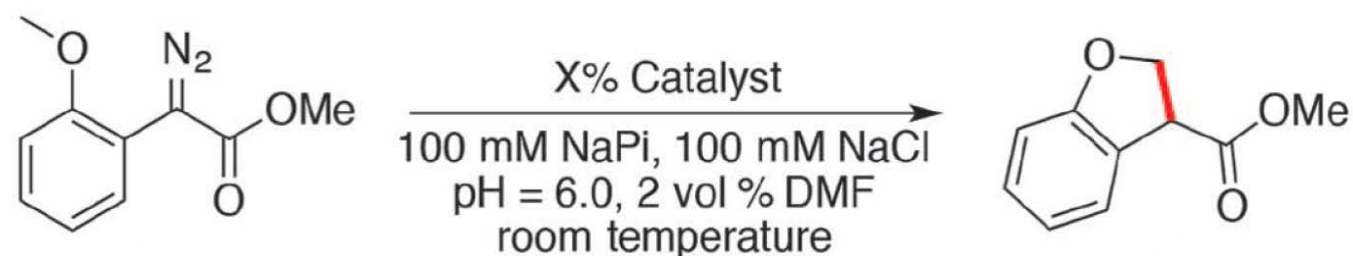


TON up to 35 000

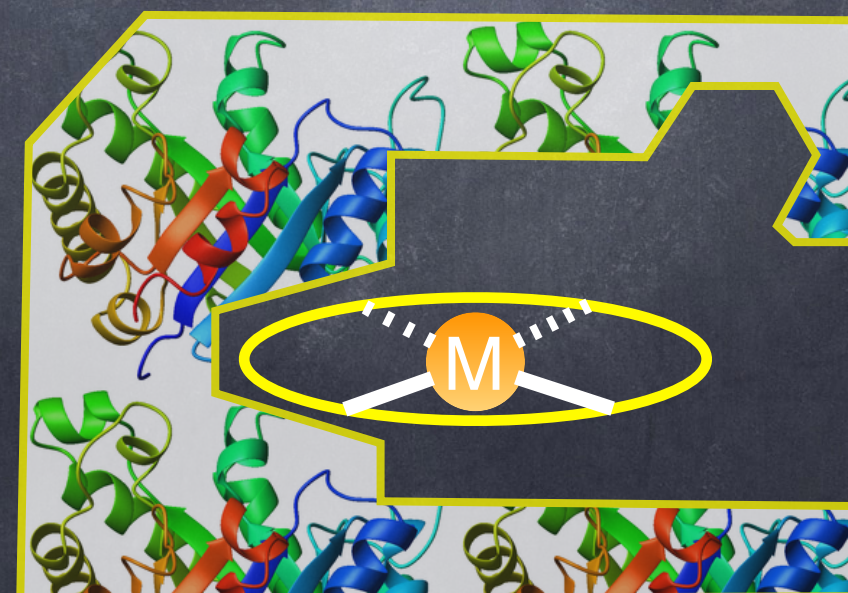
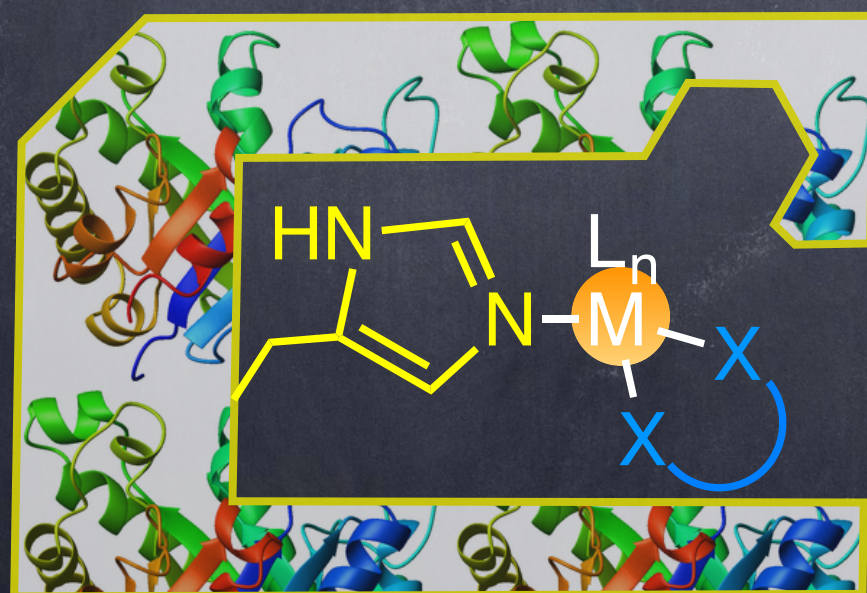
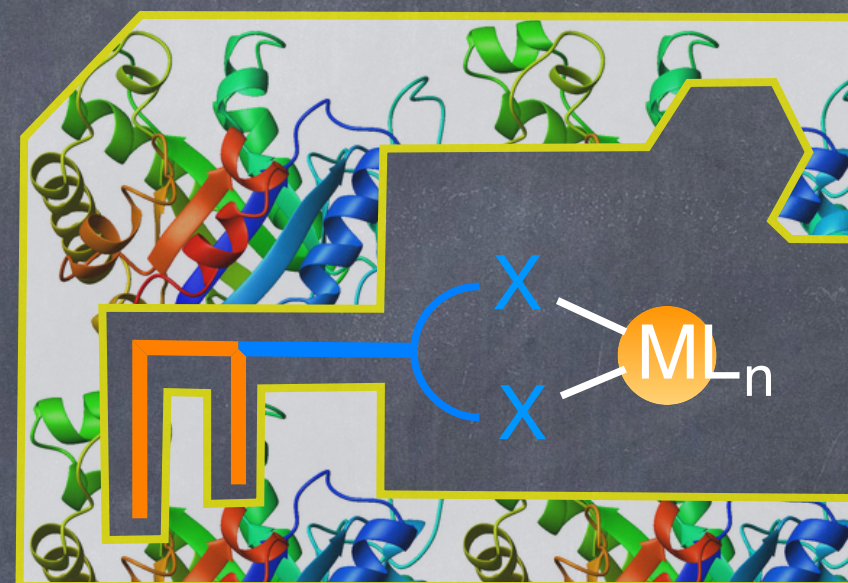
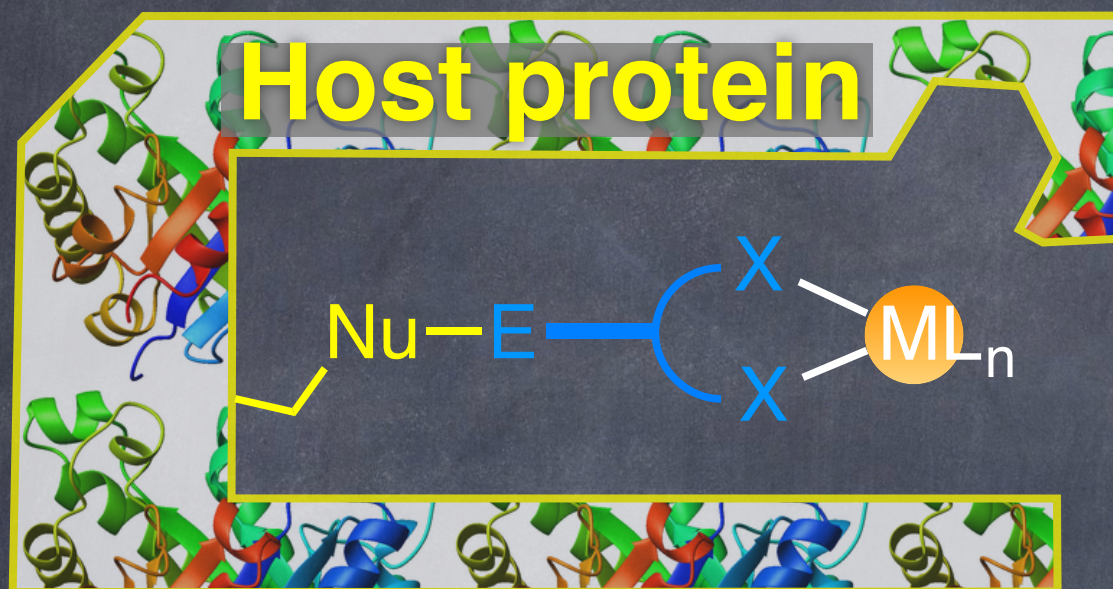
outcompetes free cofactor at low [substrate]

optimization

- exchange C317G to create space for the cofactor
- exchange active site residues for other lipophilic residues
- 24 double mutants tested
- 2 additional mutations introduced

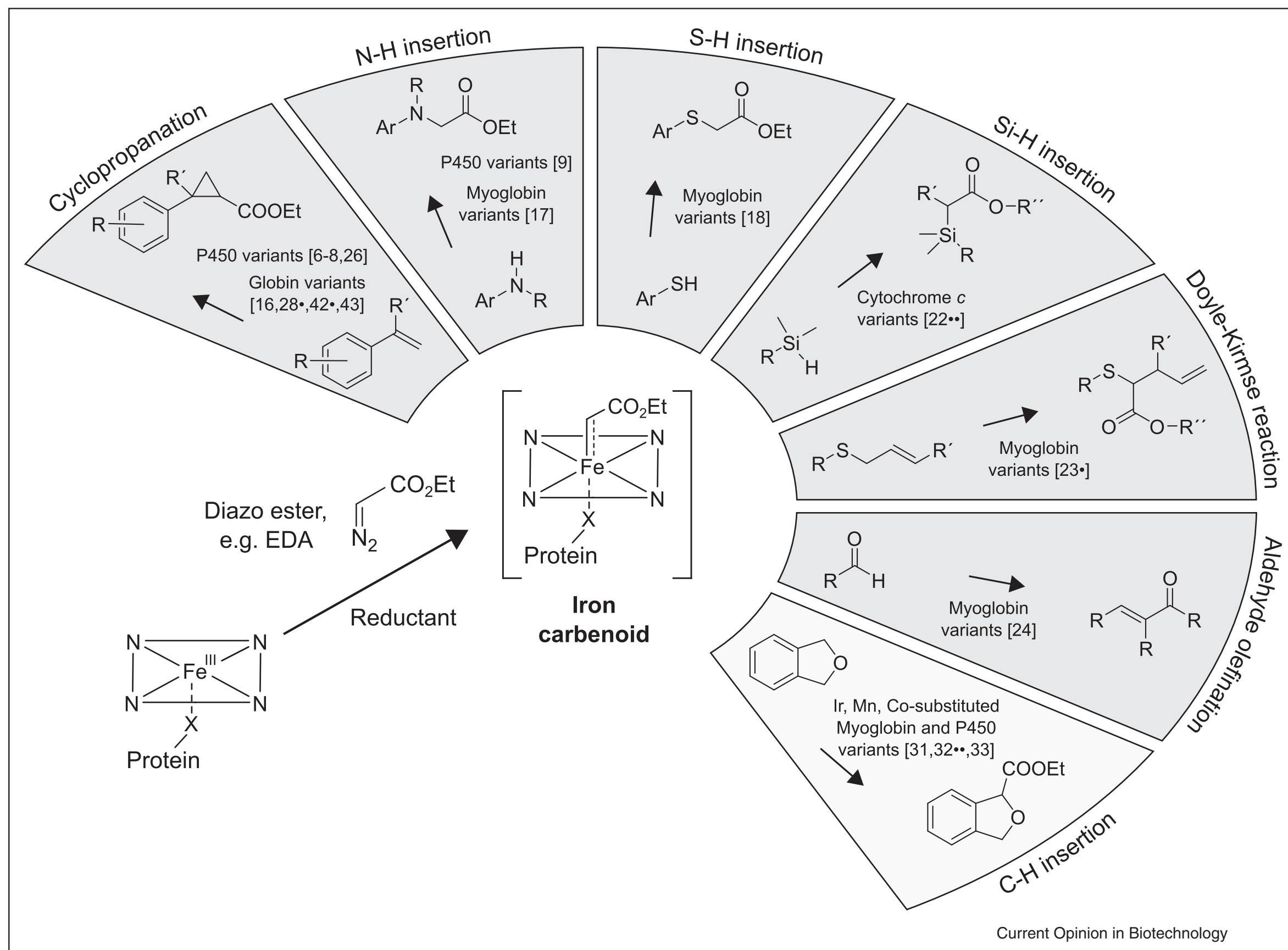


Anchoring of the Catalyst: Four Alternatives to Ensure Localization

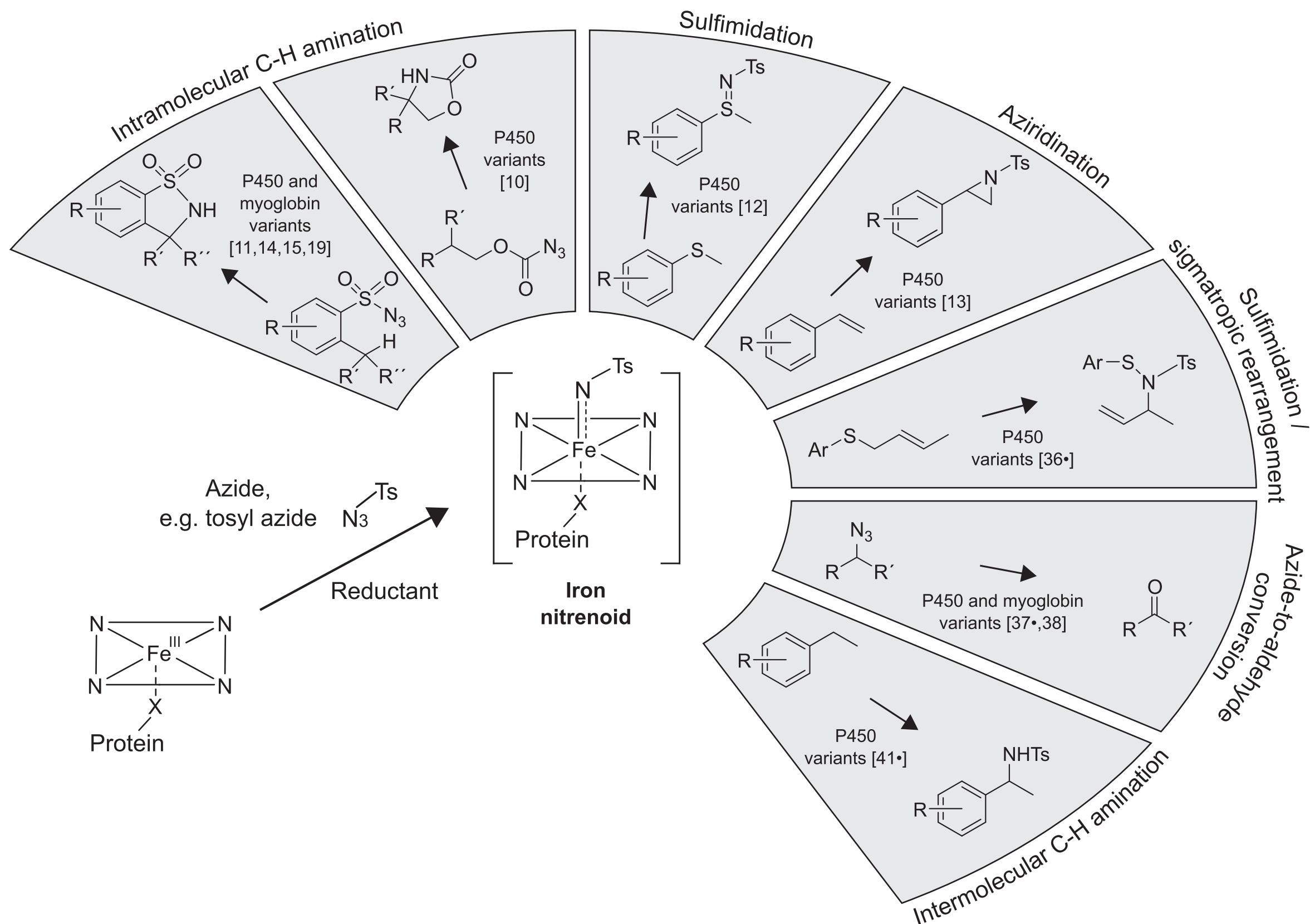


Enzyme repurposing

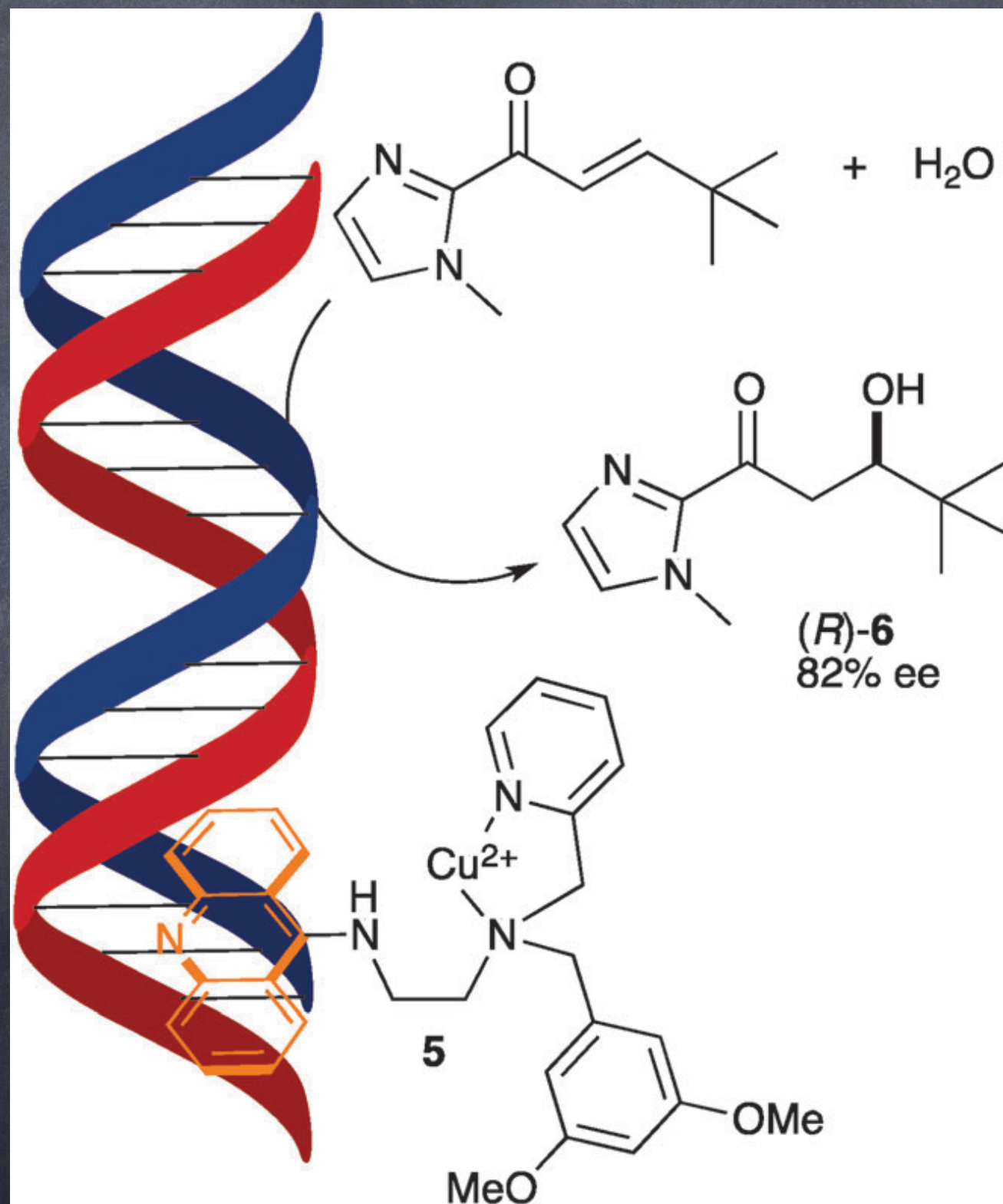
Carbene Transfer Catalyzed by Hemoproteins



Nitrene Transfer Catalyzed by Hemoproteins



DNA as Host for Enantioselective Catalysis



Artificial Metalloenzymes: Homogeneous- or Enzyme-Like?

Homogeneous

Enzymatic

Enantiomers

Both

Single

Solvent Tolerance

Organic

Aqueous

Substrate Specificity

Broad

Narrow

Optimization

Chemical

Genetic

Catalyst Lifetime

Limited

Extended

In vivo Compatibility

Limited

Excellent